



”ATOMIC SPECTROSCOPY AND OPACITY”

- 3 Weeks Lecture Course
- Textbook: ”Atomic Astrophysics and Spectroscopy”
-A.K. Pradhan and S.N. Nahar (Cambridge, 2011)

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- 3 Weeks Lecture Course
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Lecture Topics:

Week 1:

- Physics and Astronomy Connection
 - i) Elements and Spectroscopy
 - ii) Opacity and Observations
 - iii) The Sun and Evolution of Spectroscopy
 - iv) Electron and Photon Distribution Functions
 - v) The Plasma Universe
- Atomic Structure and Spectroscopy
 - i) Hydrogen - Single electron energy and wave functions
 - ii) Quantum numbers n, l, m, s , parity
 - iii) Spectral lines - Rydberg formula

Week 2:

- Atomic Structure, Computation with SUPERSTRUCTURE & Atomic Processes
 - i) Multi-electron Systems - Angular States
 - ii) Angular Couplings, Quantum Defects
 - iii) Hartree-Fock Approximation
 - iv) Central Field Approximations
 - v) Dirac Equation, Breit-Pauli Approximation, Dirac-Fock

Approximation

vi) Atomic Processes of Plasma Opacity

Week 3:

- Plasma Opacity, R-matrix Calculations for Atomic Processes

- i) Autoionization and Resonances and Opacity

- ii) Plasma Opacity

- iii) Close-Coupling Approximation and R-matrix Method

- iv) R-matrix Calculations

i) Elements and Spectroscopy: ELEMENTS (Key Information)

PERIODIC TABLE
Atomic Properties of the Elements

NIST
National Institute of Standards and Technology
Technology Administration, U.S. Department of Commerce

Frequently used fundamental physical constants

For the most accurate values of these and other constants, visit physics.nist.gov/constants

1 second = 9 192 631 770 periods of radiation corresponding to the transition between the two hyperfine levels of the ground state of ¹³³Cs

speed of light in vacuum	<i>c</i>	299 792 458 m s ⁻¹	(exact)
Planck constant	<i>h</i>	6.6261 × 10 ⁻³⁴ J s	(<i>h</i> = <i>h</i> /2π)
elementary charge	<i>e</i>	1.6022 × 10 ⁻¹⁹ C	
electron mass	<i>m_e</i>	9.1094 × 10 ⁻³¹ kg	
	<i>m_ec²</i>	0.5110 MeV	
proton mass	<i>m_p</i>	1.6726 × 10 ⁻²⁷ kg	
fine-structure constant	<i>α</i>	1/137.036	
Rydberg constant	<i>R_∞</i>	10 973 732 m ⁻¹	
	<i>R_∞c</i>	3.289 842 × 10 ¹⁵ Hz	
	<i>R_∞hc</i>	13.6057 eV	
Boltzmann constant	<i>k</i>	1.3807 × 10 ⁻²³ J K ⁻¹	

■ Solids	■ Liquids	■ Gases	■ Artificially Prepared
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Group	PERIODIC TABLE																National Institute of Standards and Technology Technology Administration, U.S. Department of Commerce																																																																																																																	
	1 IA	2 IIA		3 IIIB	4 IVB	5 VB	6 VIB	7 VIIB	8 VIII			9 IB	10 IIB	13 IIIA	14 IVA	15 VA	16 VIA	17 VIIA	18 VIIIA																																																																																																															
1	1 ¹ H Hydrogen 1.00794 1s													5 ¹⁰ B Boron 10.811 1s ² 2s ² 2p	6 ¹² C Carbon 12.0107 1s ² 2s ² 2p	7 ¹⁴ N Nitrogen 14.0067 1s ² 2s ² 2p	8 ¹⁶ O Oxygen 15.9994 1s ² 2s ² 2p	9 ¹⁸ F Fluorine 18.9984032 1s ² 2s ² 2p	10 ²⁰ Ne Neon 20.1797 1s ² 2s ² 2p	13 ¹³ Al Aluminum 26.981538 [Ne]3s ² 3p	14 ¹⁴ Si Silicon 28.0855 [Ne]3s ² 3p	15 ¹⁵ P Phosphorus 30.973761 [Ne]3s ² 3p	16 ¹⁶ S Sulfur 32.065 [Ne]3s ² 3p	17 ¹⁷ Cl Chlorine 35.453 [Ne]3s ² 3p	18 ¹⁸ Ar Argon 39.948 [Ne]3s ² 3p	19 ¹⁹ K Potassium 39.0983 [Ar]4s	20 ²⁰ Ca Calcium 40.078 [Ar]4s	21 ²¹ Sc Scandium 44.955910 [Ar]3d ¹ 4s ²	22 ²² Ti Titanium 47.867 [Ar]3d ² 4s ²	23 ²³ V Vanadium 50.9415 [Ar]3d ³ 4s ²	24 ²⁴ Cr Chromium 51.9961 [Ar]3d ⁵ 4s ¹	25 ²⁵ Mn Manganese 54.938049 [Ar]3d ⁵ 4s ²	26 ²⁶ Fe Iron 55.845 [Ar]3d ⁶ 4s ²	27 ²⁷ Co Cobalt 58.933200 [Ar]3d ⁷ 4s ²	28 ²⁸ Ni Nickel 58.6934 [Ar]3d ⁸ 4s ²	29 ²⁹ Cu Copper 63.546 [Ar]3d ¹⁰ 4s ¹	30 ³⁰ Zn Zinc 65.409 [Ar]3d ¹⁰ 4s ²	31 ³¹ Ga Gallium 69.723 [Ar]3d ¹⁰ 4s ² 4p	32 ³² Ge Germanium 72.64 [Ar]3d ¹⁰ 4s ² 4p	33 ³³ As Arsenic 74.92160 [Ar]3d ¹⁰ 4s ² 4p	34 ³⁴ Se Selenium 78.96 [Ar]3d ¹⁰ 4s ² 4p	35 ³⁵ Br Bromine 79.904 [Ar]3d ¹⁰ 4s ² 4p	36 ³⁶ Kr Krypton 83.798 [Ar]3d ¹⁰ 4s ² 4p																																																																																							
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5	37 ³⁷ Rb Rubidium 85.4678 [Kr]5s	38 ³⁸ Sr Strontium 87.62 [Kr]5s	39 ³⁹ Y Yttrium 88.90585 [Kr]4d ¹ 5s ²	40 ⁴⁰ Zr Zirconium 91.224 [Kr]4d ² 5s ²	41 ⁴¹ Nb Niobium 92.90638 [Kr]4d ⁴ 5s ¹	42 ⁴² Mo Molybdenum 95.94 [Kr]4d ⁵ 5s ¹	43 ⁴³ Tc Technetium (98) [Kr]4d ⁵ 5s ²	44 ⁴⁴ Ru Ruthenium 101.07 [Kr]4d ⁷ 5s ¹	45 ⁴⁵ Rh Rhodium 102.90550 [Kr]4d ⁸ 5s ¹	46 ⁴⁶ Pd Palladium 106.42 [Kr]4d ¹⁰	47 ⁴⁷ Ag Silver 107.8682 [Kr]4d ¹⁰ 5s ¹	48 ⁴⁸ Cd Cadmium 112.411 [Kr]4d ¹⁰ 5s ²	49 ⁴⁹ In Indium 114.818 [Kr]4d ¹⁰ 5s ² 5p	50 ⁵⁰ Sn Tin 118.710 [Kr]4d ¹⁰ 5s ² 5p	51 ⁵¹ Sb Antimony 121.760 [Kr]4d ¹⁰ 5s ² 5p	52 ⁵² Te Tellurium 127.60 [Kr]4d ¹⁰ 5s ² 5p	53 ⁵³ I Iodine 126.90447 [Kr]4d ¹⁰ 5s ² 5p	54 ⁵⁴ Xe Xenon 131.293 [Kr]4d ¹⁰ 5s ² 5p																																																																																																																
6	55 ⁵⁵ Cs Cesium 132.90545 [Xe]6s	56 ⁵⁶ Ba Barium 137.327 [Xe]6s	72 ⁷² Hf Hafnium 178.49 [Xe]4f ¹⁴ 5d ² 6s ²	73 ⁷³ Ta Tantalum 180.9479 [Xe]4f ¹⁴ 5d ³ 6s ²	74 ⁷⁴ W Tungsten 183.84 [Xe]4f ¹⁴ 5d ⁴ 6s ²	75 ⁷⁵ Re Rhenium 186.207 [Xe]4f ¹⁴ 5d ⁵ 6s ²	76 ⁷⁶ Os Osmium 190.23 [Xe]4f ¹⁴ 5d ⁶ 6s ²	77 ⁷⁷ Ir Iridium 192.217 [Xe]4f ¹⁴ 5d ⁷ 6s ²	78 ⁷⁸ Pt Platinum 195.078 [Xe]4f ¹⁴ 5d ⁹ 6s ¹	79 ⁷⁹ Au Gold 196.96655 [Xe]4f ¹⁴ 5d ¹⁰ 6s ¹	80 ⁸⁰ Hg Mercury 200.59 [Xe]4f ¹⁴ 5d ¹⁰ 6s ²	81 ⁸¹ Tl Thallium 204.3833 [Xe]6p	82 ⁸² Pb Lead 207.2 [Xe]6p	83 ⁸³ Bi Bismuth 208.98038 [Xe]6p	84 ⁸⁴ Po Polonium (209) [Xe]6p	85 ⁸⁵ At Astatine (210) [Xe]6p	86 ⁸⁶ Rn Radon (222) [Xe]6p																																																																																																																	
7	87 ⁸⁷ Fr Francium (223) [Rn]7s	88 ⁸⁸ Ra Radium (226) [Rn]7s	104 ¹⁰⁴ Rf Rutherfordium (261) [Rn]5f ¹⁴ 6d ² 7s ²	105 ¹⁰⁵ Db Dubnium (262) [Rn]5f ¹⁴ 6d ³ 7s ²	106 ¹⁰⁶ Sg Seaborgium (266) [Rn]5f ¹⁴ 6d ⁴ 7s ²	107 ¹⁰⁷ Bh Bohrium (264) [Rn]5f ¹⁴ 6d ⁵ 7s ²	108 ¹⁰⁸ Hs Hassium (277) [Rn]5f ¹⁴ 6d ⁶ 7s ²	109 ¹⁰⁹ Mt Meitnerium (268) [Rn]5f ¹⁴ 6d ⁷ 7s ²	110 ¹¹⁰ Uun Ununnilium (281) [Rn]5f ¹⁴ 6d ⁸ 7s ²	111 ¹¹¹ Uuu Unununium (272) [Rn]5f ¹⁴ 6d ⁹ 7s ²	112 ¹¹² Uub Ununbium (285) [Rn]5f ¹⁴ 6d ¹⁰ 7s ²	114 ¹¹⁴ Uuq Ununquadium (289) [Rn]5f ¹⁴ 6d ¹⁰ 7s ² 7p ²	116 ¹¹⁶ Uuh Ununhexium (292) [Rn]5f ¹⁴ 6d ¹⁰ 7s ² 7p ⁴																																																																																																																					
			<table border="1" style="width: 100%; border-collapse: collapse; text-align: center;"> <thead> <tr> <th colspan="2">Lanthanides</th> <th colspan="14">Actinides</th> </tr> <tr> <th>Atomic Number</th> <th>Ground-state Level</th> <th>57</th><th>58</th><th>59</th><th>60</th><th>61</th><th>62</th><th>63</th><th>64</th><th>65</th><th>66</th><th>67</th><th>68</th><th>69</th><th>70</th><th>71</th> <th>89</th><th>90</th><th>91</th><th>92</th><th>93</th><th>94</th><th>95</th><th>96</th><th>97</th><th>98</th><th>99</th><th>100</th><th>101</th><th>102</th><th>103</th> </tr> </thead> <tbody> <tr> <td>58</td> <td>¹G₄</td> <td>La</td><td>Ce</td><td>Pr</td><td>Nd</td><td>Pm</td><td>Sm</td><td>Eu</td><td>Gd</td><td>Tb</td><td>Dy</td><td>Ho</td><td>Er</td><td>Tm</td><td>Yb</td><td>Lu</td> <td>Ac</td><td>Th</td><td>Pa</td><td>U</td><td>Np</td><td>Pu</td><td>Am</td><td>Cm</td><td>Bk</td><td>Cf</td><td>Es</td><td>Fm</td><td>Md</td><td>No</td><td>Lr</td> </tr> <tr> <td></td> <td></td> <td>Lanthanum 138.9055</td><td>Cerium 140.116 [Xe]4f¹5d¹6s²</td><td>Praseodymium 140.90765 [Xe]4f²6s²</td><td>Neodymium 144.24 [Xe]4f³6s²</td><td>Promethium (145) [Xe]4f⁴6s²</td><td>Samarium 150.36 [Xe]4f⁶6s²</td><td>Europium 151.964 [Xe]4f⁷6s²</td><td>Gadolinium 157.25 [Xe]4f⁷5d¹6s²</td><td>Terbium 158.92534 [Xe]4f⁹6s²</td><td>Dysprosium 162.500 [Xe]4f¹⁰6s²</td><td>Holmium 164.93032 [Xe]4f¹¹6s²</td><td>Erbium 167.259 [Xe]4f¹²6s²</td><td>Thulium 168.93421 [Xe]4f¹³6s²</td><td>Ytterbium 173.04 [Xe]4f¹⁴6s²</td><td>Lutetium 174.967 [Xe]4f¹⁴5d¹6s²</td> <td>Actinium (227) [Rn]6d¹7s²</td><td>Thorium 232.0381 [Rn]6d²7s²</td><td>Protactinium 231.03688 [Rn]5f²6d¹7s²</td><td>Uranium 238.02891 [Rn]5f³6d¹7s²</td><td>Neptunium (237) [Rn]5f⁴6d¹7s²</td><td>Plutonium (244) [Rn]5f⁶7s²</td><td>Americium (243) [Rn]5f⁷7s²</td><td>Curium (247) [Rn]5f⁸6d¹7s²</td><td>Berkelium (247) [Rn]5f⁹7s²</td><td>Californium (251) [Rn]5f¹⁰7s²</td><td>Einsteinium (252) [Rn]5f¹¹7s²</td><td>Fermium (257) [Rn]5f¹²7s²</td><td>Mendelevium (258) [Rn]5f¹³7s²</td><td>Nobelium (259) [Rn]5f¹⁴7s²</td><td>Lawrencium (262) [Rn]5f¹⁴7s²7p¹</td> </tr> </tbody> </table>																Lanthanides		Actinides														Atomic Number	Ground-state Level	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	89	90	91	92	93	94	95	96	97	98	99	100	101	102	103	58	¹ G ₄	La	Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu	Ac	Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr			Lanthanum 138.9055	Cerium 140.116 [Xe]4f ¹ 5d ¹ 6s ²	Praseodymium 140.90765 [Xe]4f ² 6s ²	Neodymium 144.24 [Xe]4f ³ 6s ²	Promethium (145) [Xe]4f ⁴ 6s ²	Samarium 150.36 [Xe]4f ⁶ 6s ²	Europium 151.964 [Xe]4f ⁷ 6s ²	Gadolinium 157.25 [Xe]4f ⁷ 5d ¹ 6s ²	Terbium 158.92534 [Xe]4f ⁹ 6s ²	Dysprosium 162.500 [Xe]4f ¹⁰ 6s ²	Holmium 164.93032 [Xe]4f ¹¹ 6s ²	Erbium 167.259 [Xe]4f ¹² 6s ²	Thulium 168.93421 [Xe]4f ¹³ 6s ²	Ytterbium 173.04 [Xe]4f ¹⁴ 6s ²	Lutetium 174.967 [Xe]4f ¹⁴ 5d ¹ 6s ²	Actinium (227) [Rn]6d ¹ 7s ²	Thorium 232.0381 [Rn]6d ² 7s ²	Protactinium 231.03688 [Rn]5f ² 6d ¹ 7s ²	Uranium 238.02891 [Rn]5f ³ 6d ¹ 7s ²	Neptunium (237) [Rn]5f ⁴ 6d ¹ 7s ²	Plutonium (244) [Rn]5f ⁶ 7s ²	Americium (243) [Rn]5f ⁷ 7s ²	Curium (247) [Rn]5f ⁸ 6d ¹ 7s ²	Berkelium (247) [Rn]5f ⁹ 7s ²	Californium (251) [Rn]5f ¹⁰ 7s ²	Einsteinium (252) [Rn]5f ¹¹ 7s ²	Fermium (257) [Rn]5f ¹² 7s ²	Mendelevium (258) [Rn]5f ¹³ 7s ²	Nobelium (259) [Rn]5f ¹⁴ 7s ²	Lawrencium (262) [Rn]5f ¹⁴ 7s ² 7p ¹
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[†]Based upon ¹²C. () indicates the mass number of the most stable isotope.

For a description of the data, visit physics.nist.gov/data

NIST SP 966 (September 2003)

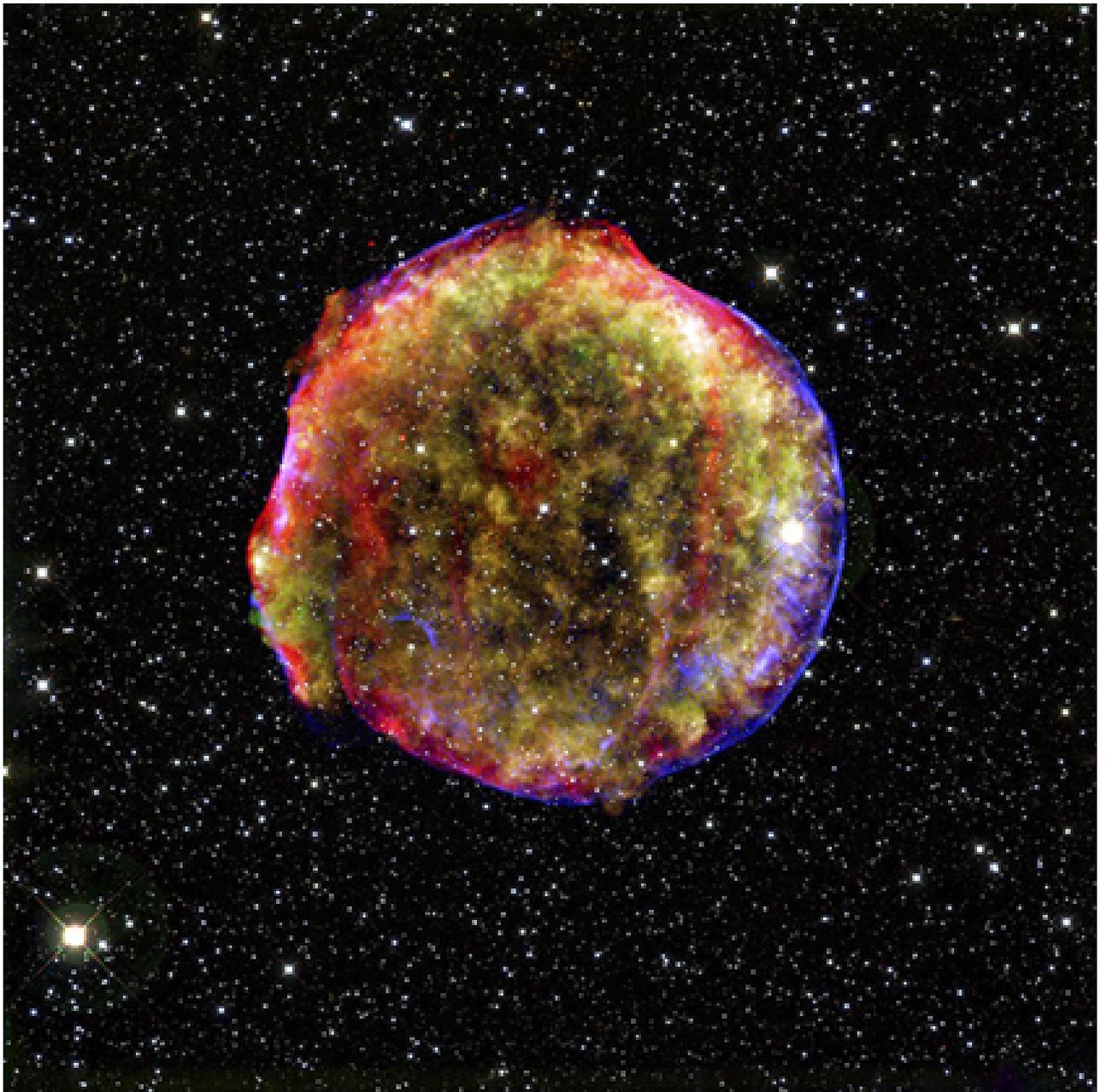
- Elements: Gases (pink), Solids (white), Liquids (blue), Artificially Prepared (yellow)

- Abundant Elements: H (X - 90%, mass:70%), He (Y - 7%, mass:28%)
- Other elements (metals) (Z- 3%, mass: 2%): Li, Be, B, C, N, O, F, Ne, Na, Mg, Al, Si, S, Ar, Ca, Ti, Cr, Fe, Ni
 - Fusion cycle in a star usually ends at Fe (strongest nuclear force)
- Study elements through spectroscopy
- Although constitute only 2% of the plasma, the metals are responsible for most of the spectral features
 - they crucially determine properties such as the plasma opacity that governs the transfer of radiation through the source

- Heavier elements - Supernova explosion
- Solar system made from debris of supernova explosions

TYCHO SUPERNOVA REMNANT IN CASSIOPIA

Low Resolution Observation: Spitzer (red - Infrared - dust),
Calar Alto (O, white stars),
Chandra (X-rays - yellow, green - Expanding
debris, blue - Ultra-energetic electrons)



SOLAR ABUNDANCES

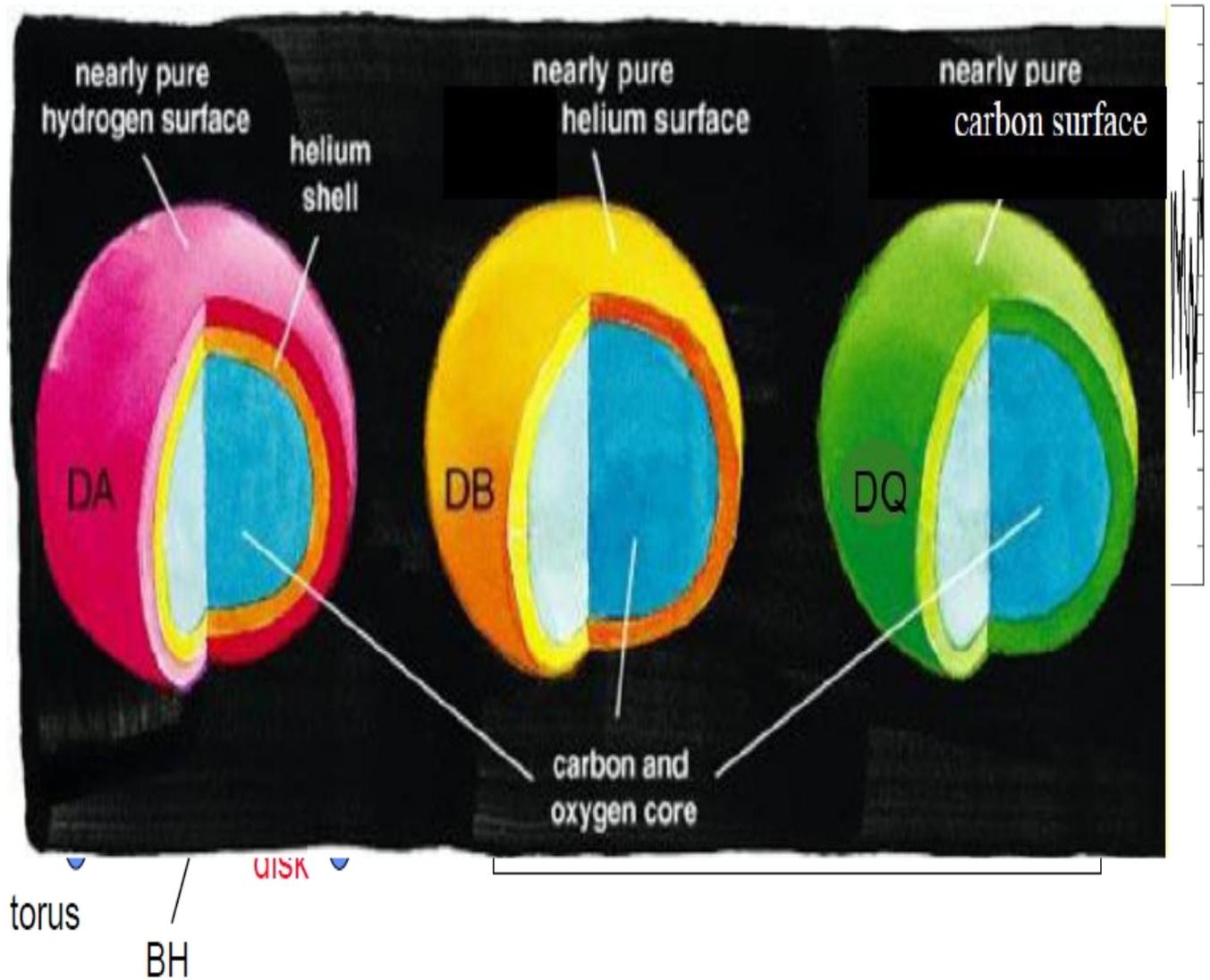
Solar abundances (A_k , usually in log scale):

- H - 90 % (by number), 70% (by mass fraction)
- He - 10 % (by number), 28 % (by mass)
- Metals - 2% (by mass). O (highest), C, N, Ne, Mg, Si, S, Ar, Fe
- Traditionally H abundance: $\log(A_H) = 12$, other elements are scaled relative to it
- Table: A mixture of “standard” solar abundances (10% uncertainties, Seaton et al. 1994)

Element (k)	Log A_k	A_k/A_H
H	12.0	1.0
He	11.0	1.00(-1)
C	8.55	3.55(-4)
N	7.97	9.33(-5)
O	8.87	7.41(-4)
Ne	8.07	1.18(-4)
Na	6.33	2.14(-6)
Mg	7.58	3.80(-5)
Al	6.47	2.95(-6)
Si	7.55	3.55(-5)
S	7.21	1.62(-5)
Ar	6.52	3.31(-6)
Ca	6.36	2.29(-6)
Cr	5.67	4.68(-7)
Mn	5.39	2.46(-7)
Fe	7.51	3.24(-5)
Ni	6.25	1.78(-6)

- **A supernova remnant:** an outer shimmering shell of expelled material, and a core skeleton of a once-massive star
 - Heavy elements, beyond Iron, are generated from nuclear fusion during supernova explosions and are scattered into interstellar medium
 - Two pure H & He clouds, Pristine Clouds, detected first in 2010
 - End of a star (98%): Below 1.4 solar mass (Chandra limit): Core becomes a white dwarf - diomond formation (electron repulsion equals gravity)
 - Above 1.4 but below 4 solar mass: neutron star (electrons pressed to Neutron formation - largher gravity)
- Larger mass - blackhole (extreme gravity)

White Dwarf stars -¿ cosmic clocks. The age of the universe is at least as big as the oldest stars in it. This principle can map the ages of galactic compoments



ii) Opacity and Observation

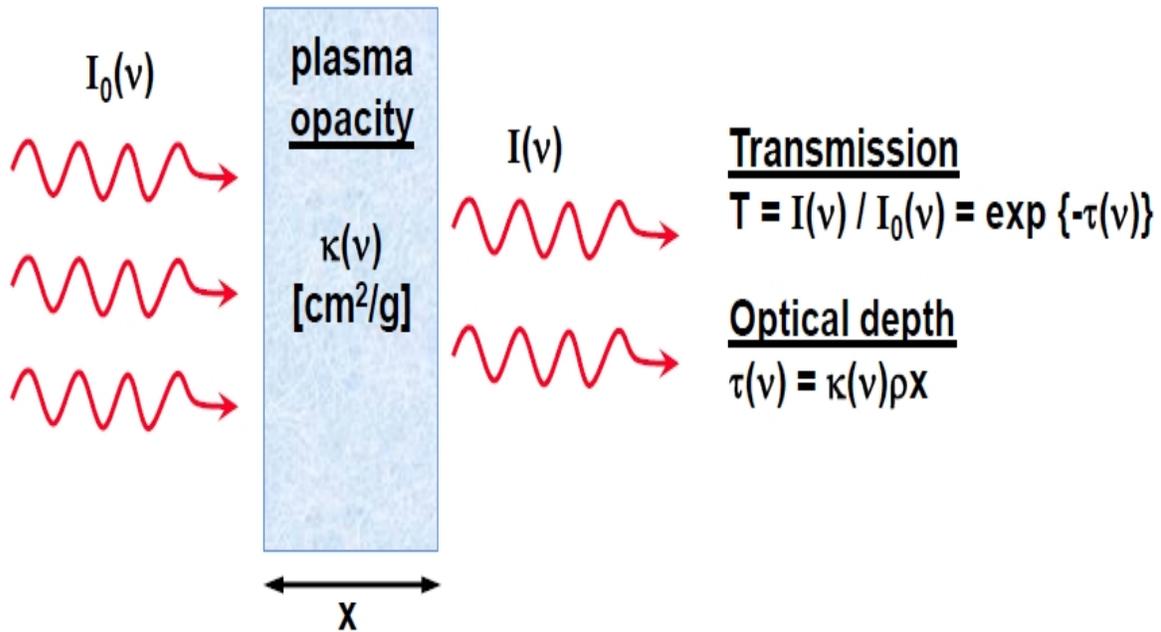
STUDYING ASTRONOMICAL OBJECTS

ASTRONOMICAL objects are studied in three ways:

- **Imaging:** - Beautiful pictures or images of astronomical objects, Stars, Nebulae, Active Galactic Nuclei (AGN), Blackhole Environments, etc
→ Provides information of size and location of the objects
- **Photometry:** Bands of Electromagnetic Colors ranging from X-ray to Radio waves
→ macroscopic information and low resolution spectroscopy
 - **Spectroscopy:** Taken by spectrometer - Provides most of the detailed knowledge: temperature, density, extent, chemical composition, etc. of astronomical objects

Spectroscopy is underpinned by Atomic & Molecular Physics

OPACITY

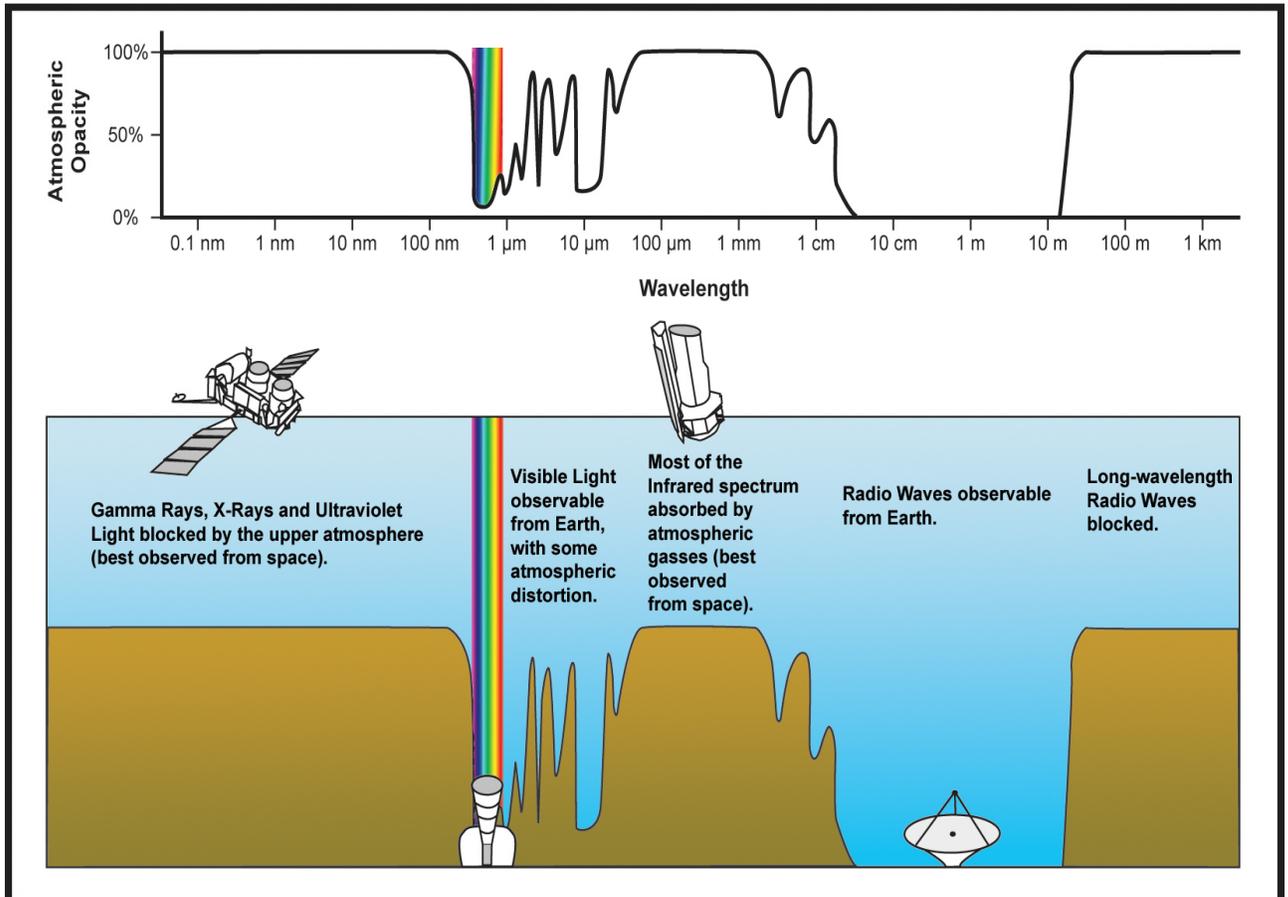


Stellar structure depends on opacities that have never been measured

- Opacity is a measure of radiation transfer
- Higher opacity - less radiation and lower opacity - more radiation transfer

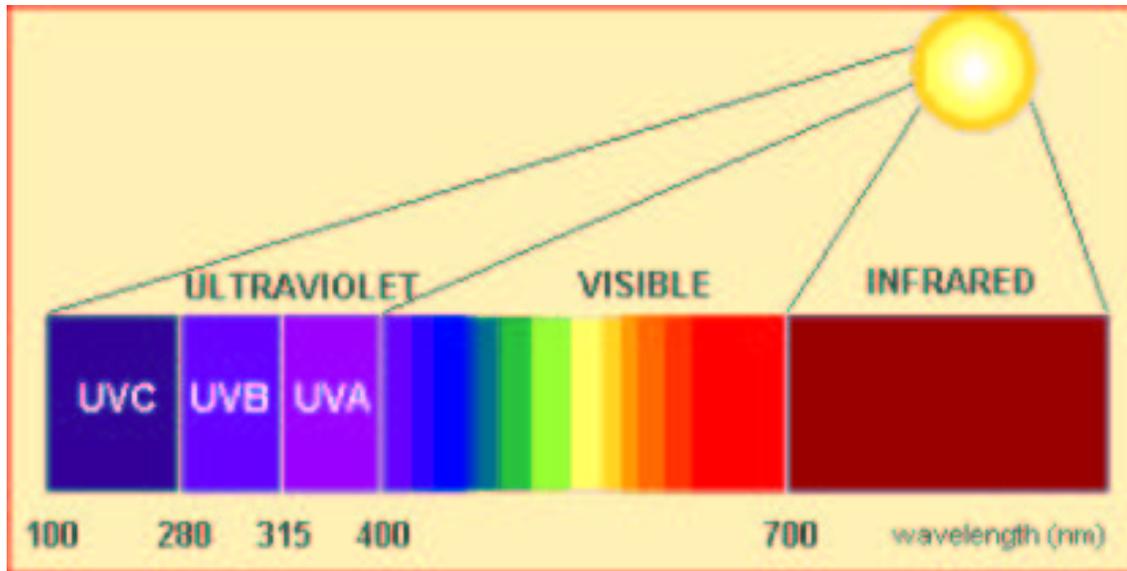
ATMOSPHERIC OPACITY

(www.ipac.caltech.edu/Outreach/Edu)

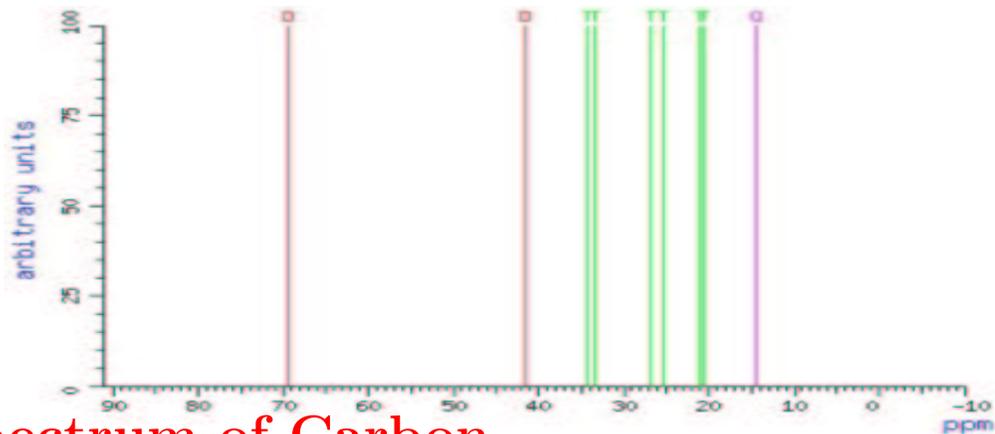


- Higher opacity - less radiation and lower opacity - more radiation reaching earth's surface
- Opacity determines types of telescopes - earth based or space based
- Gamma, X-ray, UV are blocked while visible light passes through
- CO₂, H₂O vapor, other gases absorb most of the infrared frequencies
- Part of radio frequencies is absorbed by H₂O and O₂, and part passes through

iii) The Sun and Evolution of Spectroscopy



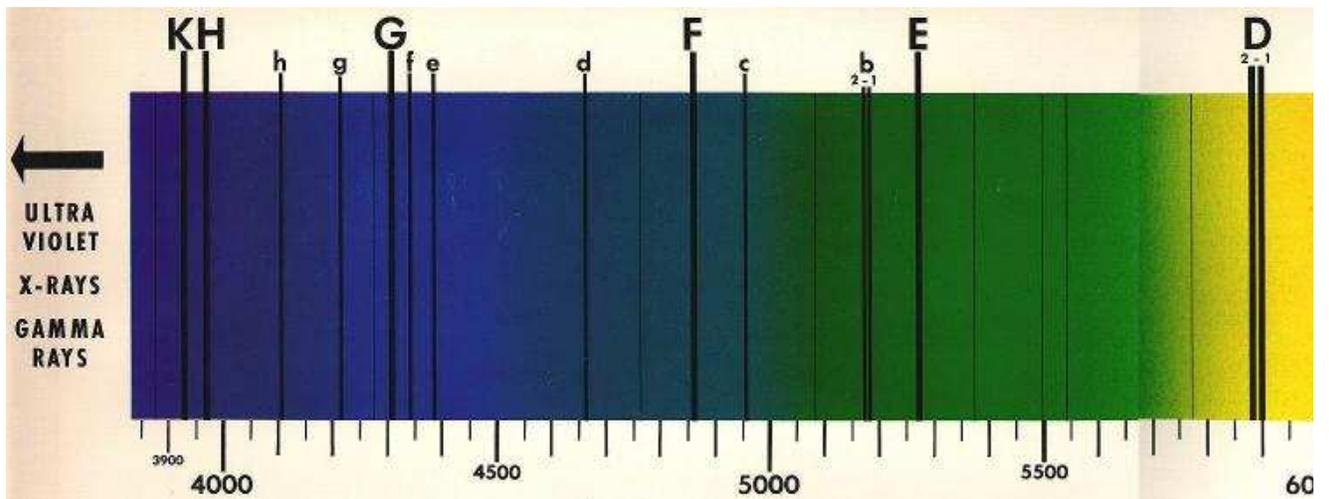
- Light is a mixture of colors - Spectrum: splitting of colors
For example: Rainbow is the spectrum of white sunlight
- Each atom gives out light with its own set of colors
- Spectrum is lines of colors, like the barcode on an item
- The lines → key information of astronomical objects the blackhole or other astrophysical objects



Spectrum of Carbon

Solar Spectra: Absorption & Emission Lines

- Absorption line - by absorption of a photon as an electron jumps to a higher level
- Emission line - by emission of a photon as an electron dumps to a lower level
- An absorption or emission line from the same two energy levels appear at the same energy position in the spectrum



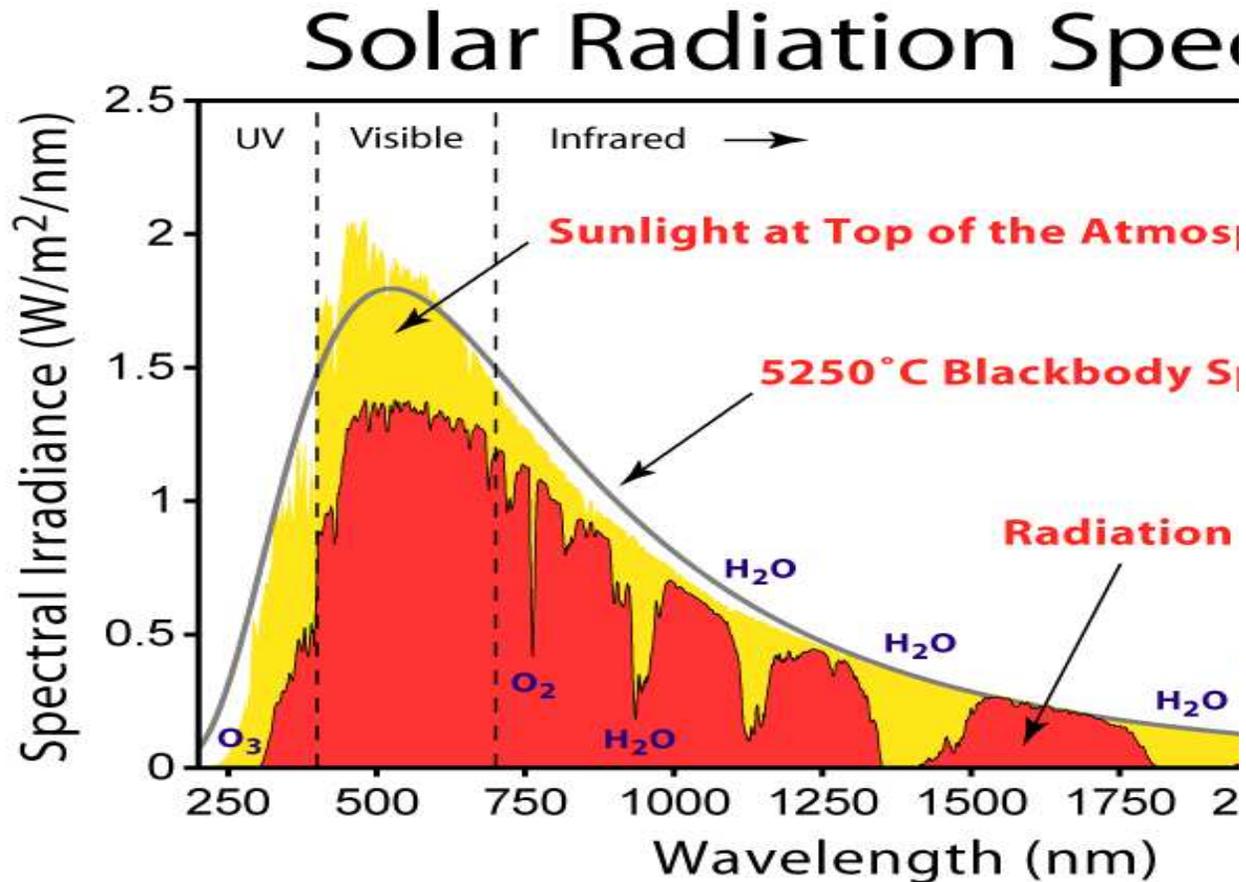
Fraunhofer lines (1815) in the Solar Spectrum

Used alphabet - no element assignment; Later spectroscopy with quantum mechanics: A (7594 Å, O), B (6867 Å, O) (air), C (6563 Å H), D1 & D2 (5896, 5890 Å Na, yellow sun), E (Fe I), G (CH), H & K (Ca II)

[(Courtesy: Institute for Astronomy, University of Hawaii)]

“SUNLIGHT” ON THE EARTH SURFACE

(http://globalwarmingart.com/wiki/Image:Solar_Spectrum.png)



- The Sun - Blackbody (constant T), gives out radiation in Planck distribution
- Solar surface $T_* = 5770 \text{ K} \rightarrow$ peak blackbody emission - yellow $\sim 5500 \text{ \AA}$.
- H_2O has inefficient absorption windows: 3 are molecular IR bands - J, H, and K bands
- Solar UV flux drops off rapidly by *ozone effect* preventing it to reach earth

iv) Electron and Photon Distribution Functions

- A plasma of charged particles and a radiation field of photons treated with two distribution functions, Planck and Maxwell
- ‘Temperature’ is defined through a radiation (photon) or the particle (electron) *energy distribution* → *kinetic* T defined from

$$E = h\nu \sim kT, \text{ or } E = 1/2mv^2 = 3/2kT$$

- Consider a star ionizing a molecular cloud into a gaseous nebula
 - The two objects, the star & the nebula, have different T distribution functions
 - Energy of the radiation emitted by the star - Planck distribution function
 - Energies of electrons in the surrounding ionized gas heated by the star - Maxwell distribution

PLANCK DISTRIBUTION FUNCTION

- total energy emitted by an object is related to its T by the Stefan-Boltzmann Law

$$E = \sigma T^4 \quad (1)$$

$\sigma = 5.67 \times 10^{-8}$ Watts/(m² K⁴) = Stefan constant

- Radiation of a star, a blackbody, is given by the Planck distribution function

$$B_\nu(T_*) = \frac{2h\nu^3}{c^2} \frac{1}{\exp(h\nu/kT_*) - 1}, \quad (2)$$

T_* = radiation temperature, ν = frequency of the photons.

MAXWELL DISTRIBUTION FUNCTION

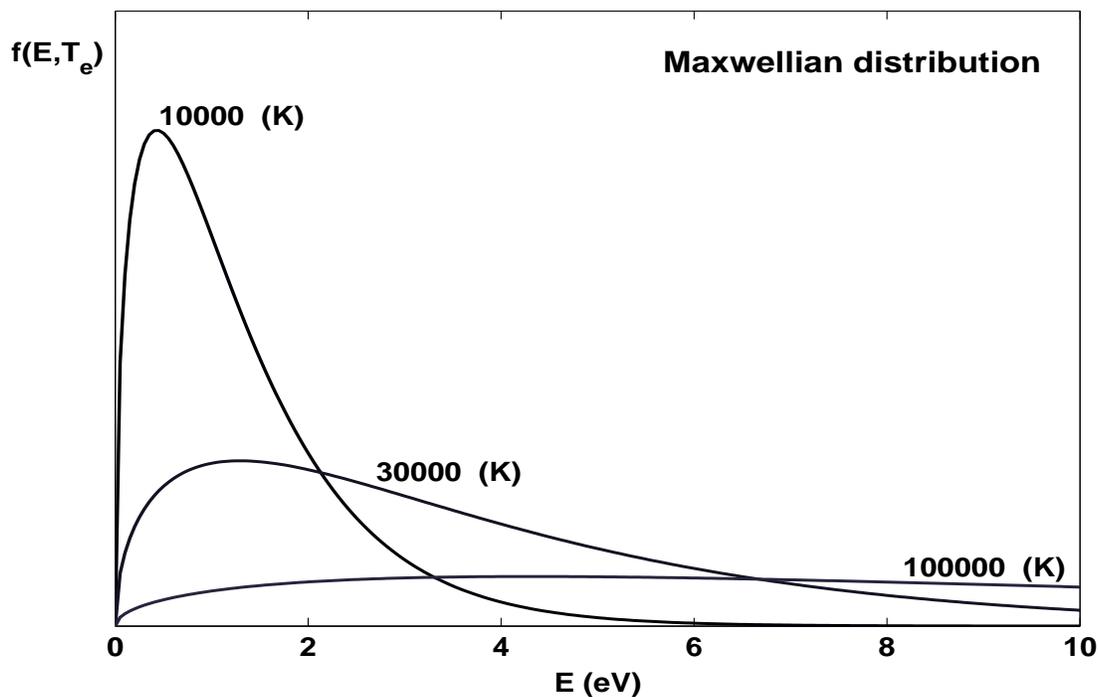
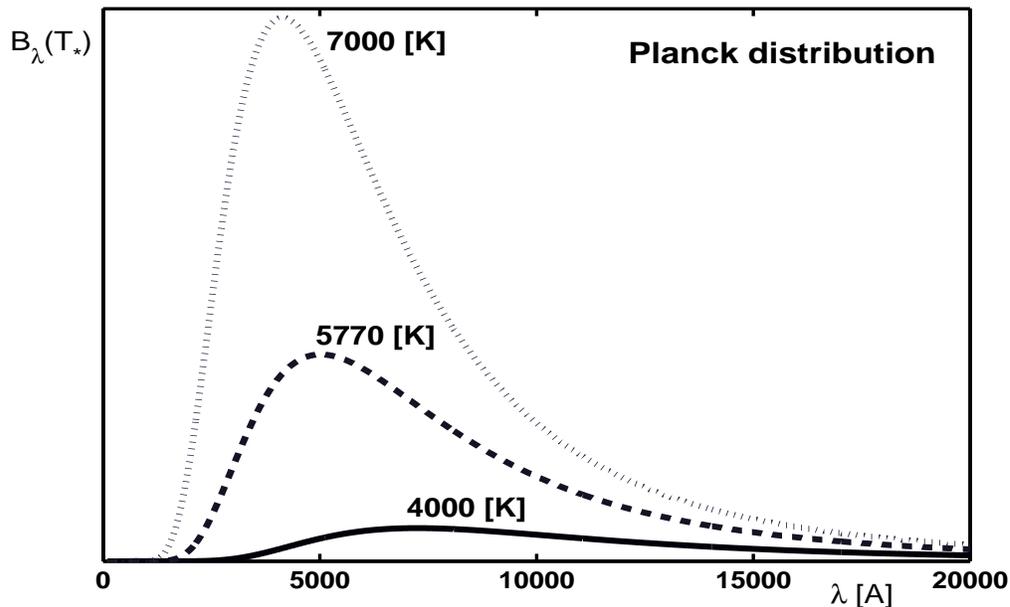
The charged particles in the plasma ionized by a star in an H II region has an *electron temperature* T_e associated with the mean kinetic energy of the electrons kT_e

$$\frac{1}{2}mv^2 = \frac{3}{2}kT_e. \quad (3)$$

The velocity (or energy) distribution of electrons is characterized by a Maxwellian function at temperature T_e ,

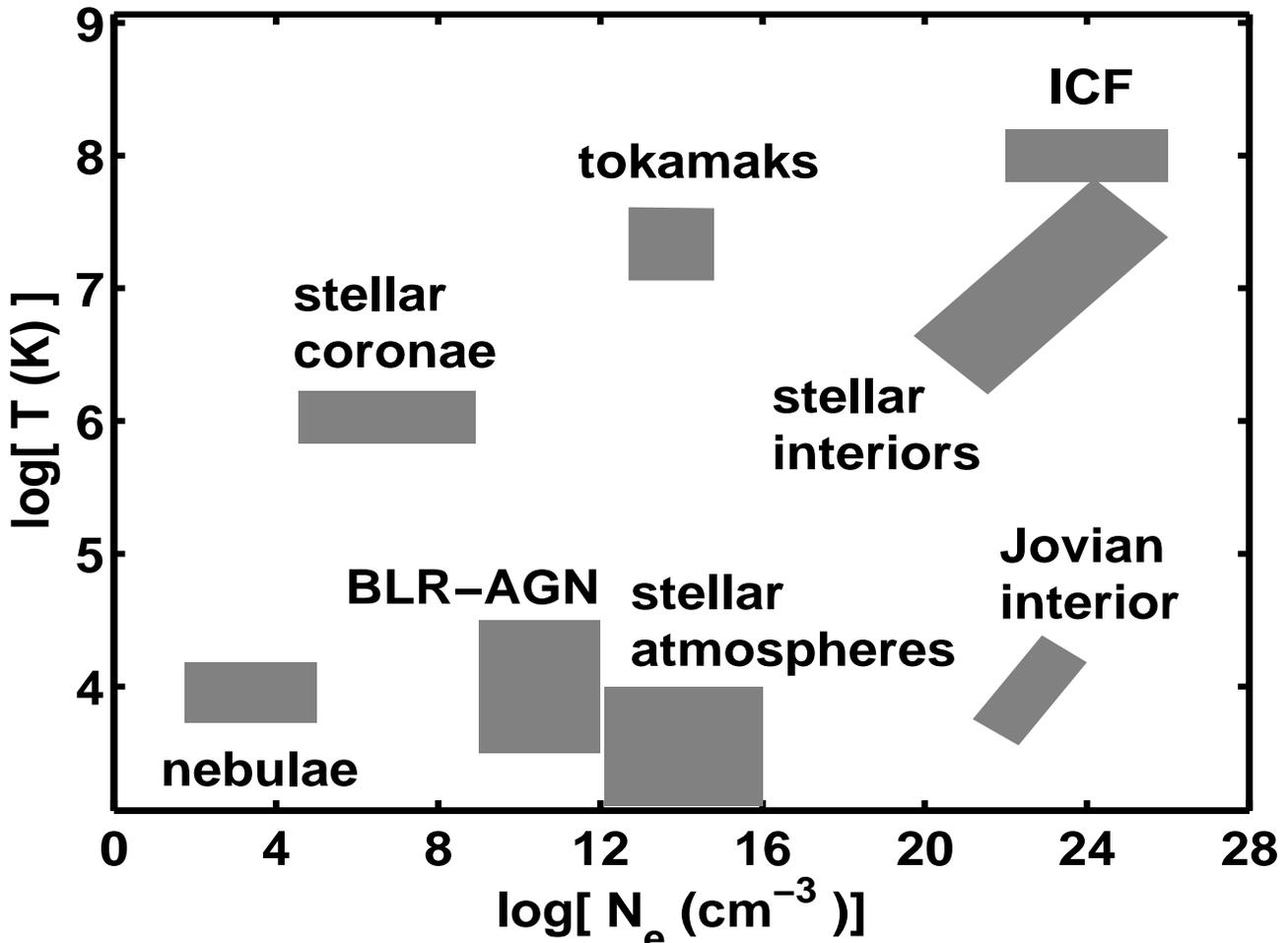
$$\mathbf{f}(\mathbf{v}) = \frac{4}{\sqrt{\pi}} \left(\frac{m}{2kT} \right)^{3/2} \mathbf{v}^2 \exp \left(-\frac{m\mathbf{v}^2}{2kT} \right). \quad (4)$$

PLANCK & MAXWELL DISTRIBUTIONS



- H II region (nebula): Ionized by black-body radiation of $T \sim 30000 - 40000$ K , electron kinetic energy of Maxwellian distribution at $T_e \approx 10000 - 20000$ K

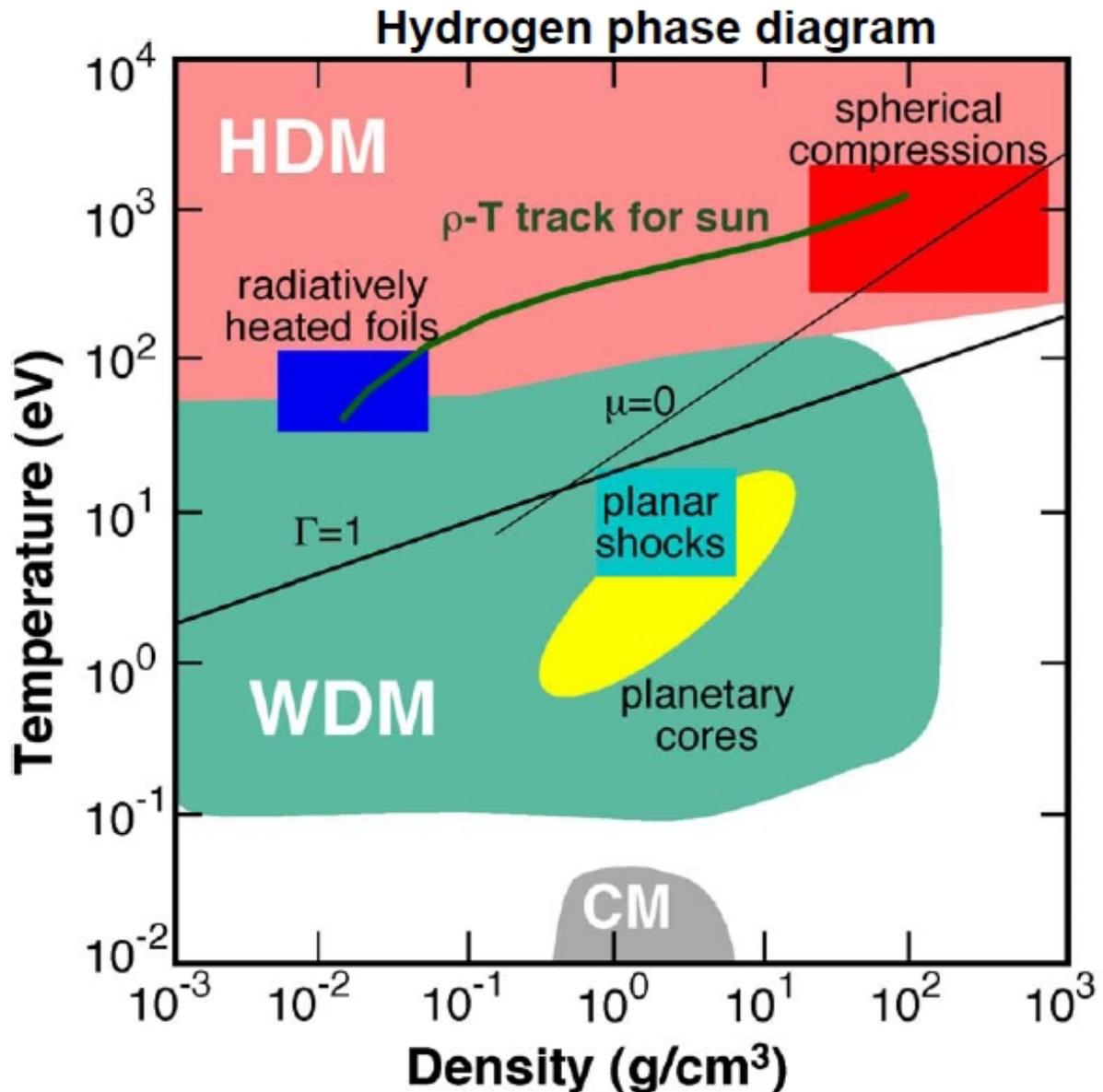
v) The Plasma Universe
PLASMA COVERS VAST REGION (99%) IN
T- ρ PHASE SPACE
(AAS, Pradhan & Nahar, 2011)



- Temperature-density regimes of plasmas in astrophysical objects - BLR-AGN (“broad-line regions) in active galactic nuclei”, where many spectral features associated with the central massive black hole activity manifest themselves
- Laboratory plasmas - tokamaks (magnetic confinement fusion devices), Z-pinch machines (inertial confinement fusion (ICF) devices)

WARM & HOT DENSE MATTER

(HEDLP-FESAC report)



Hot Dense Matter (HDM):

- Sun's ρ -T track, Supernovae, Stellar Interiors, Accretion Disks, Blackhole environments
- Laboratory plasmas in fusion devices: inertial confinement
- laser produced (NIF) and Z pinches (e.g. Sandia), magnetic confinement (tokamaks)

• Warm Dense Matter (WDM):

- cores of large gaseous planets

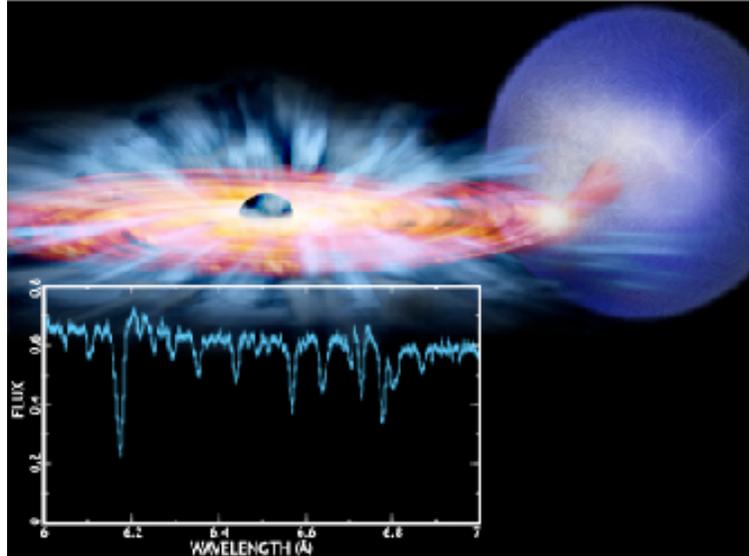
X-RAYS FROM A BLACKHOLE - CENTAURUS A GALAXY

(Observed by Chandra space telescope)

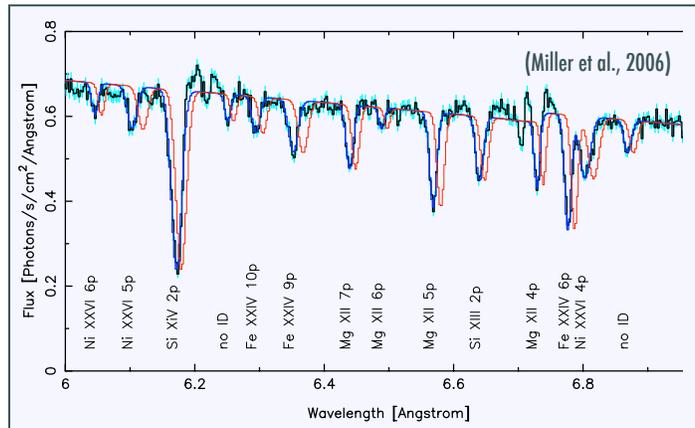


- Photometric image: red - low-energy X-rays, green - intermediate-energy X-rays, and blue - the highest-energy X-rays. The dark green and blue bands are dust lanes that absorb X-rays.
- Highly energetic SUPERHOT ATOMS encircling the blackhole are in a plasma state & emit bright K_{α} X-rays
- Electrons are ripped off - highly charged state, e.g. He- & Li-like Ne, Si, S, Fe; - emit mainly K_{α} (1s - 2p) photons
- Sucked in materials are ejected as a jet (L & E conservation)
- Blasting from the black hole a jet of a billion solar-masses extending to 13,000 light years

SPECTRUM of the Wind near Blackhole: GRO J1655-40 Binary Star System

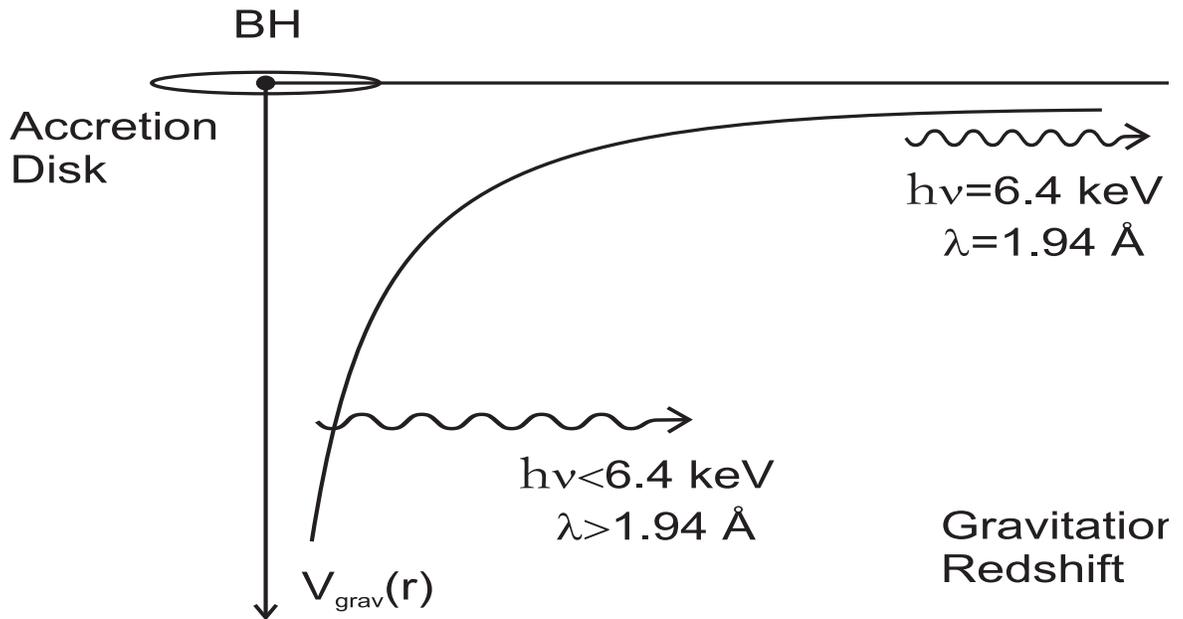
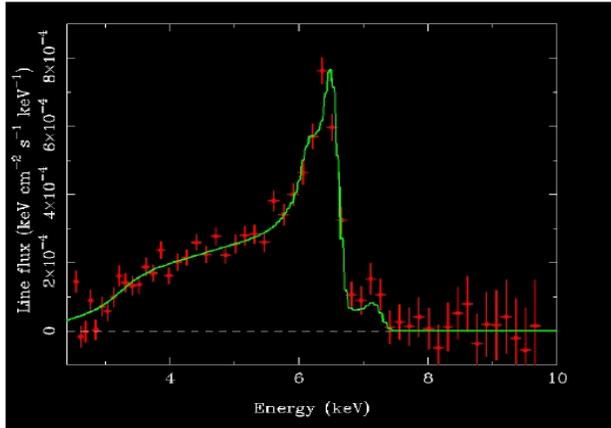


- Materials from the large star is sucked into the companion blackhole - form wind as they spiral to it



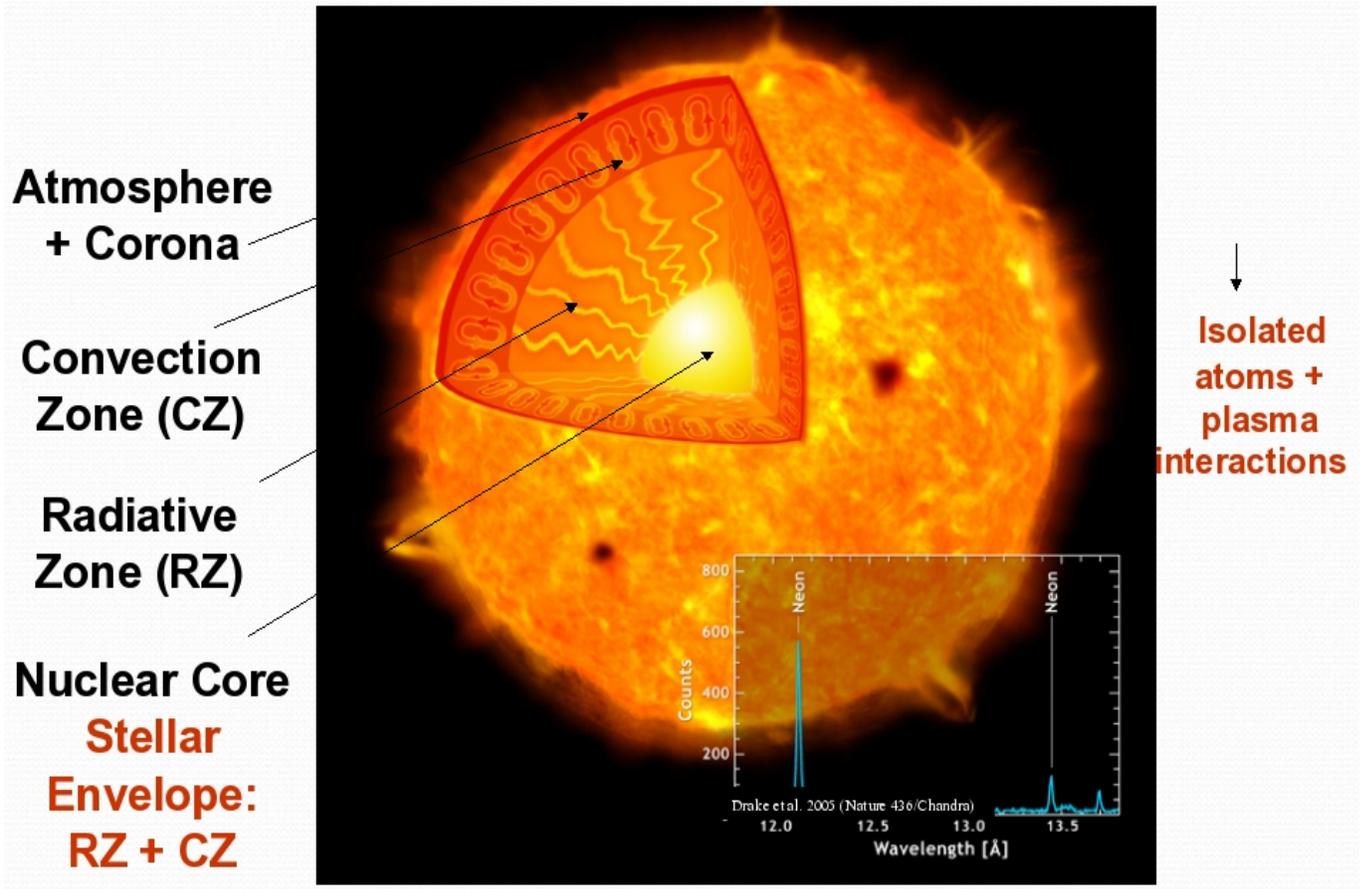
Spectrum: Highly charged Mg, Si, Fe, Ni lines
 Red Spectrum - Elements in natural widths
 Doppler Blue Shift - Wind is blowing toward us

Signature of a Blackhole



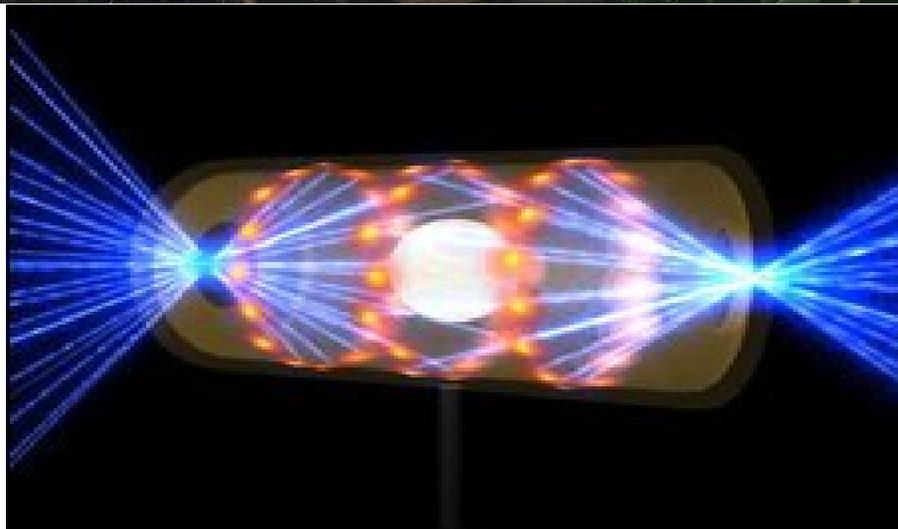
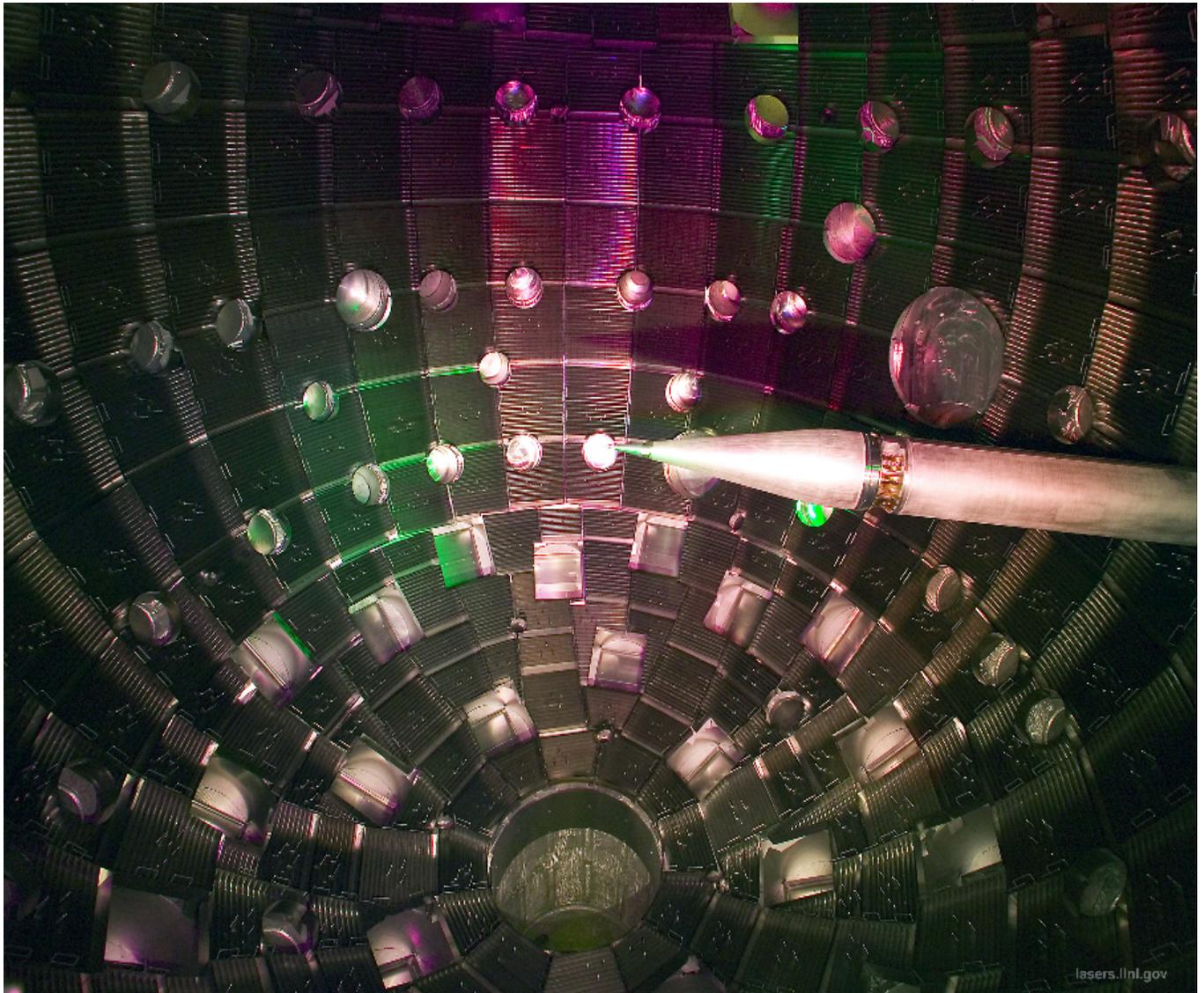
- The well-known K_{α} (1s-2p) transition array lines of iron near a black hole in Seyfert I galaxy MCG-6-30-15 6 (ASCA & Chandra obs)
- The maximum energy for a 1s-2p transition in iron is 6.4 keV. However, the large extension of the lines toward low energy means that the escaped photons have lost energies in the black hole.
- Gravitational broadening of Fe K_{α} emission line from close vicinity of the black hole.

SOLAR ABUNDANCES & HED PLASMA



- Sun's interior - nuclear core to the end of convection zone beyond which the radiation escapes
- At the convection zone boundary, R_{CZ} , the temperature $T_e \sim 193$ eV, density $n_e \sim 10^{23}/cm^3$ (HED - High Energy Density - Fe ions (Fe XVII-XIX))
- Takes over a million years for radiation created in the core (gamma rays) to travel to surface: **Reason - OPACITY**
- Although the most studied star, we can not explain all observations. Recent determination of solar abundances, from measurements and 3D hydro NLTE models, show 30-40% lower abundances of C, N, O, Ne, Ar than the standard abundances; contradict the accurate helioseismology data

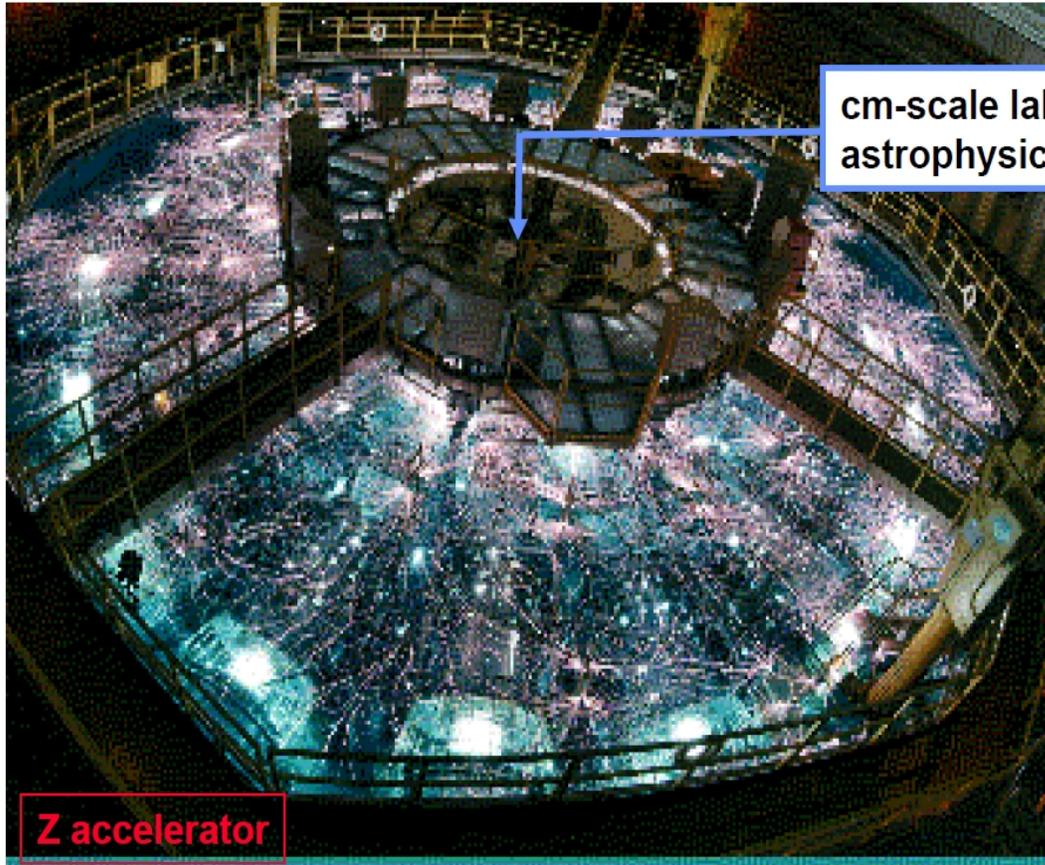
CREATION OF HED PLASMA: NIF, LLNL



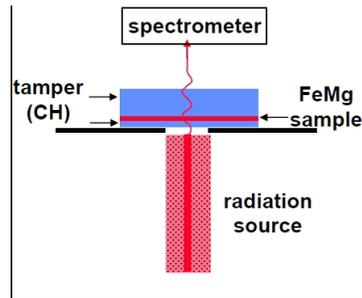
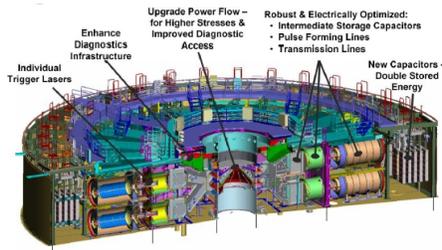
- NIF: Pulses from 192 high-power lasers arrive (in 10^{-12} s of each other) (TOP, spike - target position) to center target (mm-size)- 1.875MJ shot (Bottom)

HED PLASMA, Z PINCH, Sandia Natl lab

The 20million Amp current provided by the Z accelerator enables this research



40 m



New Z

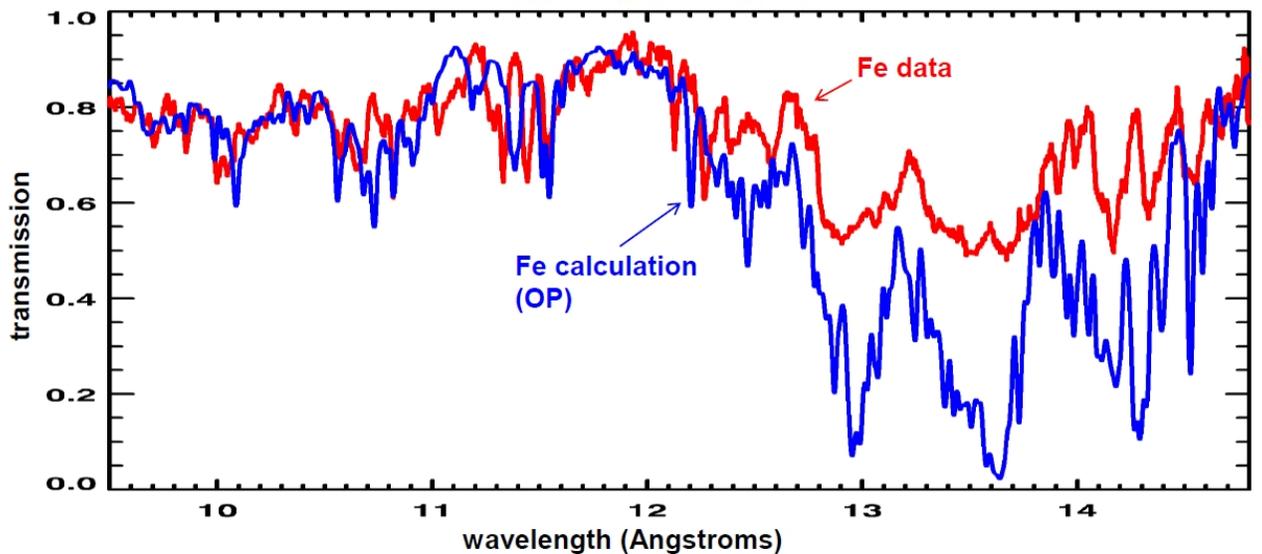
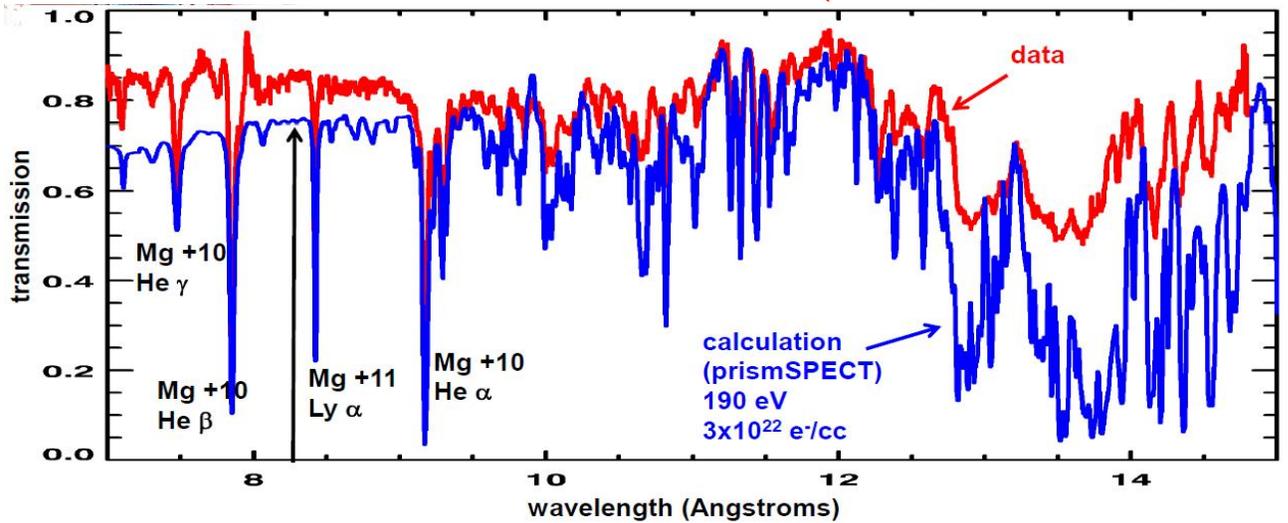
The refurbished Z delivers 24 million Amps to the load
50% increase in electrical energy for present day experiments

New sample design

Increasing the rear tamper thickness delays expansion onset
This leads to higher density and higher temperature

- Magnetically driven z-pinch implosions convert electrical energy into radiation. Internal shock heating → X-rays - Energy: 1.5 MJ, Power: 200 TW (Latest: ~ 27 Mamp)
- $T_e \sim 190 \text{ eV}$, density $n_e \sim 2.8 \times 10^{22} / \text{cm}^3$, similar to that at solar (R_{RZ})

DISCREPANCY BETWEEN EXPERIMENTAL & THEORETICAL OPACITIES (Bailey, Sandia lab)



- Z PINCH spectra, SNL: Plasma - $T_e \sim 193$ eV, Density $n_e \sim 10^{23}/\text{cm}^3$ - similar at solar R_{CZ}
- Observed (red), Calculated (blue). Top: Diagnostic lines, Bottom: Iron - Large differences **Reason: OPACITY**
- Serious discrepancies for Fe opacity
- Experimental n_e is wrong OR bound-free absorption (photoionization) is inaccurate

”ATOMIC PROCESSES FOR ASTROPHYSICAL SPECTROSCOPY”

1. Photo-Excitation & De-excitation:



- An Electron jumps up or down to level, but does not leave the atom
- **Fig:Top-** Energy levels C I, C II

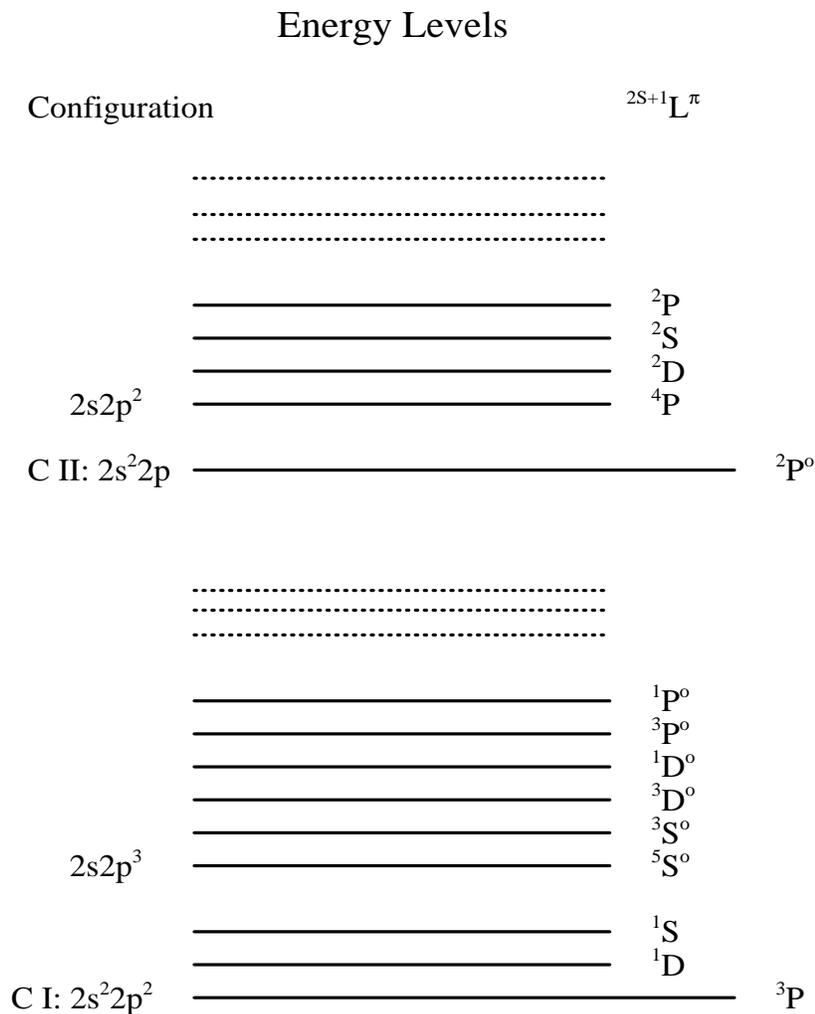
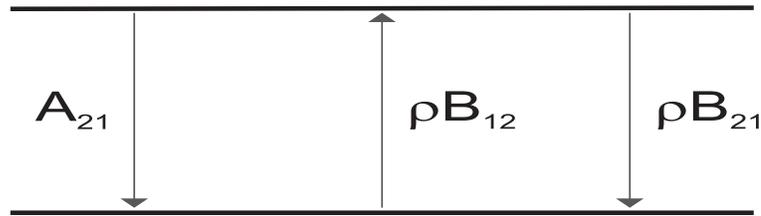


Photo-Excitation and Opacity



A_{21} - Spontaneous Decay

B_{21} - Stimulated Decay with a radiation field

- Introduces absorption or emission lines
- Atomic quantities (constant numbers):
 - Oscillator Strength (f)
 - Radiative Decay Rate (A -value)
- Monochromatic opacity (κ_ν) depends on f_{ij}

$$\kappa_\nu(\mathbf{i} \rightarrow \mathbf{j}) = \frac{\pi e^2}{mc} N_i f_{ij} \phi_\nu \quad (5)$$

N_i = ion density in state i , ϕ_ν = profile factor

- κ includes thousands to millions of transitions

ALLOWED & FORBIDDEN TRANSITION

Determined by angular momentum selection rules

i) Allowed: Electric Dipole (E1) transitions

($\Delta J = 0, \pm 1$, $\Delta L = 0, \pm 1, \pm 2$; parity changes, same-spin & different spin or intercombination transition)

Forbidden:

ii) Electric quadrupole (E2) transitions

($\Delta J = 0, \pm 1, \pm 2$, parity does not change)

iii) Magnetic dipole (M1) transitions

($\Delta J = 0, \pm 1$, parity does not change)

iv) Electric octupole (E3) transitions

($\Delta J = \pm 2, \pm 3$, parity changes)

v) Magnetic quadrupole (M2) transitions

($\Delta J = \pm 2$, parity changes)

Allowed transitions \gg *Forbidden transitions*

EX: ALLOWED & FORBIDDEN TRANSITIONS

Diagnostic Lines of He-like Ions: w,x,y,z

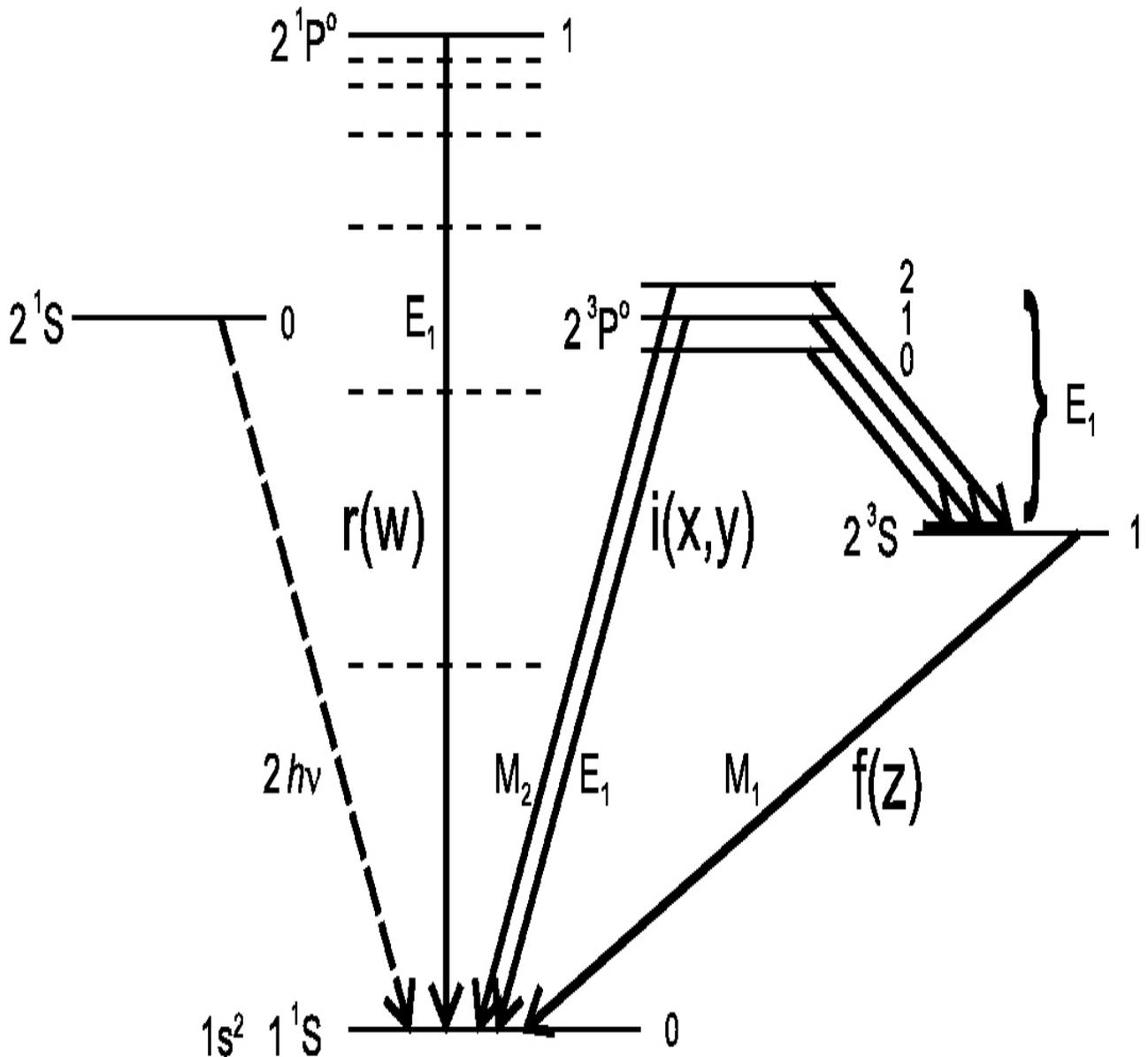
w(E1) : $1s2p(^1P_1^o) - 1s^2(^1S_0)$ (Allowed Resonant)

x(M2) : $1s2p(^3P_2^o) - 1s^2(^1S_0)$ (Forbidden)

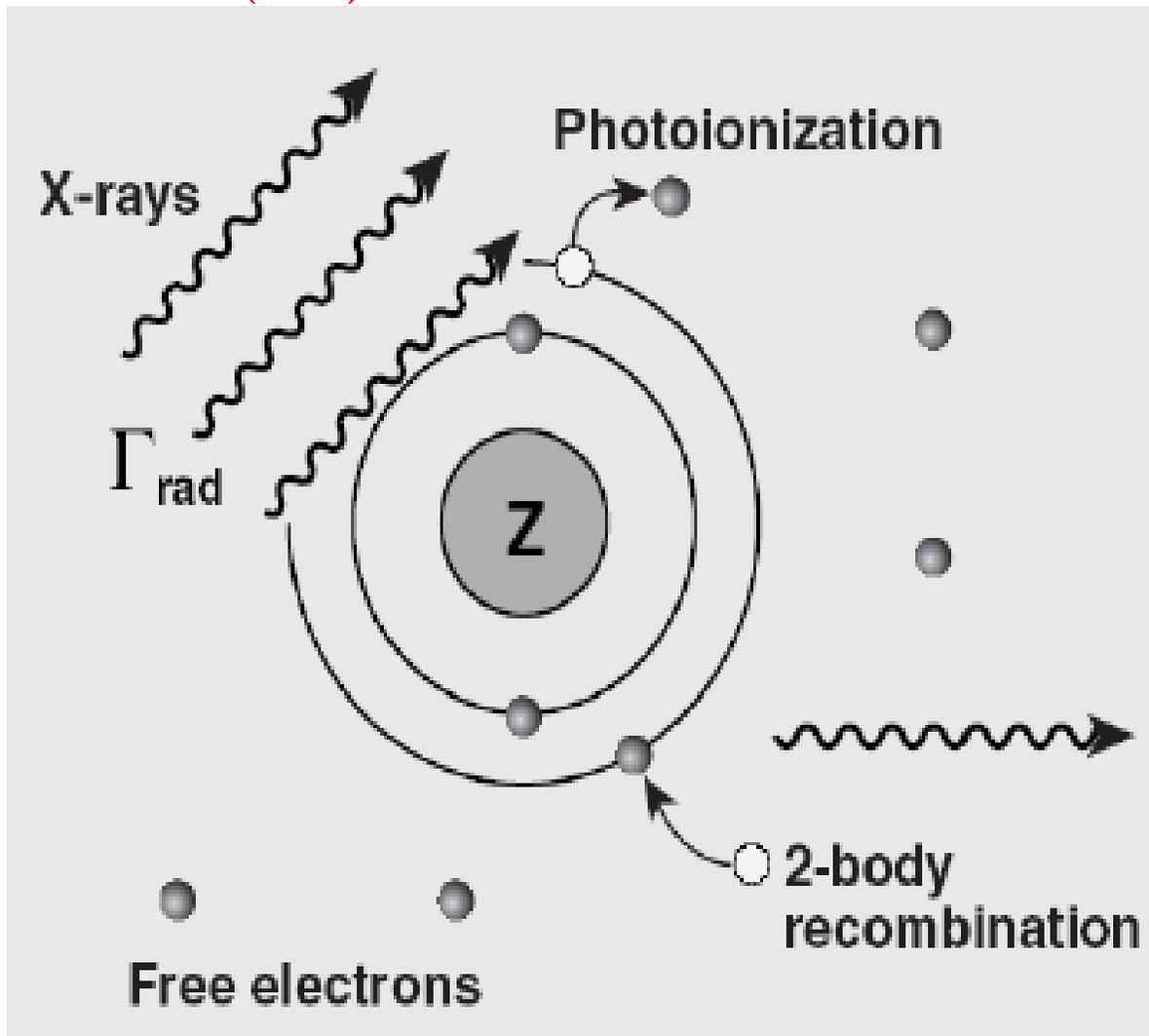
y(E1) : $1s2p(^3P_1^o) - 1s^2(^1S_0)$ (Intercombination)

z(M1) : $1s3s(^3S_1) - 1s^2(^1S_0)$ (Forbidden)

NOTE: 1s-2p are the K_α transitions



2 & 3. PHOTOIONIZATION (PI) & RECOMBINATION (RC):



Examples:

- Photoionization - in absorption spectra
- Recombination - in emission spectra

2 & 3. PHOTOIONIZATION (PI) & RECOMBINATION (RC): Inverse Processes

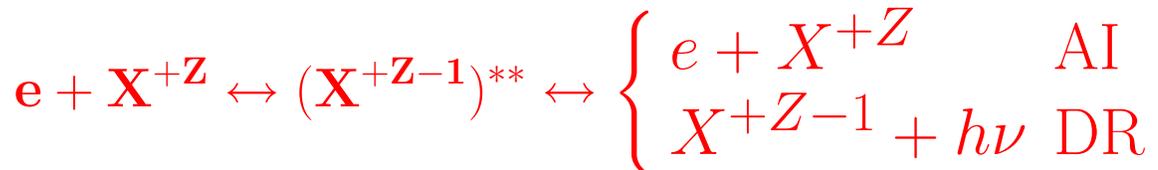
i) Direct Photoionization (PI) & Radiative Recombination (RR) (1-step Bound-Continuum transitions):



ii) Autoionization (AI) & Dielectronic recombination (DR) (2-steps with intermediate autoionizing state):



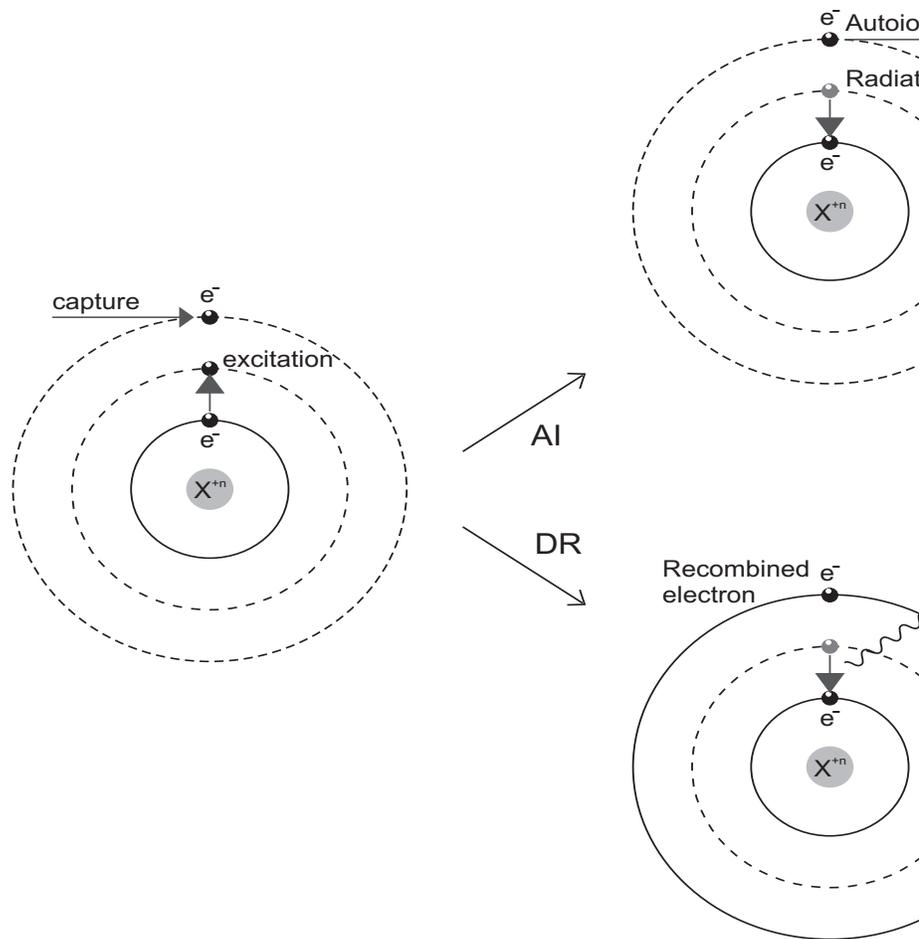
More exactly,



The doubly excited state "autoionizing state" introduces resonances

- i & ii Energetic electron leaves/combines the atom
- Photoionization Cross Sections (σ_{PI})
- Recombination Cross Sections (σ_{RC}) & Rate Coefficients (α_{RC})
- Show features with energy

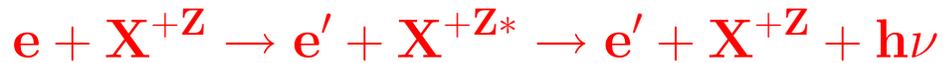
Fig. Photoionization, Autoionization, Recombination:



- Incoming electron energy - excites the core and binds the electron to an higher excited level - short lived Quasi-bound, doubly excited state (AI)
- outer electron goes free when core drops down
- core drops down with emssion of a photon -outer electron becomes bound (DR)

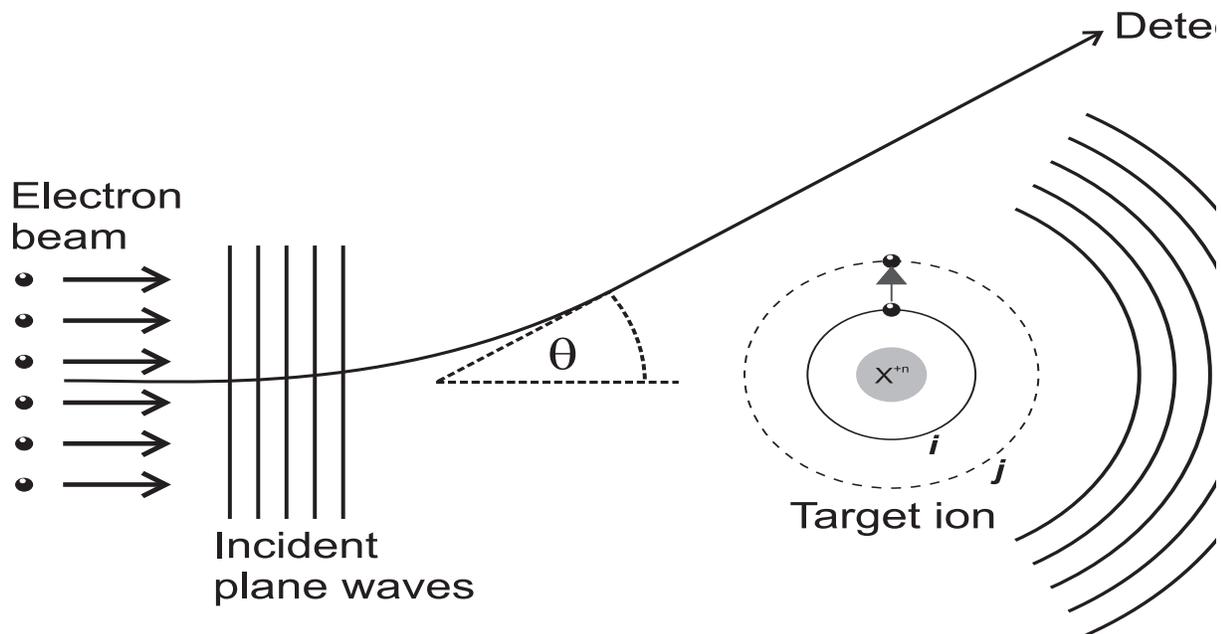
ELECTRON-ION COLLISIONS:

1. Electron-impact excitation (EIE):



- - Light is emitted as the excitation decays
- show most common lines in astrophysical spectra
- diagnostic forbidden lines
- Scattered electron shows features with energy
- Can go through an autoionizing state
- Atomic quantity: *Collision Strength* (Ω)

Fig. Excitation by electron impact:



2. Electron-impact Ionization (EII, important, but may not emit photon):



• **Atomic parameters:**

Ionization cross section (σ_{EII})

Ionization strength (S_{EII})

(often available experimentally)

THEORY

(*"Atomic Astrophysics and Spectroscopy", (AAS)
Pradhan and Nahar, Cambridge 2011*)

Hamiltonian: For a multi-electron system, the relativistic Breit-Pauli Hamiltonian is:

$$\mathbf{H}_{\text{BP}} = \mathbf{H}_{\text{NR}} + \mathbf{H}_{\text{mass}} + \mathbf{H}_{\text{Dar}} + \mathbf{H}_{\text{so}} + \frac{1}{2} \sum_{i \neq j}^{\text{N}} [\mathbf{g}_{ij}(\mathbf{so} + \mathbf{so}') + \mathbf{g}_{ij}(\mathbf{ss}') + \mathbf{g}_{ij}(\mathbf{css}') + \mathbf{g}_{ij}(\mathbf{d}) + \mathbf{g}_{ij}(\mathbf{q})]$$

where the non-relativistic Hamiltonian is

$$\mathbf{H}_{\text{NR}} = \left[\sum_{i=1}^{\text{N}} \left\{ -\nabla_i^2 - \frac{2Z}{r_i} + \sum_{j>i}^{\text{N}} \frac{2}{r_{ij}} \right\} \right] \quad (6)$$

the Breit interaction is

$$\mathbf{H}_{\text{B}} = \sum_{i>j} [\mathbf{g}_{ij}(\mathbf{so} + \mathbf{so}') + \mathbf{g}_{ij}(\mathbf{ss}')] \quad (7)$$

and one-body correction terms are

$$\mathbf{H}_{\text{mass}} = -\frac{\alpha^2}{4} \sum_i \mathbf{p}_i^4, \quad \mathbf{H}_{\text{Dar}} = \frac{\alpha^2}{4} \sum_i \nabla^2 \left(\frac{Z}{r_i} \right), \quad \mathbf{H}_{\text{so}} =$$

$$\frac{Ze^2\hbar^2}{2m^2c^2r^3}\text{L.S}$$

Wave functions and energies are obtained solving:

$$\mathbf{H\Psi = E\Psi}$$

- $\mathbf{E < 0} \rightarrow$ Bound (e+ion) states Ψ_B
- $\mathbf{E \geq 0} \rightarrow$ Continuum states Ψ_F

Transition Matrix elements with a Photon

Dipole operator: $\mathbf{D} = \sum_i \mathbf{r}_i$:

$\langle \Psi_B || \mathbf{D} || \Psi_{B'} \rangle \rightarrow$ **Photo-excitation & De-excitation**

$\langle \Psi_B || \mathbf{D} || \Psi_F \rangle \rightarrow$ **Photoionization**

Matrix element is reduced to generalized line strength

$$S = \left| \left\langle \Psi_f \left| \sum_{j=1}^{N+1} \mathbf{r}_j \right| \Psi_i \right\rangle \right|^2 \quad (8)$$

Oscillator Strength, Photoionization, Re-combination all are related to S

Evaluation of the transition matrix element with a radiation field is non-trivial because of the exponential operator $e^{i\mathbf{k}\cdot\mathbf{r}}$ describing the field

$$e^{i\mathbf{k}\cdot\mathbf{r}} = 1 + i\mathbf{k}\cdot\mathbf{r} + [i\mathbf{k}\cdot\mathbf{r}]^2/2! + \dots, \quad (9)$$

- Various terms in the exponential expansion corresponds to various multipole transitions. The first term gives the electric dipole transitions E1, 2nd term E2 and M1, ...
- The selection rules arise from the angular algebra of the transition matrices.

- The exponential factor $e^{i\mathbf{k}\cdot\mathbf{r}}$ in the transition matrix element gives the spatial dependence of the incident wave.

An electron at a distance from the nucleus is of the order of Bohr radius, 10^{-8} cm.

The magnitude of the wave vector k is $2\pi/\lambda$. For visible light k is of the order of 10^5 cm^{-1}

→ the wavelength is much larger than the size of an atom

Retaining only the term unity in the first expansion, leads to the *electric dipole approximation*:

$$e^{i\mathbf{k}\cdot\mathbf{r}} \approx 1.$$

f-, S, A-VALUES FOR VARIOUS TRANSITIONS

Allowed electric dipole (E1) transitions

The oscillator strength (f_{ij}) and radiative decay rate (A_{ji}) for the bound-bound transition are

$$f_{ij} = \left[\frac{E_{ji}}{3g_i} \right] S, \quad A_{ji}(\text{sec}^{-1}) = \left[0.8032 \times 10^{10} \frac{E_{ji}^3}{3g_j} \right] S \quad (10)$$

i) Same spin multiplicity dipole allowed ($\Delta j=0,\pm 1$, $\Delta L = 0, \pm 1, \pm 2$, $\Delta S = 0$, parity π changes)

ii) Intercombination ($\Delta j=0,\pm 1$, $\Delta L = 0, \pm 1, \pm 2$, $\Delta S \neq 0$, π changes)

Forbidden transitions

i) Electric quadrupole (E2) transitions ($\Delta J = 0,\pm 1,\pm 2$, parity does not change)

$$A_{ji}^{E2} = 2.6733 \times 10^3 \frac{E_{ij}^5}{g_j} S^{E2}(i, j) s^{-1}, \quad (11)$$

ii) Magnetic dipole (M1) transitions ($\Delta J = 0, \pm 1$, parity does not change)

$$A_{ji}^{M1} = 3.5644 \times 10^4 \frac{E_{ij}^3}{g_j} S^{M1}(i, j) s^{-1}, \quad (12)$$

iii) Electric octupole (E3) transitions ($\Delta J = \pm 2, \pm 3$, parity changes)

$$A_{ji}^{E3} = 1.2050 \times 10^{-3} \frac{E_{ij}^7}{g_j} S^{E3}(i, j) s^{-1}, \quad (13)$$

iv) Magnetic quadrupole (M2) transitions ($\Delta J = \pm 2$, parity changes)

$$A_{ji}^{M2} = 2.3727 \times 10^{-2} s^{-1} \frac{E_{ij}^5}{g_j} S^{M2}(i, j). \quad (14)$$

LIEFTIME:

The lifetime of a level can be obtained from the A-values,

$$\tau_k(s) = \frac{1}{\sum_i A_{ki}(s^{-1})}. \quad (15)$$

Photoionization & Recombination cross section:

Photoionization cross section is given by

$$\sigma_{\text{PI}}(\mathbf{K}\alpha, \nu) = \frac{4\pi^2 \mathbf{a}_0^2 \alpha \mathbf{E}_{ij}}{3 \mathbf{g}_k} \mathbf{S} \quad (16)$$

In central-field approximation, the photoionization cross section in length form is given by,

$$\sigma_{\text{L}} = \frac{4\pi^2 \alpha \omega_{ij}}{3} \frac{1}{2l_i + 1} \left[l_i \left| \mathbf{R}_{n_i l_i}^{n_j, l_i+1} \right|^2 + (l_i + 1) \left| \mathbf{R}_{n_i l_i}^{n_j, l_i-1} \right|^2 \right]. \quad (17)$$

A general expression for the photoionization cross section of level n of a hydrogenic system is (Seaton 1959)

$$\sigma_n = \frac{64\pi \alpha \mathbf{a}_0^2}{3\sqrt{3}} \left(\frac{\omega_n}{\omega} \right)^3 \frac{\mathbf{ng}(\omega, n, l, \mathbf{Z})}{\mathbf{Z}^2}, \quad (18)$$

where

$$\omega > \omega_n = \frac{\alpha^2 m c^2 \mathbf{Z}^2}{2\hbar n^2}, \quad (19)$$

and $g(\omega, n, l, Z)$ is the bound-free Gaunt factor which is approximately unity at near-threshold energies. A more detailed expression for photoionization of a state nl is

$$\sigma_{nl} = \frac{512\pi^7 m_e^{10}}{3\sqrt{3} c h^6} \frac{g(\omega, n, l, \mathbf{Z}) \mathbf{Z}^4}{n^5 \omega^3} \quad (20)$$

The photoionization cross section can be approximated by using hydrogenic wavefunctions that yield

the Kramer's formula,

$$\sigma_{\text{PI}} = \frac{8\pi}{3^{1.5}} \frac{1}{c n^5 \omega^3} \quad (21)$$

This is sometimes used to extrapolate photoionization cross sections in the high-energy region, where other features have diminished contribution with a relatively smooth background. However, it is not accurate.

Since the spin does not change in a dipole transition, and $S_i = S_j$, the angular algebra remains the same for both the bound-bound and bound-free transitions. So we have the same selection rules for photoionization,

$$\begin{aligned} \Delta L &= L_j - L_i = 0, \pm 1, \quad \Delta M = M_{L_j} - M_{L_i} = 0, \pm 1, \\ \Delta l &= l_j - l_i = \pm 1. \end{aligned} \quad (22)$$

The length photoionization cross section is,

$$\begin{aligned} \sigma_L &= \frac{\pi^2 c^2}{n_\omega \omega_{ij}^2} T_{ij} \\ &= \frac{4\pi^2 \alpha}{3} \omega_{ij} \sum_{l_j=l_i \pm 1} \sum_{L_j} (2L_j + 1) \frac{(l_i + l_j + 1)}{2} W^2(l_i L_i l_j L_j; L_1 1) | \\ &< n_j l_j | r | n_i l_i >|^2 \end{aligned} \quad (23)$$

With $\alpha = e^2/(\hbar c)$ the cross section is in units of a_o^2 and the constant in the equation is $\frac{4\pi^2 a_o^2 \alpha}{3} = 2.689$ Megabarns,

abbreviated as $Mb = 10^{-18} \text{ cm}^2$ The radial integral

$$\langle n_j l_j | r | n_i l_i \rangle = \int_0^\infty R_{n_j l_j} r R_{n_i l_i} r^2 dr, \quad (24)$$

involves an exponentially decaying initial bound state wavefunction, and an oscillating continuum free-electron wavefunction as a plane wave. The bound state wavefunctions are normalized to $|\langle i | i \rangle|^2 = 1$, and free state wavefunctions are normalized per unit energy

Recombination cross section, σ_{RC} , from principle of detailed balance:

$$\sigma_{RC} = \sigma_{PI} \frac{g_i}{g_j} \frac{h^2 \omega^2}{4\pi^2 m^2 c^2 v^2}. \quad (25)$$

The recombination rate coefficient:

$$\alpha_{RC}(\mathbf{T}) = \int_0^\infty v f(v) \sigma_{RC} dv, \quad (26)$$

$f(\mathbf{v}, \mathbf{T}) = \frac{4}{\sqrt{\pi}} \left(\frac{m}{2kT}\right)^{3/2} v^2 e^{-\frac{mv^2}{2kT}} =$ Maxwellian velocity distribution function

Recombination Rate Coefficient in terms of photoelectron energy is

$$\alpha_{RC}(\mathbf{E}) = v \sigma_{RC}$$

Total α_{RC} \rightarrow Contributions from infinite number of recombined states

- Radiative & Dielectronic Recombinations are inseparable in nature

UNIFIED TREATMENT FOR ELECTRON-ION RECOMBINATION

(Nahar & Pradhan, 1992, 1994, Zhang, Nahar, Pradhan, 1999)

EXISTING METHODS: $\alpha_{RC} = \alpha_{RR} + \alpha_{DR}$

UNIFIED METHOD: Total α_{RC} - subsumes RR and DR including interference effects

- Considers infinite number of recombined states:
 - i) Group (A) Low-n Bound States ($n \leq n_o \sim 10$):
 σ_{RC} are obtained from σ_{PI} which includes autoionizing resonances. Hence subsumes both RR & DR
 - ii) Group (B) High-n States ($n_o \leq n \leq \infty$):
 These densely packed highly excited states, dominated by DR process, are treated through DR theory (Bell & Seaton 1985, Nahar & Pradhan 1994) in close-coupling approximation
- **Advantages of the "Unified method":**
 - Self-consistent treatment for photoionization & recombination - use of same wavefunction
 - Provides "level-specific" recombination rate coefficients & photoionization cross sections for a large number of bound levels

Ionization Fractions in Astrophysical Plasma Equilibria: Unified Treatment of Photoionization & Electron-Ion Recombination

i) Plasma in Coronal Equilibrium:

Electron-Impact Ionization (EII) is balanced by Electron Ion Recombination:

$$N(z-1)S_{\text{EII}}(z-1) = N(z)\alpha_{\text{RC}}(z)$$

$S_{\text{EII}}(z-1)$ = total EII rate coefficient, α_{RC} = total recombination rate coefficient

ii) Plasma in Photoionization Equilibrium:

With a radiation flux J_ν , photoionization is balanced by electron-ion recombination:

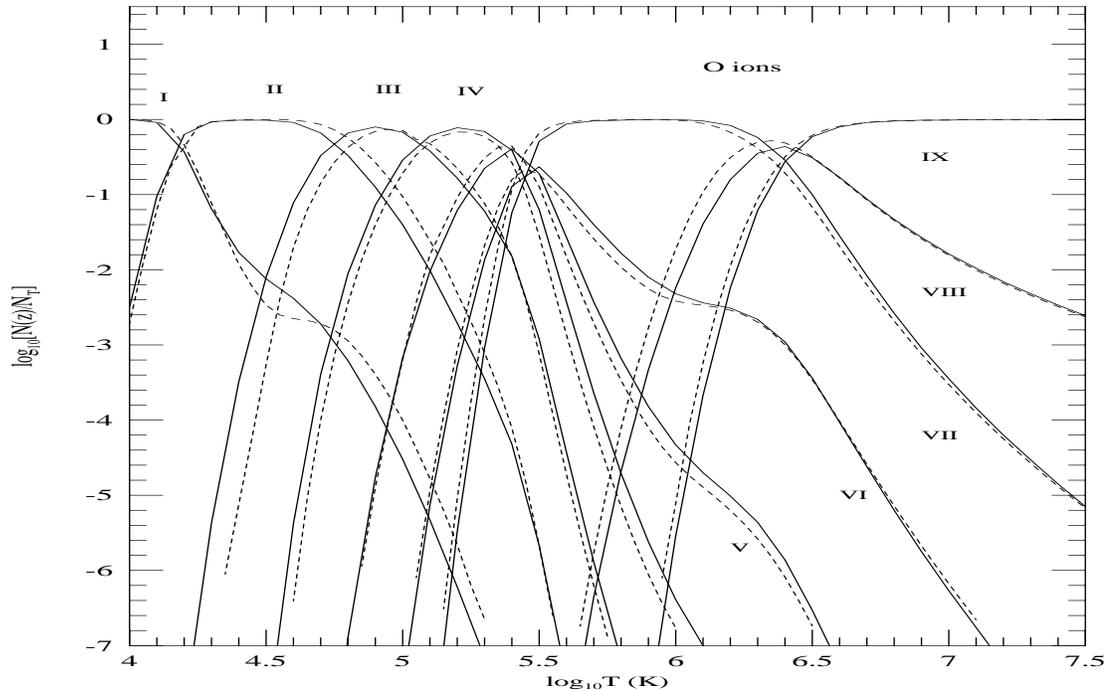
$$N(z) \int_{\nu_0}^{\infty} \frac{4\pi J_\nu}{h\nu} \sigma_{\text{PI}}(z, \nu) d\nu = N_e N(z+1) \alpha_{\text{RC}}(z, T_e)$$

ν_0 = Ionization potential of the ion, σ_{PI} = Ground state photoionization cross sections

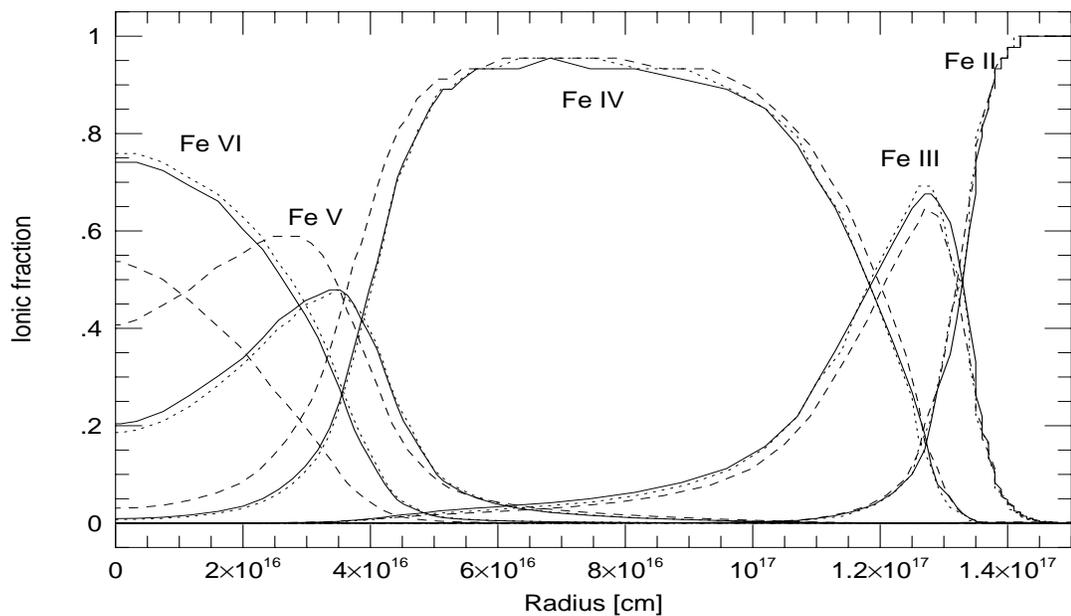
Accurate ionization fractions, N_i/N , requires

- Unified treatment of *total* (RR+DR) rate coefficient
- Self-consistency between these inverse processes

Ionization Fractions at Coronal Equilibrium: O-ions (Nahar 1998)



Iron Ionization Fractions at Photoionization Equilibrium in a typical PNe: Fe IV dominates in most of the nebula, Fe V fraction is reduced by half (Nahar & Bautista 1995)



EIE:

Scattering matrix $\mathbf{S}_{SL\pi}(i, j)$ is obtained from the wave function phase shift. The EIE collision strength is

$$\Omega(\mathbf{S}_i\mathbf{L}_i - \mathbf{S}_j\mathbf{L}_j) = \frac{1}{2} \sum_{SL\pi} \sum_{l_i l_j} (2S + 1)(2L + 1) |\mathbf{S}_{SL\pi}(\mathbf{S}_i\mathbf{L}_i l_i - \mathbf{S}_j\mathbf{L}_j l_j)|^2$$

The **effective collision strength** $\Upsilon(\mathbf{T})$ is the Maxwellian averaged collision strength:

$$\Upsilon(\mathbf{T}) = \int_0^{\infty} \Omega_{ij}(\epsilon_j) e^{-\epsilon_j/k\mathbf{T}} d(\epsilon_j/k\mathbf{T}) \quad (27)$$

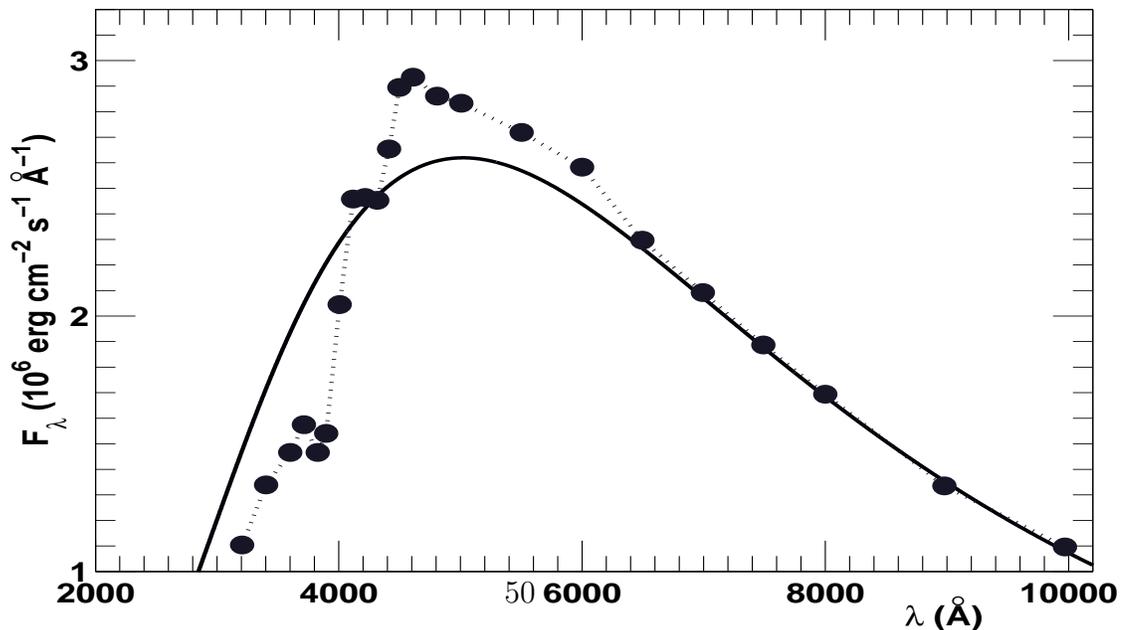
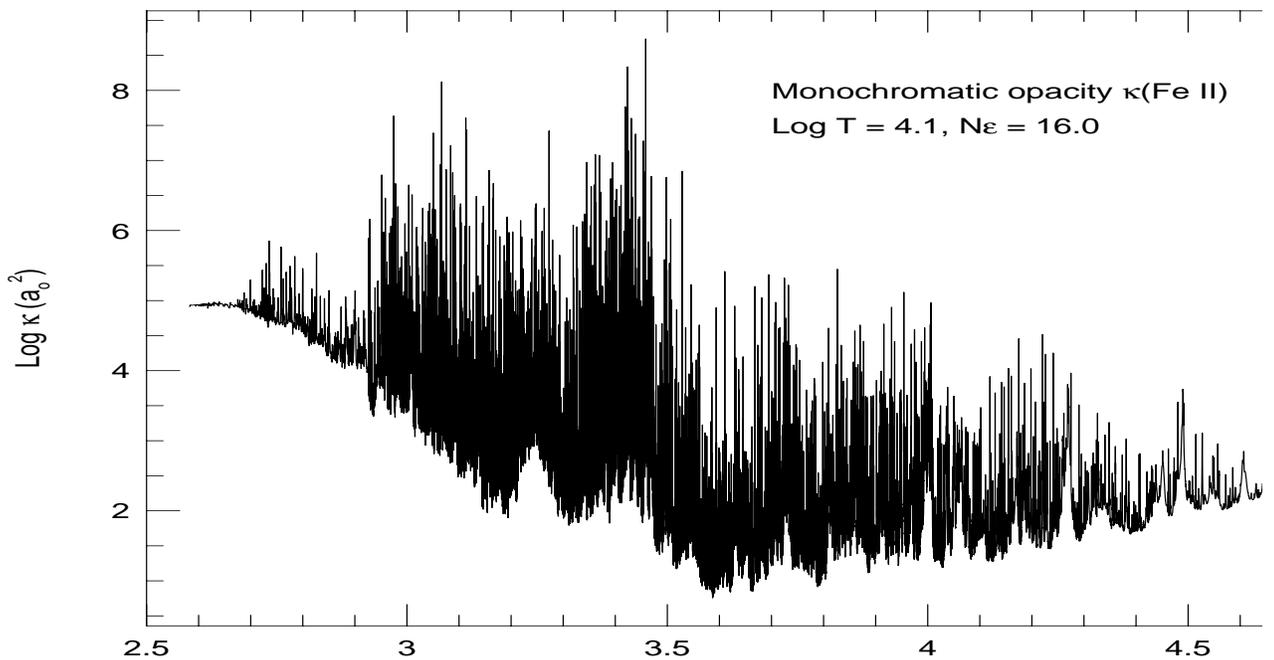
The **excitation rate coefficient** $q_{ij}(\mathbf{T})$ is

$$q_{ij}(\mathbf{T}) = \frac{8.63 \times 10^{-6}}{g_i \mathbf{T}^{1/2}} e^{-E_{ij}/k\mathbf{T}} \Upsilon(\mathbf{T}) \text{ cm}^3 \text{ s}^{-1} \quad (28)$$

$E_{ij} = E_j - E_i$ and $E_i < E_j$ in Rydbergs

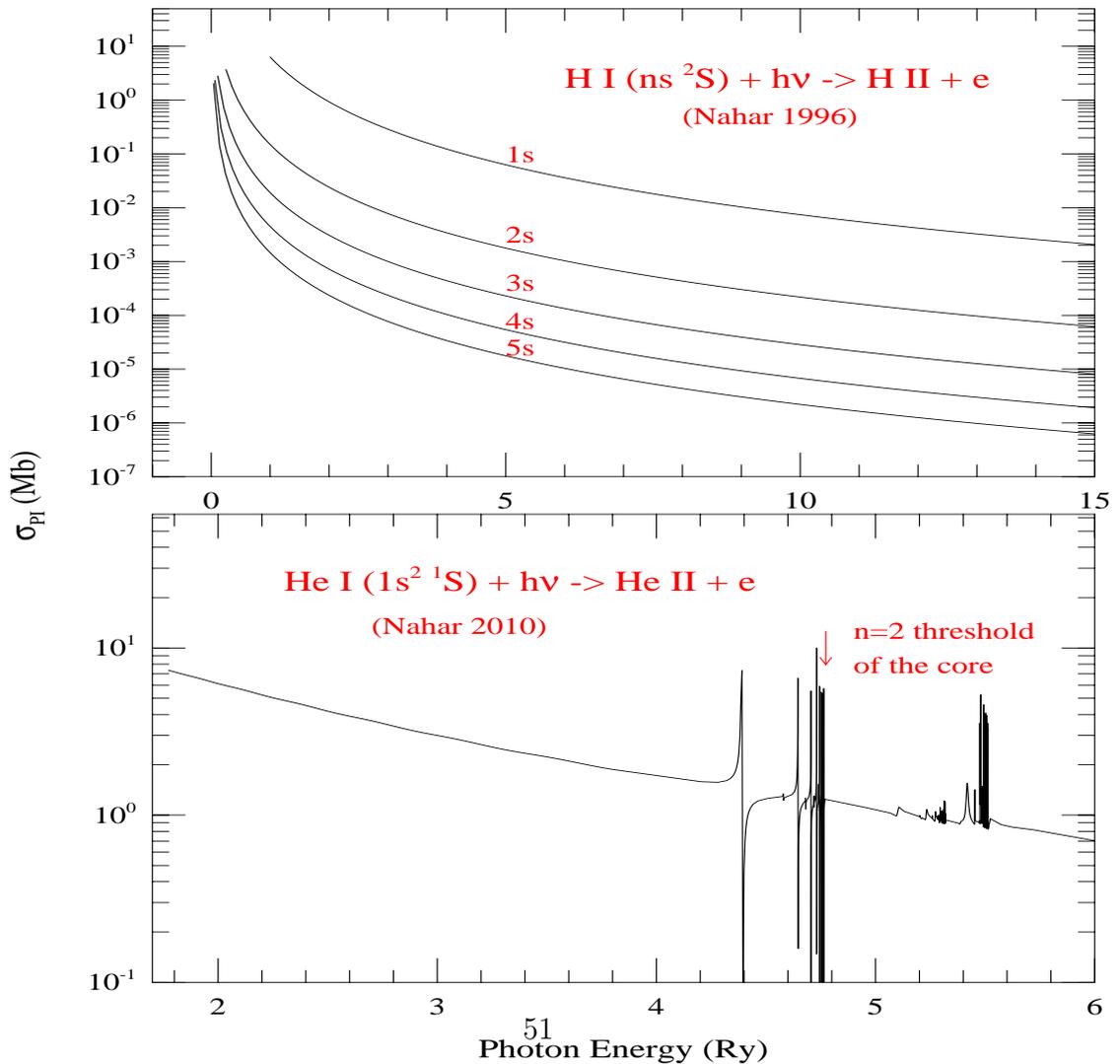
Monochromatic Opacities κ_{ν} of Fe II: (Nahar & Pradhan 1993)

TOP: Complex structure in monochromatic opacity of Fe II at the Z-bump temperature and density
Bottom: The high opacity in 2-3.5 region indicates the absorption by Fe I and Fe II in the solar blackbody radiation.



PHOTOIONIZATION

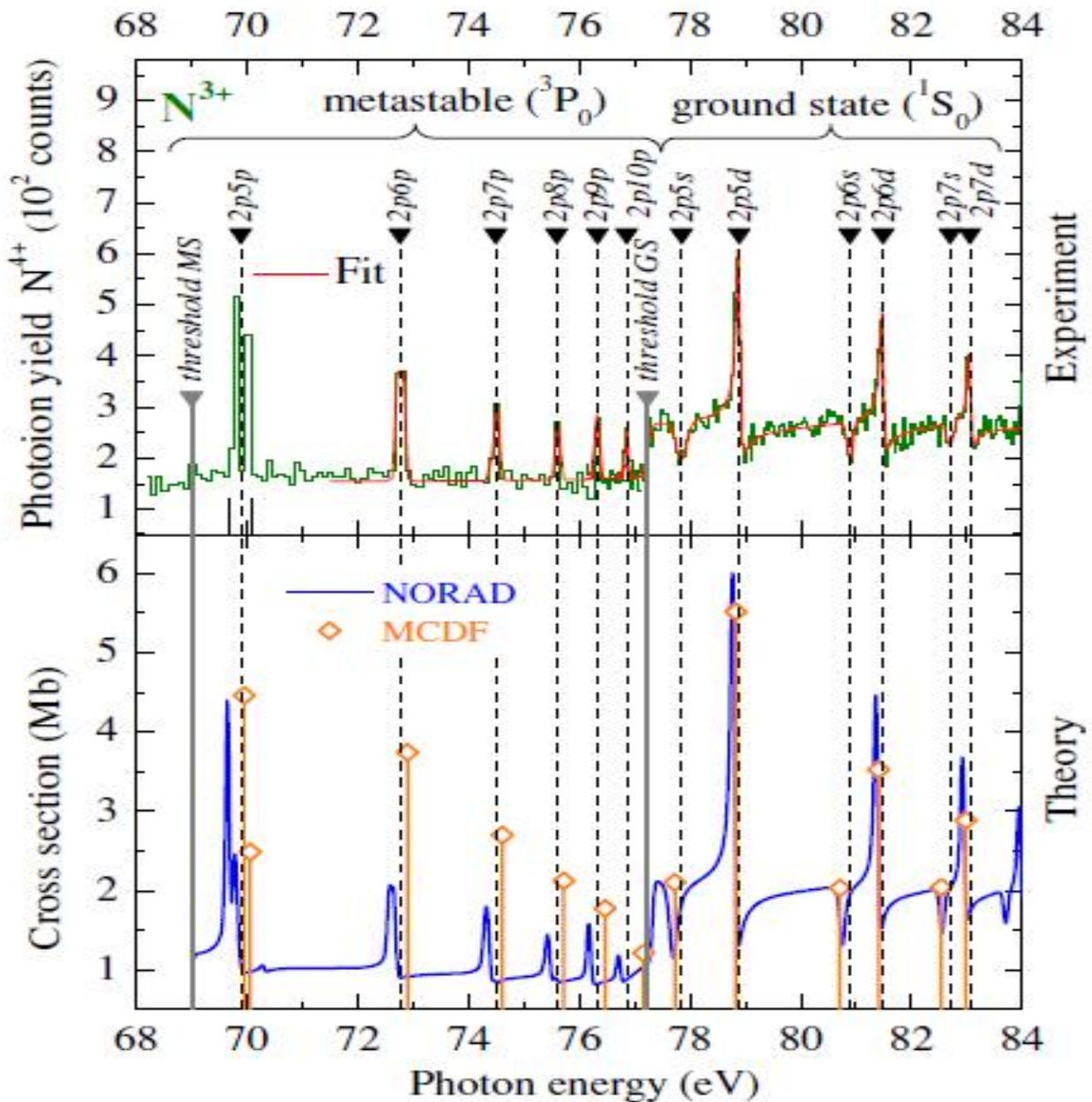
- An atom/ion with more than 1 electron will have resonances in σ_{PI}
- Earlier calculations for photoionization cross sections under the Opacity Project considered low-lying resonances
- Core excitations to higher states during photoionization are thought to be weaker and behave hydrogenic
- Fine structure introduce new features that can not be obtained in LS coupling
- New calculations under the Iron Project are showing new and dominating features not studied before - these are expected to resolve some longstanding problems



• **Physics Experiment:** Photoionization cross sections of N IV:

TOP: measured at synchrotron facility BESSY II by Simon et al (2010)

BOTTOM: Comparison with NORAD-Atomic-Data (Nahar & Praadhan 1997, blue). MCDF (Orange drop lines)

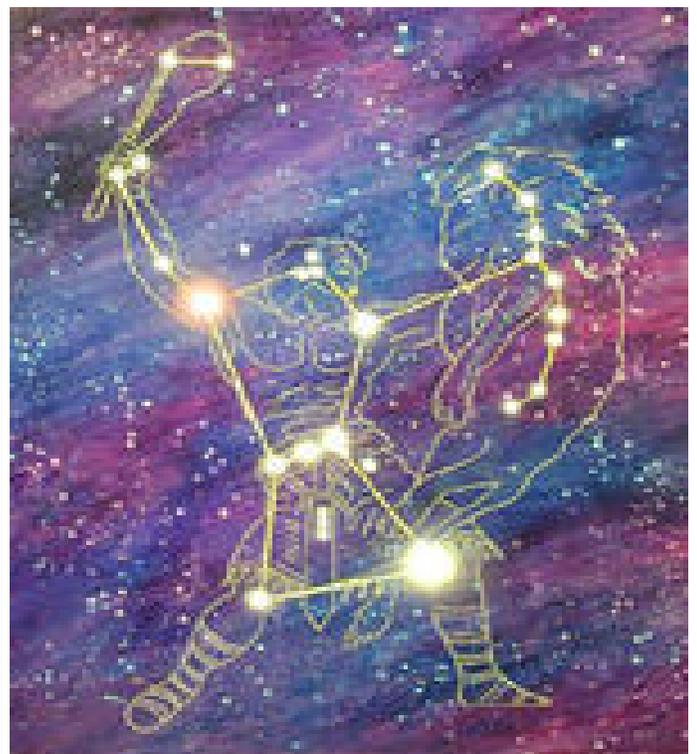
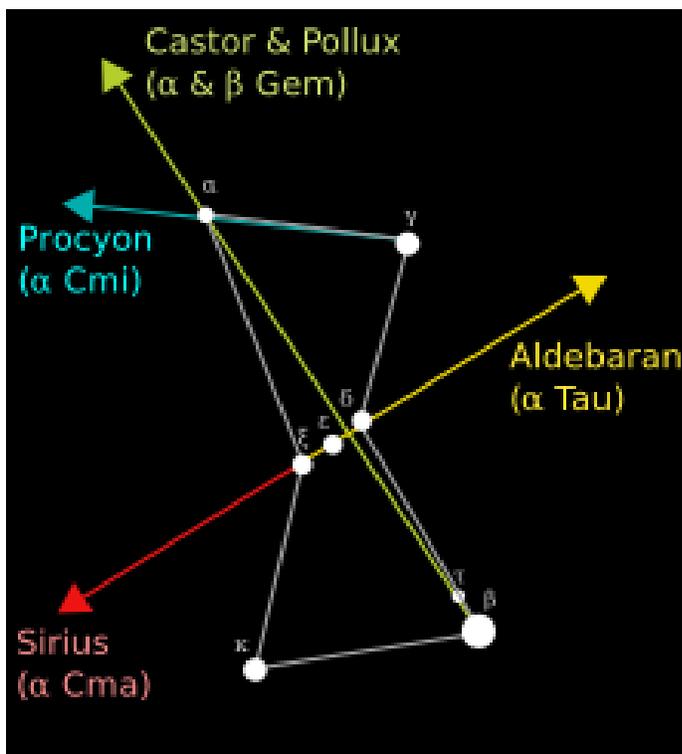


Photometric Image: Orion Constellation

Red gaseous nebula in the background - "Orion Nebula"
(reddish patch) below the belt of 3 stars (on the sword)



Below: Imaginary figure - "Orion the Hunter"



Determination of Abundance of Elements

Orion Nebula - Birthplace of Stars

[Composed by images from Spitzer & Hubble]



- Orion Nebula \sim 1500 Lyr away, closest cosmic cloud to us
 - Center bright & yellow gas - illuminated ultraviolet (UV) radiation
- Images: Spitzer - Infrared (red & orange) C rich molecules - hydrocarbons, Hubble - optical & UV (swirl green) of H, S
- Small dots - infant stars; over 1000 young stars
 - Detected lines of O II, O III

Determination of Abundance of Elements

- Longstanding problems on determination of abundances of elements - lack of accurate atomic data
- One is Oxygen abundance in Nebular plasmas Ex: Planetary Nebula (PNe) K 4-55 (below)



- Planetary nebula (PNe): last stage of a star
- Has condensed central star: very high $T \sim 100,000$ K ($\gg T \leq 40,000$ K of a typical star)
- Envelope: thin gas radiatively ejected & illuminated by radiation of the central star: red (N), blue (O)
- Common lines of low ionization states: O III and O II - low density and low T

Determination of OXYGEN Abundance:

- Collisionally Excited Lines (CEL):



- Recombination lines (REL):



- The intensity of a CEL of ion X_I due to transition between a and b (*Atomic Astrophysics and Spectroscopy*, Pradhan & Nahar, 2011)

$$I_{ba}(\mathbf{X}_i, \lambda_{ba}) = \frac{h\nu}{4\pi} n_e n_{\text{ion}} q_{ba} \quad (29)$$

q_{ba} - EIE rate coefficient in cm^3/sec . Hence population of a level b can be written as

$$N(\mathbf{b}) = n_{\text{ion}} q_{ba} = \frac{1}{n_e} \frac{4\pi}{h\nu} I_{ba} \quad (30)$$

The abundance, $n(\mathbf{X})/n(\mathbf{H})$ of the element, X , with respect to H, can be obtained from the intensity I

$$I(\mathbf{X}_i, \lambda_{ba}) = \frac{h\nu}{4\pi} A_{ba} \frac{N(\mathbf{b})}{\sum_j N_j(\mathbf{X}_i)} \frac{n(\mathbf{X}_i)}{n(\mathbf{X})} \left[\frac{n(\mathbf{X})}{n(\mathbf{H})} \right] n(\mathbf{H})$$

A_{ba} - radiative decay rate, sum over N_j - total populations of all excited levels, $N(\mathbf{b})/\sum_j N_j(\mathbf{X}_i)$ - population fraction, (obtained

from a radiative -collisional model), $\frac{n(X_i)}{n(X)}$ - ionization fraction
(come from photoionization model code)

- Similarly abundance can be obtained from the intensity of a REL from emissivity

$$\epsilon(\lambda_{pj}) = N_e N(X^+) \alpha_{\text{eff}}(\lambda_{pj}) h\nu_{pj} [\text{erg cm}^{-1}\text{s}^{-1}]$$

intensity, cascade matrix, and formula similar to that of CEL but for REL

Problem: $N_e N(\text{O III}) q_{\text{EIE}} < \alpha_{\text{R}}(\text{T}) N_e N(\text{O II})$

- **ONE SOLUTION:** Recombination rate for $(e + \text{O III}) \rightarrow \text{O II}$ - higher at low T

Close-Coupling (CC) Approximation & the R-matrix method

- CC approximation:
- Ion treated as a (N+1) electron system:
 - a target or the ion core of N electrons and
 - an additional interacting (N+1)th electron:
- Total wavefunction expansion is:

$$\Psi_{\mathbf{E}}(\mathbf{e} + \mathbf{ion}) = A \sum_i^N \chi_i(\mathbf{ion})\theta_i + \sum_j c_j \Phi_j(\mathbf{e} + \mathbf{ion})$$

$\chi_i \rightarrow$ target ion or core wavefunction

$\theta_i \rightarrow$ interacting electron wavefunction (continuum or bound)

$\Phi_j \rightarrow$ correlation functions of (e+ion)

- The complex resonant structures in the atomic processes are included through channel couplings.
- Substitution of $\Psi_{\mathbf{E}}(\mathbf{e} + \mathbf{ion})$ in $\mathbf{H}\Psi_{\mathbf{E}} = \mathbf{E}\Psi_{\mathbf{E}}$ results in a set of coupled equations
- Coupled equations are solved by R-matrix method

Close-Coupling Approximation

The wavefunction is an expansion:

$$\Psi_{\mathbf{E}}(\mathbf{e} + \text{ion}) = \mathbf{A} \sum_{\mathbf{i}}^{\mathbf{N}} \chi_{\mathbf{i}}(\text{ion})\theta_{\mathbf{i}} + \sum_{\mathbf{j}} \mathbf{c}_{\mathbf{j}}\Phi_{\mathbf{j}}(\mathbf{e} + \text{ion})$$

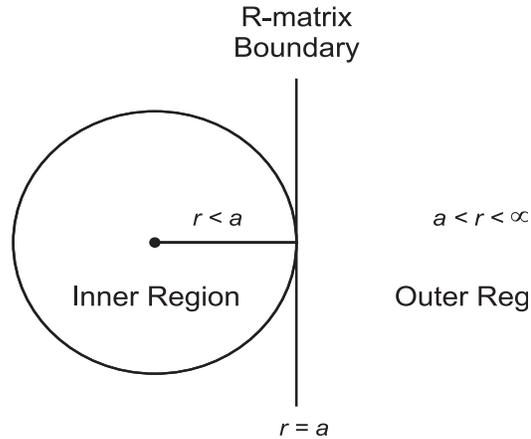
- $\chi_{\mathbf{i}} \rightarrow$ core ion wavefunction, $\Phi_{\mathbf{j}} \rightarrow$ correlation functions
 $\theta_{\mathbf{i}} \rightarrow$ interacting electron wavefunction (continuum or bound)
 • Resonant Structures - manifested through channel couplings

Table (Example of $\Psi_{\mathbf{E}}$): Levels and energies (E_t) of the core O III in wave function expansion of O II; SS - calculated energies and NIST - observed energies

Level	J_t	$E_t(\text{Ry})$ NIST	$E_t(\text{Ry})$ SS
1	0	0.0	0.
2	1	0.0010334	0.0011497
3	2	0.0027958	0.003384
4	2	0.18472	0.21215
5	2	0.39352	0.38420
6	2	0.54972	0.46200
7	3	1.0938	1.12584
8	2	1.0940	1.12576
9	1	1.0941	1.12573
10	2	1.2975	1.32510
11	1	1.2975	1.32500
12	0	1.2976	1.32495
13	2	1.7045	1.83934
14	1	1.7960	1.89708
15	1	1.9178	2.02463
16	0	2.4354	2.33186
17	1	2.4365	2.33300
18	2	2.4388	2.33186
19	1	2.4885	2.40908

R-matrix Method

Substitution of $\Psi_E(\text{e} + \text{ion})$ in $\mathbf{H}\Psi_E = \mathbf{E}\Psi_E \rightarrow$ a set of coupled equations - solved by the R-matrix method



- Divide the space in two regions, the inner and the outer regions, of a sphere of radius r_a with the ion at the center. r_a is large enough for to include the electron-electron interaction potential. Wavefunction at $r > r_a$ is Coulombic due to perturbation
- In the inner region, the radial part $F_i(r)/r$ of the outer electron wave function (θ) is expanded in terms of a basis set, called the R-matrix basis,

$$\mathbf{F}_i = \sum \mathbf{a}_k \mathbf{u}_k$$

$$\left[\frac{d^2}{dr^2} - \frac{l(l+1)}{r^2} + \mathbf{V}(r) + \epsilon_{lk} \right] \mathbf{u}_{lk} + \sum_n \lambda_{nlk} \mathbf{P}_{nl}(r) = \mathbf{0}. \quad (31)$$

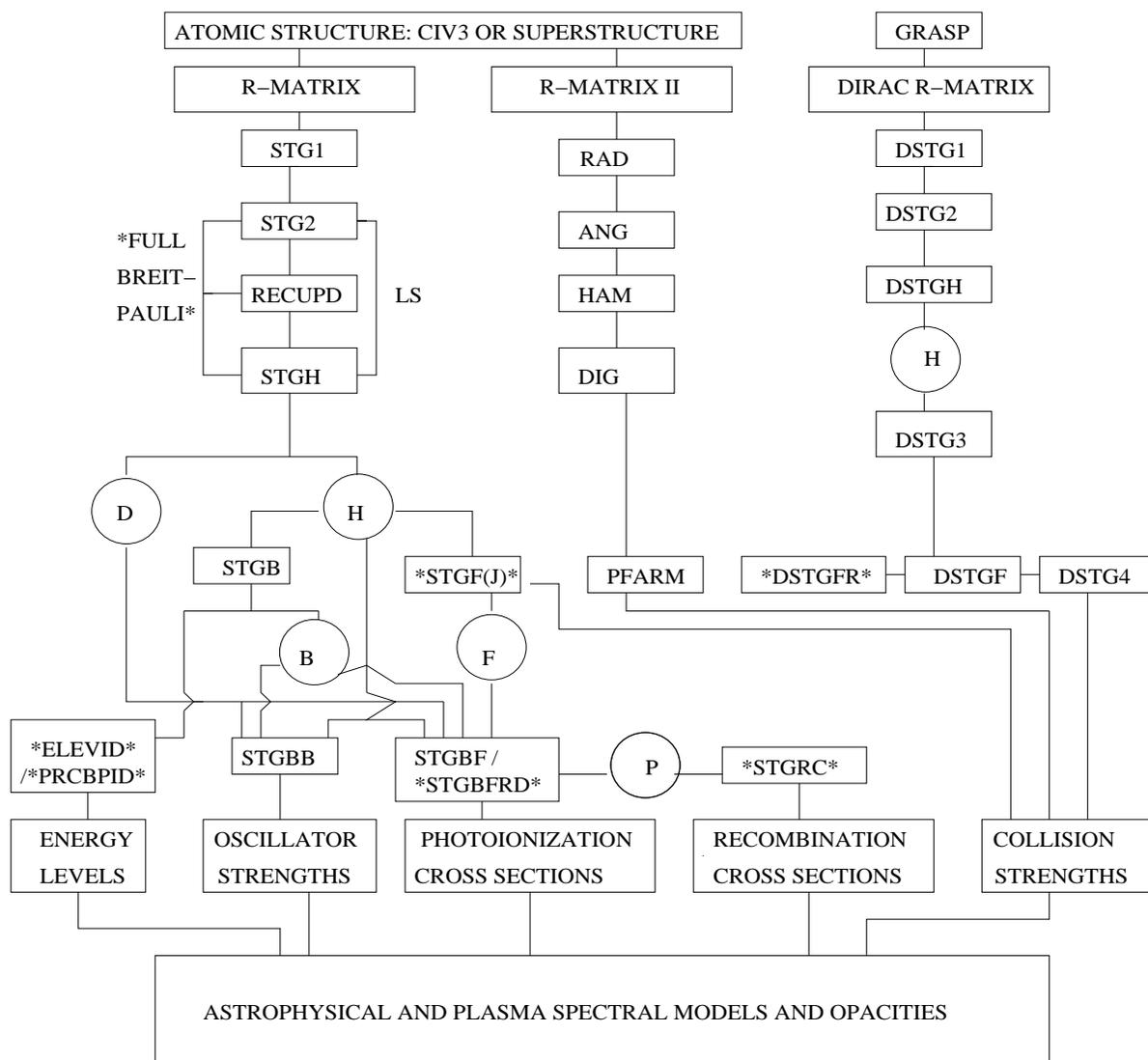
and is made continuous at r_a with the Coulomb functions outside r_a

R-Matrix Codes For Large-Scale Atomic Calculations at the Ohio Supercomputer Center

VARIOUS COMPUTATIONAL STAGES

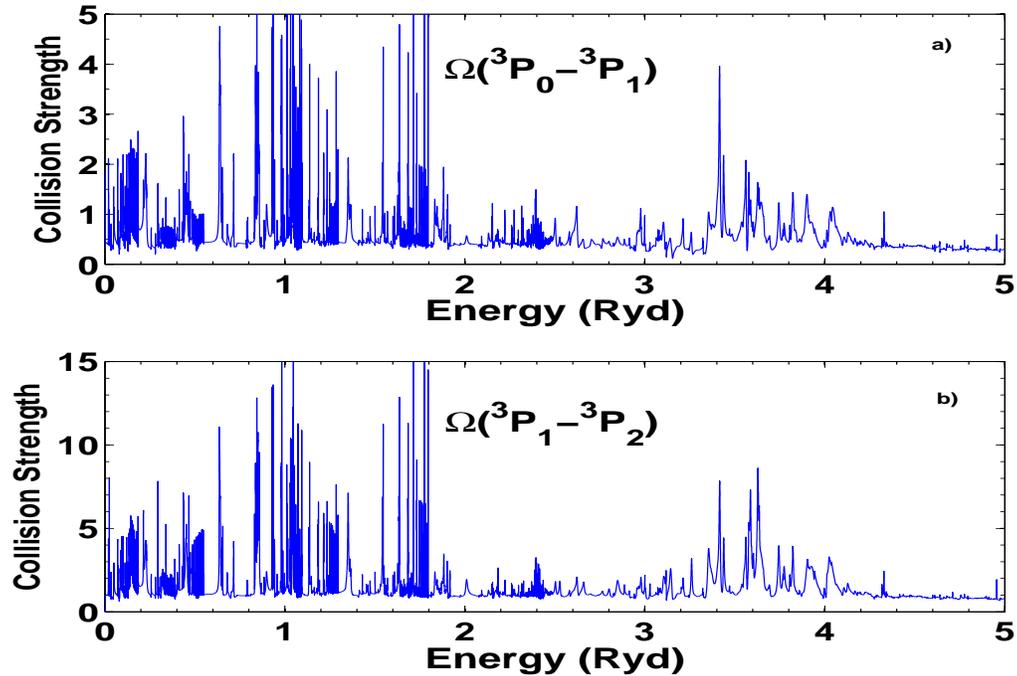
- R-matrix calculations can proceed in 3 branches - 1) LS coupling & relativistic Breit-Pauli, 2) LS coupling R-matrix II for Large configuration interaction, 3) DARC for Full Dirac relativistic
- Results - 1) Energy Levels, 2) Oscillator Strengths, 3) Photoionization Cross sections, 4) Recombination Rate Coefficients, 5) Collision Strengths; - Astrophysical Models

THE R-MATRIX CODES AT OSU

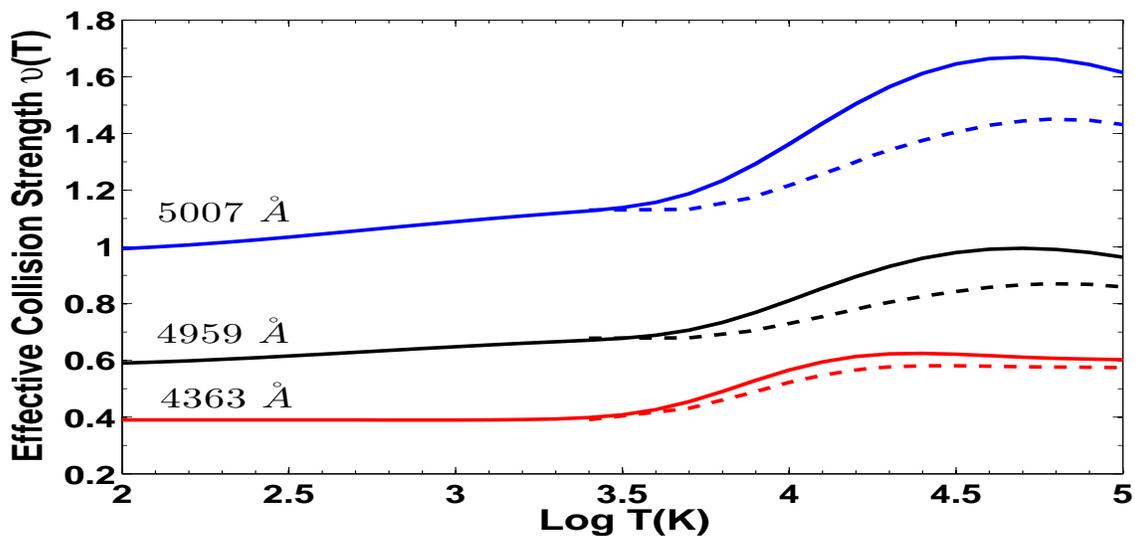


ELECTRON IMPACT EXCITATIONS (EIE)

- Fig Collision strengths Ω (EIE) of O III: $2p^2(^3P_0 - ^3P_1)$ ($88\mu\text{m}$) and $2p^2(^3P_1 - ^3P_2)$ ($52\mu\text{m}$) (IR) (Palay et al 2012)
- The near threshold resonances are seen for the first time

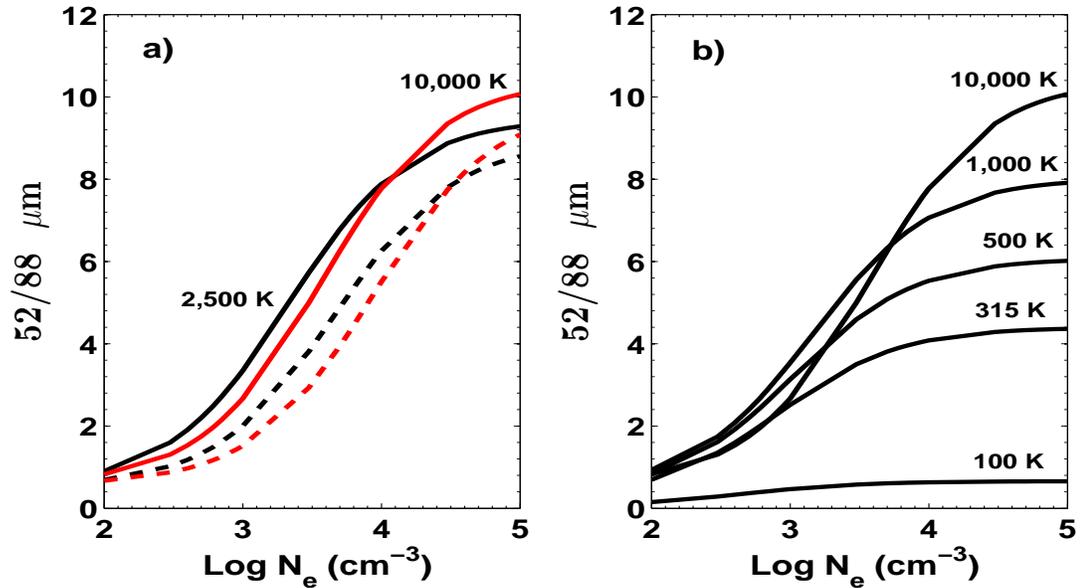


- Fig Effective $\Omega(T)$ of 3 optical lines; Solid: BPRM (present), Dashed: R-matrix(LS) (Aggarwal & Keenan)
- Differences to affect T and density diagnostics

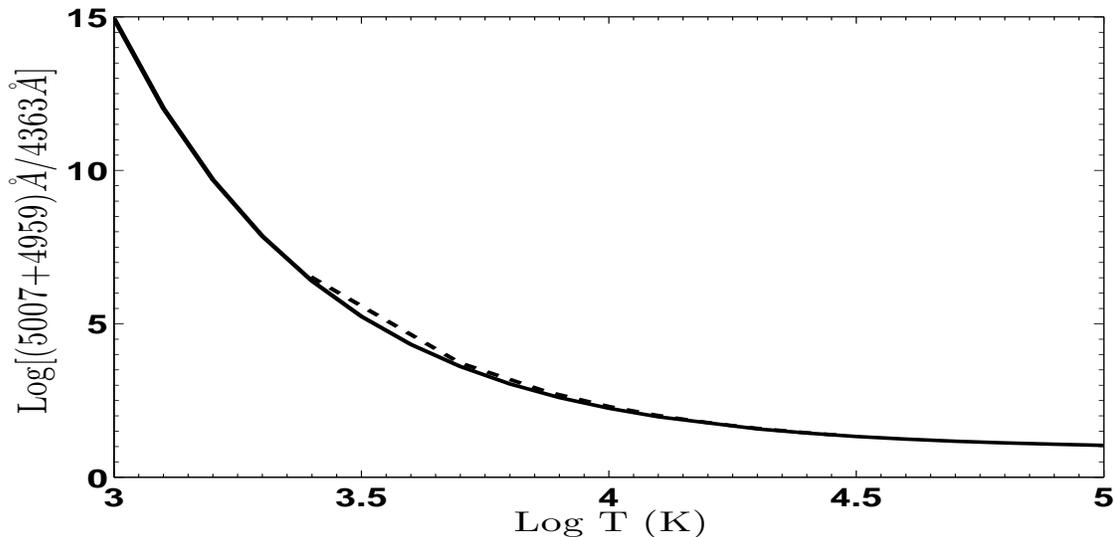


LINE RATIO: DENSITY & T DIAGNOSTICS

- Intensity ratio of two observed lines can be compared to the calculated curves for density (ρ) & T diagnostics
- **Fig** Significant FS effect on ρ diagnostics, 100 - 10,000 K



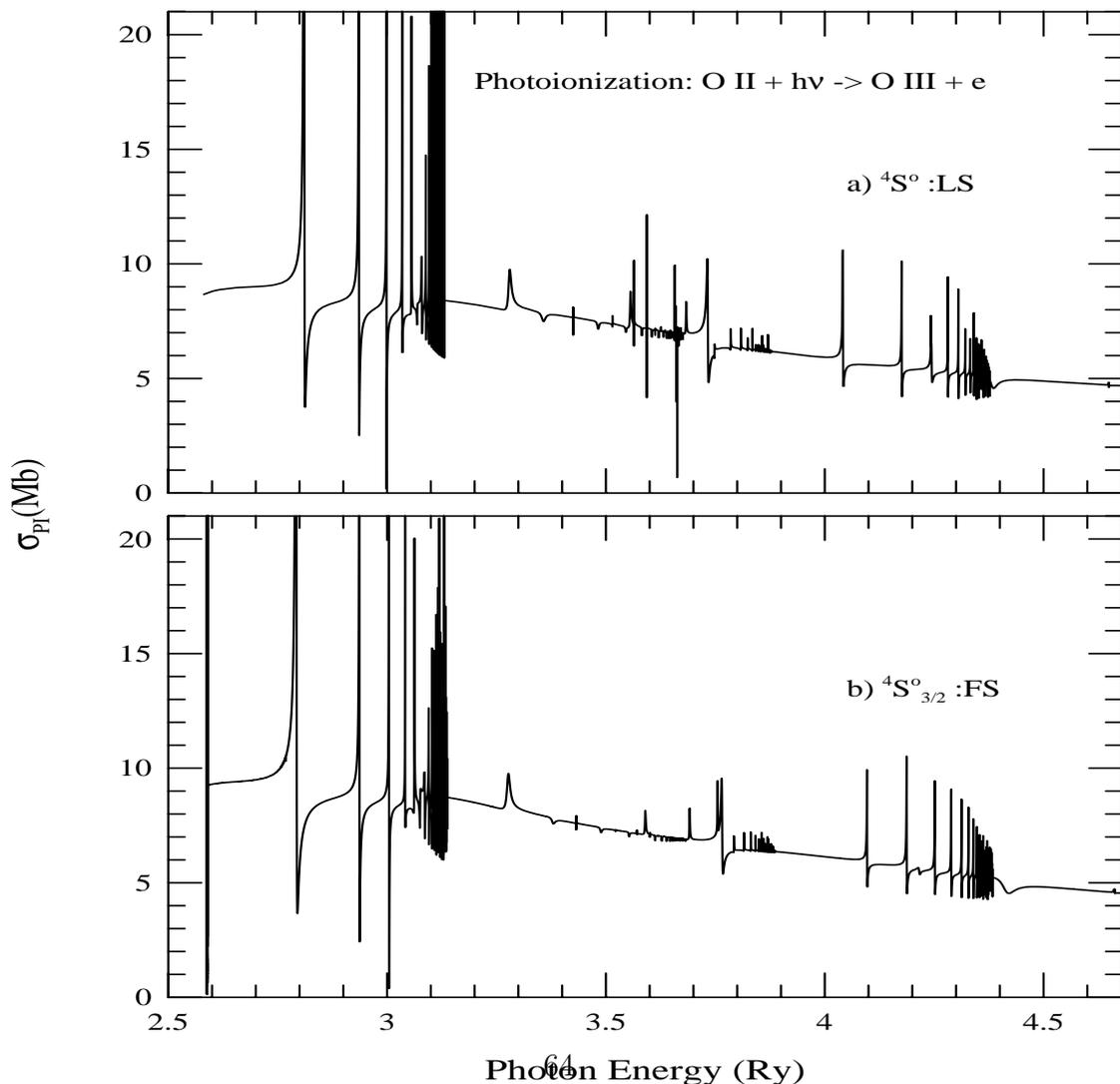
- Blended line ratio with T at $N_e = 10^3 \text{ cm}^{-3}$. The diagnostic indicates considerable rise in low T region



Fine Structure Effects on Low Z ion: O II

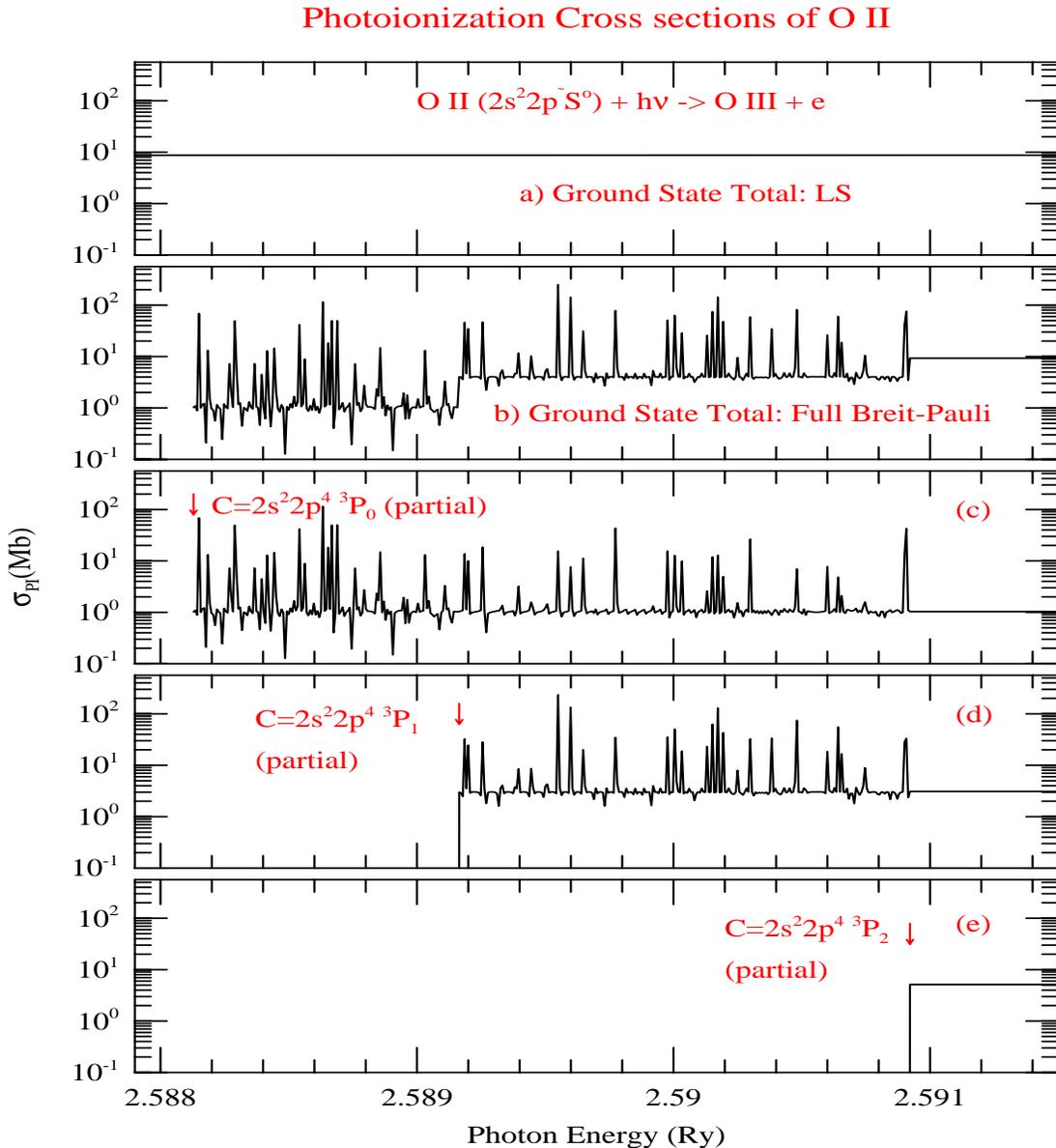
(Nahar et al. 2010)

- Previous study of σ_{PI} for O II in LS coupling R-matrix (Nahar 2004) showed good agreement with experiment (ALS: Covington et al. 2001)
- However, problem remained with O II abundance at low T astrophysical plasmas
- σ_{PI} in fine structure (FS) is very similar to that in LS coupling except the resonant structure at the threshold
- The new low energy features in $\sigma_{PI}(\text{FS})$ should narrow down the discrepancy



GROUND STATE PHOTOIONIZATION OF O II:

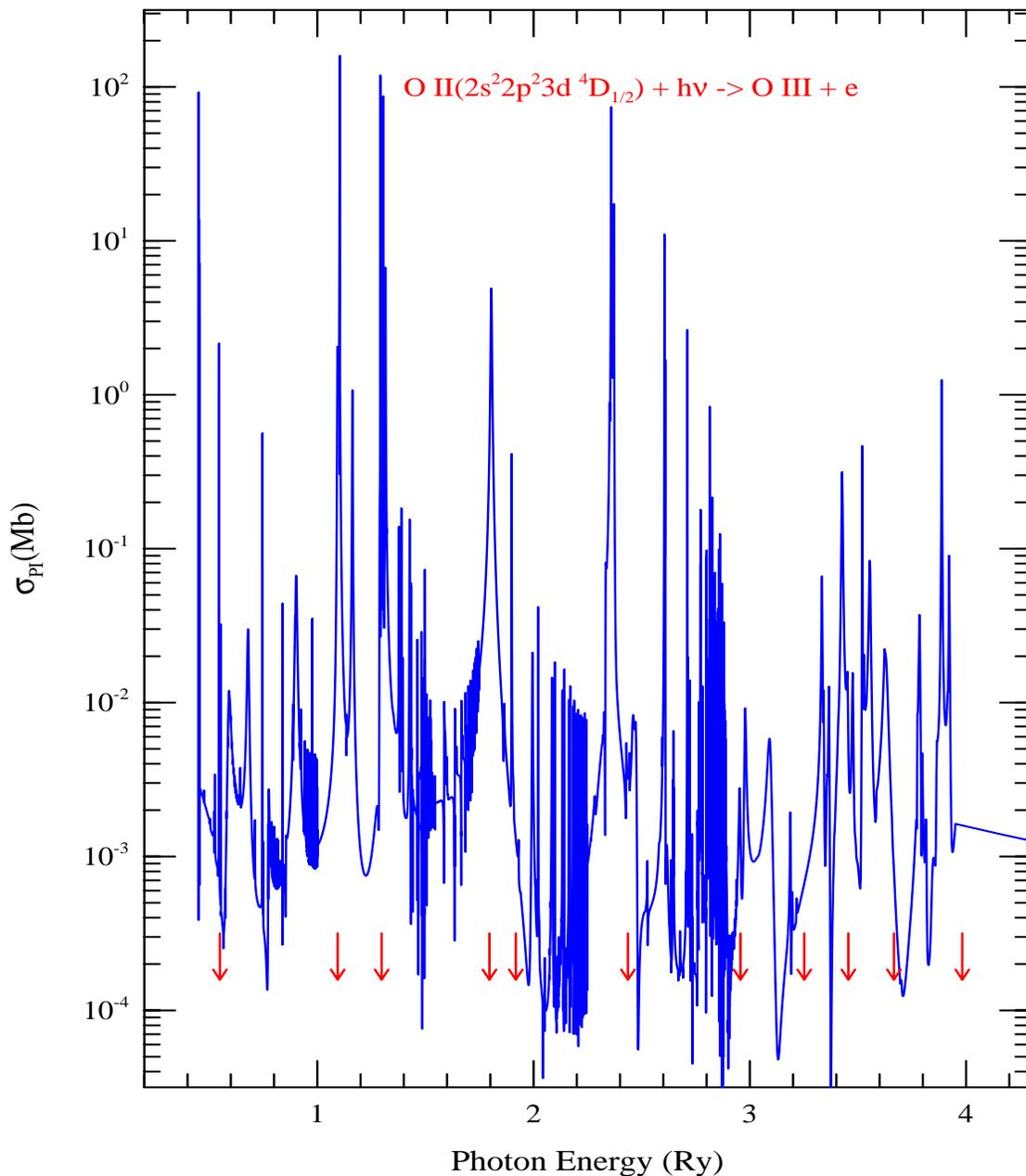
- Photoionization of Ground state $4S^o$ of O II in small energy region near threshold
- Panel a) (Top) Photoionization in LS coupling - no structure in the region (Nahar 2004)
- Panel b) Total σ_{PI} , of the ground level $4S^o_{3/2}$: i) filled with resonant structures, ii) background shifts (Nahar et al. 2010)
- Panels c,d,e) Partial photoionization into fine structure components $2p^2\ ^3P_{0,1,2}$ of O III, arrows \rightarrow ionization thresholds.



PHOTOIONIZATION OF O II: PEC RESONANCES

- PEC (Photo- Excitation-of-Core) resonances appear photoionization of single electron valence excited levels
- Figure shows that PEC resonances are strong and enhance the background cross sections by orders of magnitude
- These PEC resonances will affect photoionization and recombination rates of plasmas

Photoionization of O II: PEC Resonances

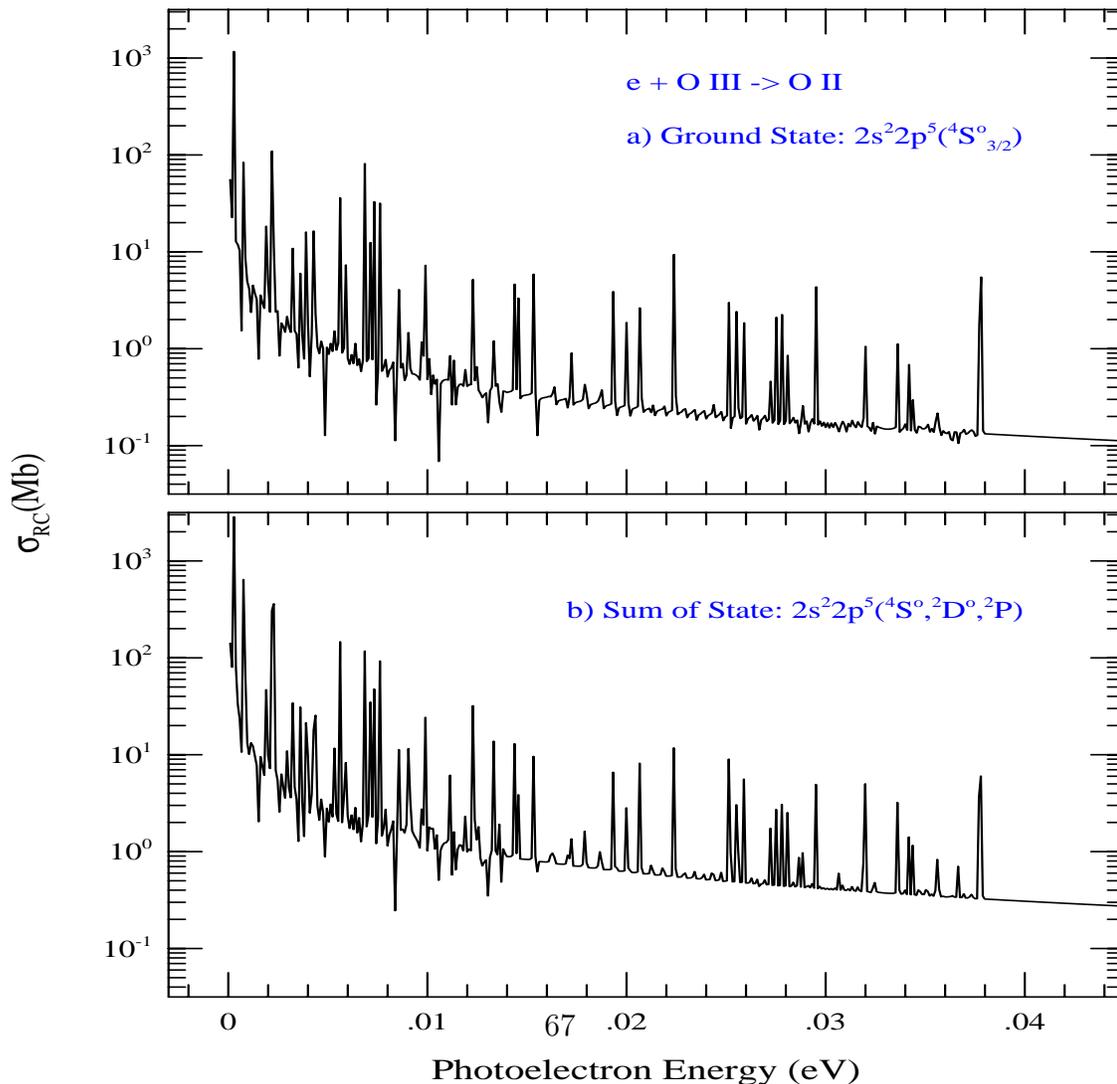


New RECOMBINATION CROSS SECTION AT LOW T: O II

(Nahar et al. 2010)

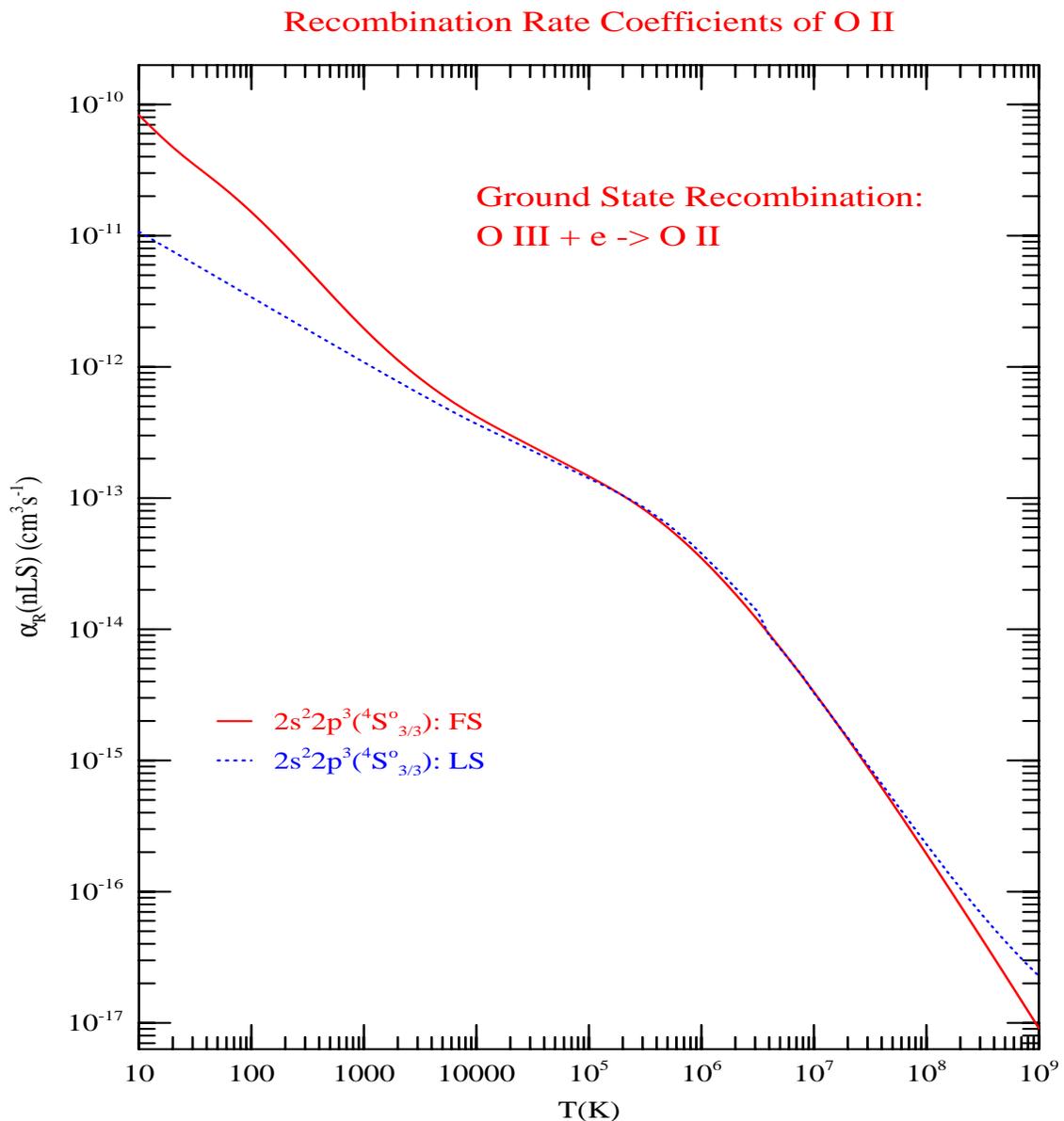
- Recombination cross section σ_{RC} show many resonances at very low energy
- Recombination rate from these will increase the recombination rate
- Higher recombination rate will decrease the oxygen abundance and hence will resolve or narrow down the existing gap in nebular problem

Recombination Cross Sections of O II



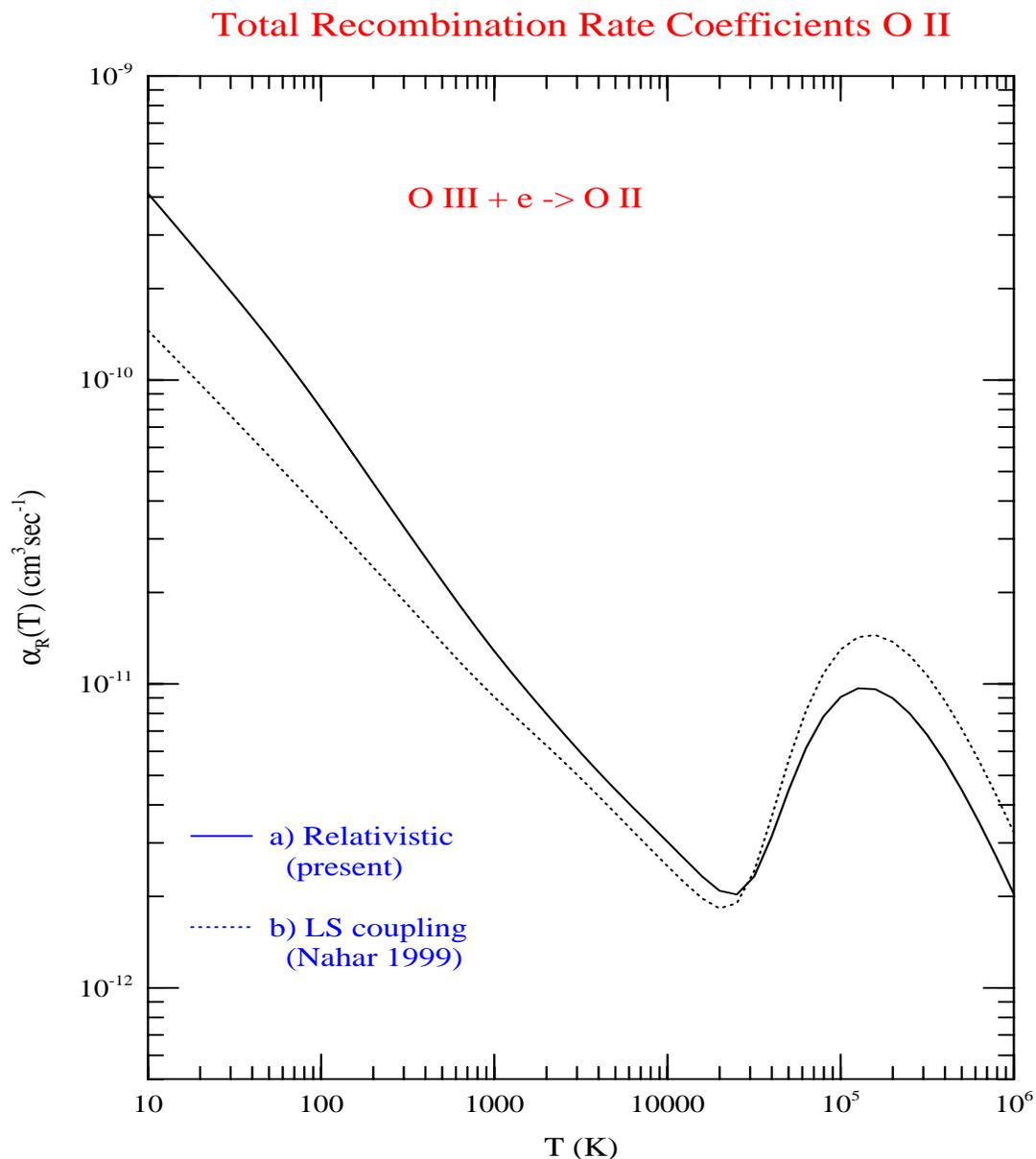
O II: Ground State RECOMBINATION (RR+DR):

- Fig: Recombination rate coefficients (RRC) $\alpha_{RC}(T)$ for the ground state
- Fine structure effects have increased RRC considerably at low temperature
- This indicates lower abundance of oxygen

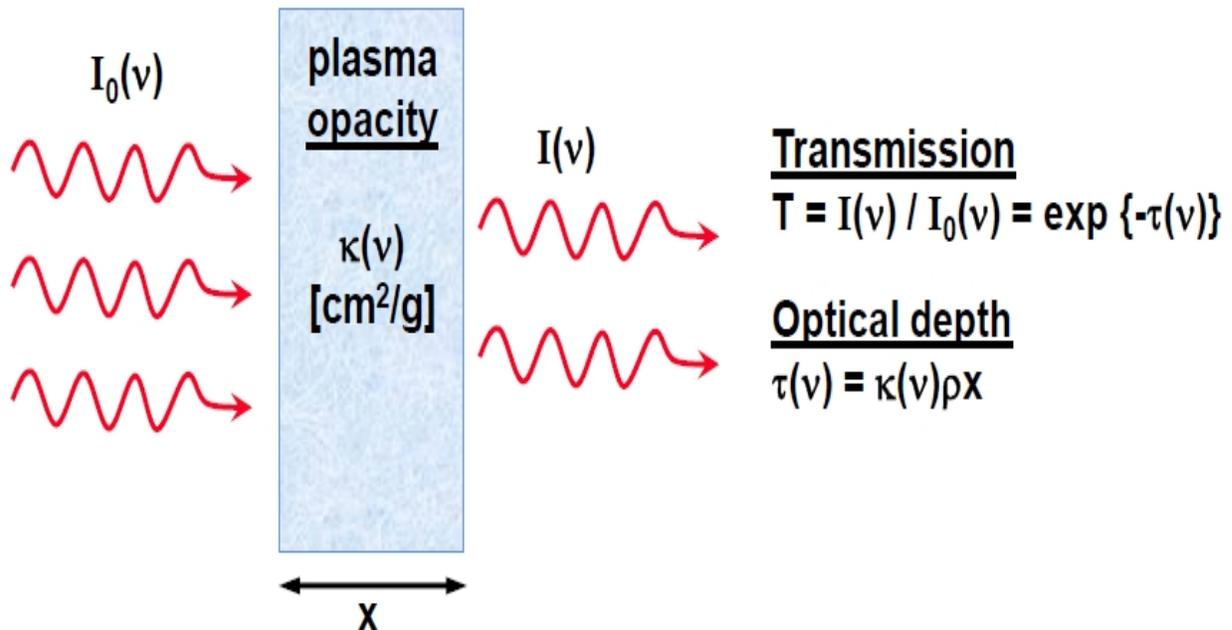


O II: TOTAL RECOMBINATION (RR+DR):

- Fig: Total recombination rate coefficients (RRC) $\alpha_{RC}(T)$
- Fine structure effects have increased RRC considerably, by 2.8 times around 10 K and 2.2 at 100 K,
- The high temperature DR peak has been reduced some. The reason could be some radiation damping effect included in the present work. The resolution of the narrow fine structure resonances could also affect the decrease.



PLASMA OPACITY



Stellar structure depends on opacities that have never been measured

- Opacity is a fundamental quantity for plasmas - it determines radiation transport in plasmas
- Opacity is the resultant effect of repeated absorption and emission of the propagating radiation by the constituent plasma elements.
 - Microscopically opacity depends on two radiative processes: i) photo-excitation (bound-bound transition) & ii) photoionization (bound-free transition)
- The total $\kappa(\nu)$ is obtained from summed contributions of all possible transitions from all ionization stages of all elements in the source.
- Calculation of accurate parameters for such a large number of transitions has been the main problem for obtaining accurate opacities.

The OPACITY Project & The IRON Project

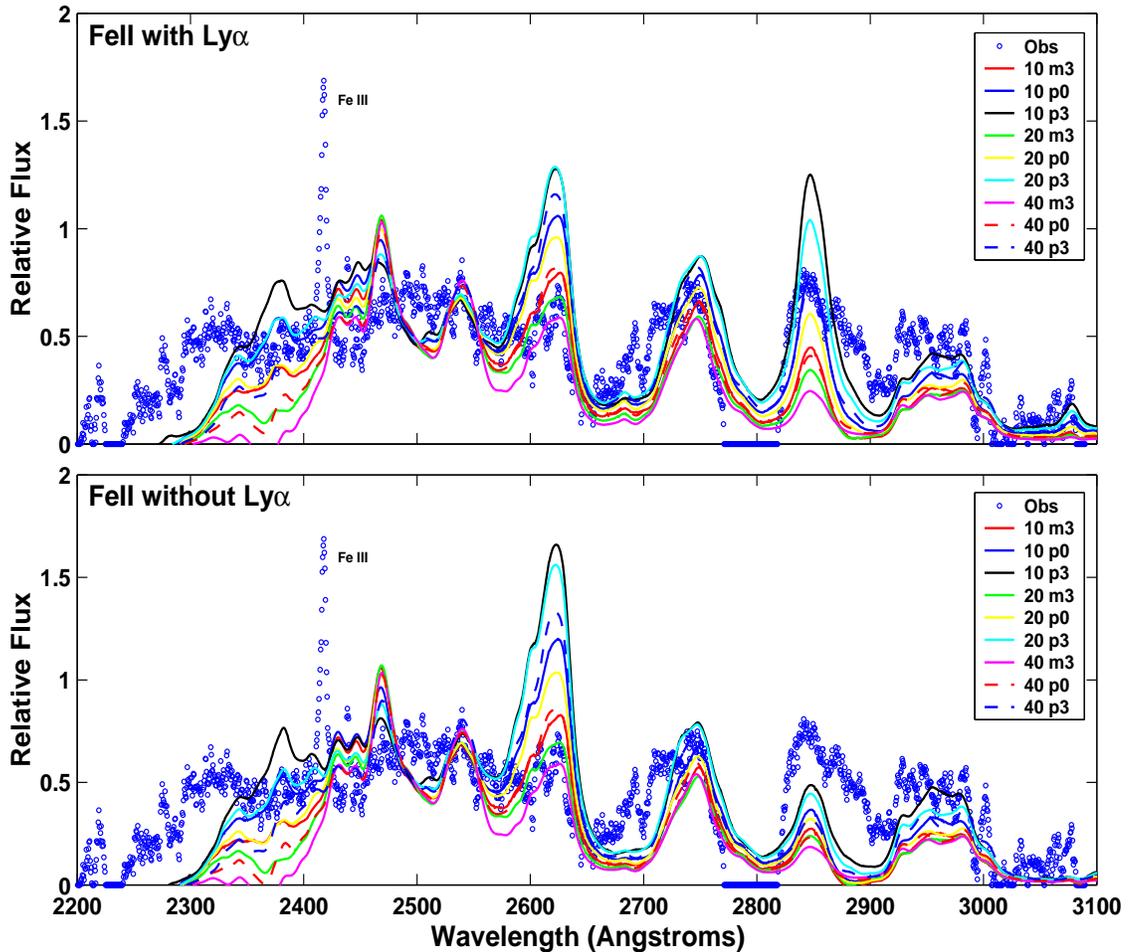
AIM: Accurate Study of Atoms & Ions, Applications to Astronomy

- International Collaborations: France, Germany, U.K., U.S., Venezuela, Canada, Belgium
- Earlier opacities were incorrect by factors of 2 to 5 → inaccurate stellar models → initiation of the Opacity Project in 1981
- **THE OPACITY PROJECT - OP (1981 - 2006):**
 - Studied radiative atomic processes for (E, f, σ_{PI})
 - Elements: H to Fe
 - Calculated opacities of astrophysical plasmas
- **THE IRON PROJECT - IP (1993 -):** collisional & radiative processes of Fe & Fe peak elements
- **RMAX:** Under IP, study X-ray atomic astrophysics
- Atomic & Opacity Databases (from OP & IP)
 - TOPbase (OP) at CDS:
<http://vizier.u-strasbg.fr/topbase/topbase.html>
 - TIPbase (IP) at CDS:
<http://cdsweb.u-strasbg.fr/tipbase/home.html>
 - OPserver for opacities at the OSC: <http://opacities.osc.edu/>
 - Latest radiative data at NORAD-Atomic-Data at OSU:
http://www.astronomy.ohio-state.edu/~nahar/nahar_radiativeatomicdata/index.html

ACHIEVEMENTS OF THE OP & THE IP

- The first detailed study of the atomic processes for most of the atoms and ions
- Found new features in photoionization cross sections
- OP opacities agreed with those computed under the OPAL, and both solved the outstanding problem on pulsations of cepheids
- Results from the OP and the IP continue to solve many outstanding problems, e.g., spectral analysis of blackhole environment, abundances of elements, opacities in astrophysical plasmas, dark matter
- **HOWEVER**, these are not complete and sufficiently accurate enough to solve all astrophysical problems

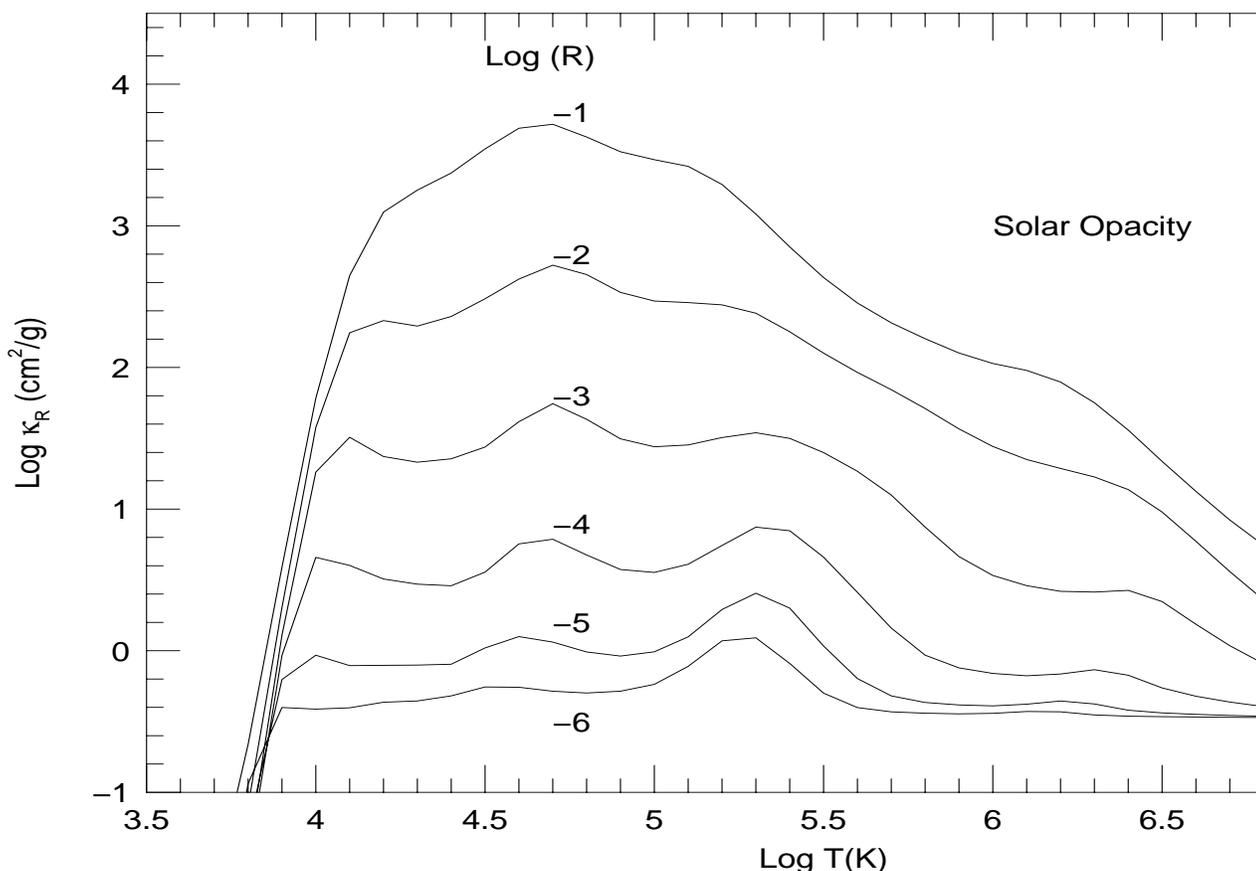
Observation & Modeling: Emission spectra of Fe I - III in active galaxy 1 Zwicky 1 (Sigut, Pradhan, Nahar 2004)



- Blue - Observation; Curves - Various Models with 1000 energy levels, millions of transitions
- With (top) and without (bottom) Lyman-alpha fluorescent excitation of Fe II by recombining H-atoms. The models reproduce many of the observed features, but discrepancies indicate need for more accurate calculations.

SOLAR OPACITY

(Seaton et al 1993)

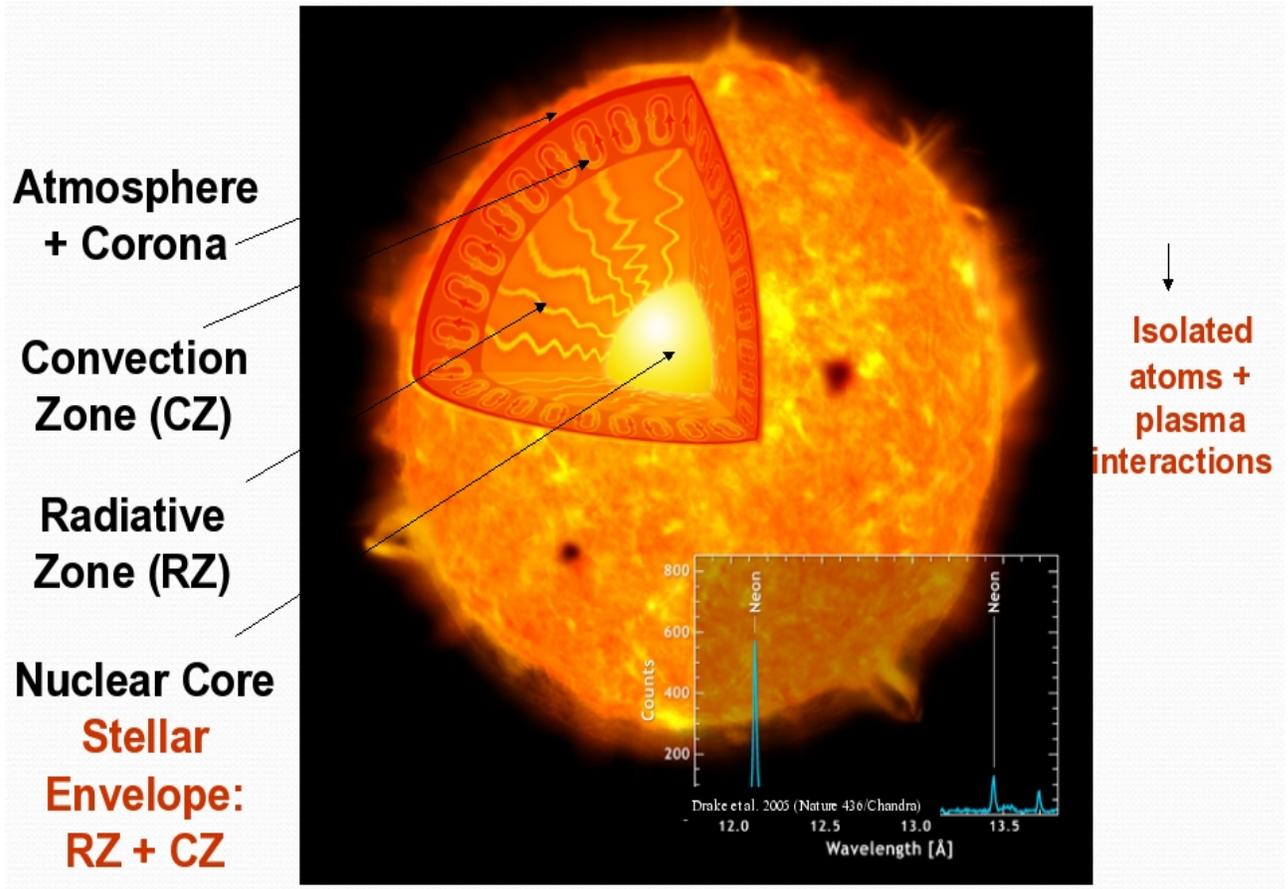


- The Rosseland mean opacity κ_R in several temperature-density regimes throughout the solar interior characterized by the parameter

$$\mathbf{R} = \rho(\text{g/cc})/\mathbf{T}_6^3, \quad \mathbf{T}_6 = \mathbf{T} * 10^{-6}$$

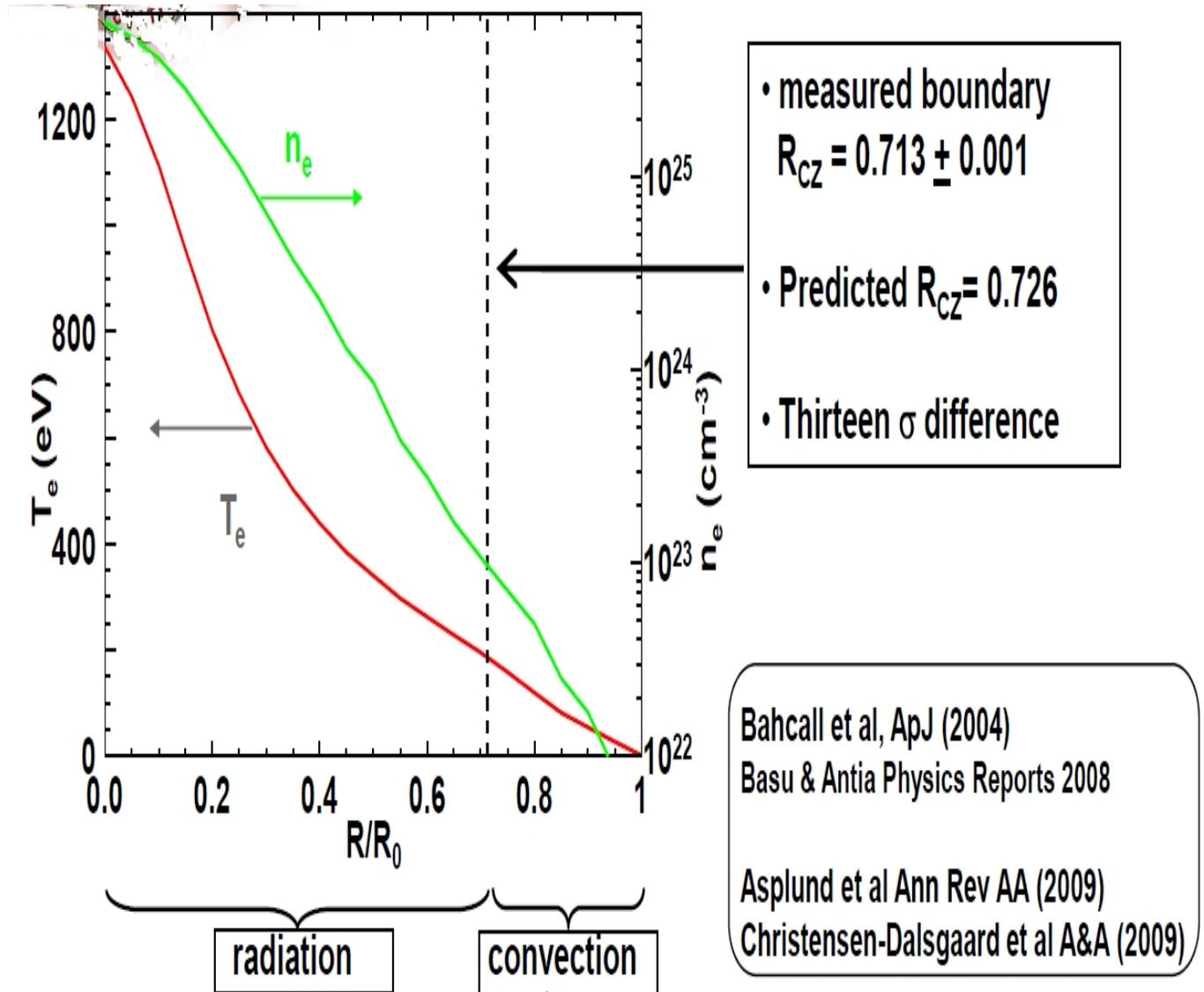
- For the sun, $\mathbf{-6} \leq \mathbf{R} \leq \mathbf{-1}$
- The bumps and kinks in the curves represent higher opacities due to excitation/ionization of different atomic species at those temperatures (H-, He-, Z-, and inner-shell bumps)

SOLAR OPACITY & ABUNDANCES

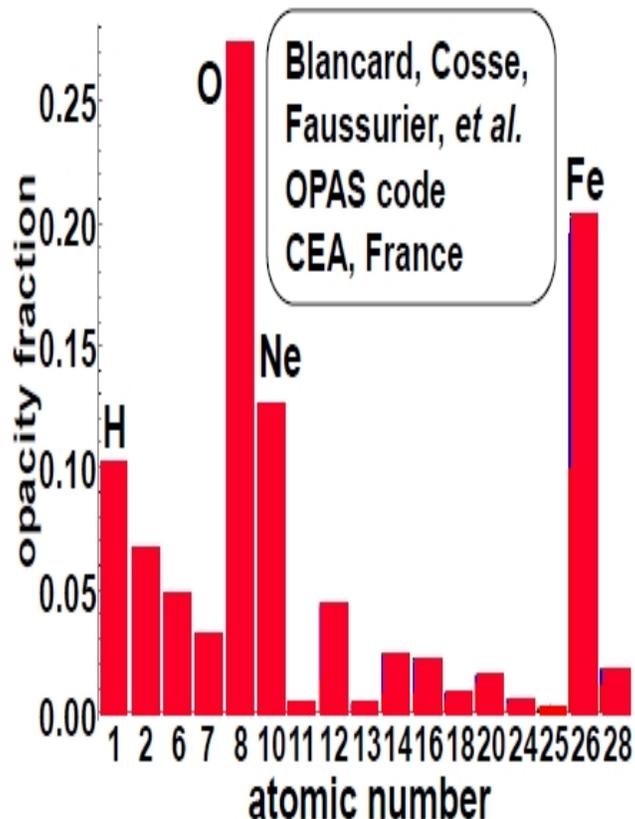
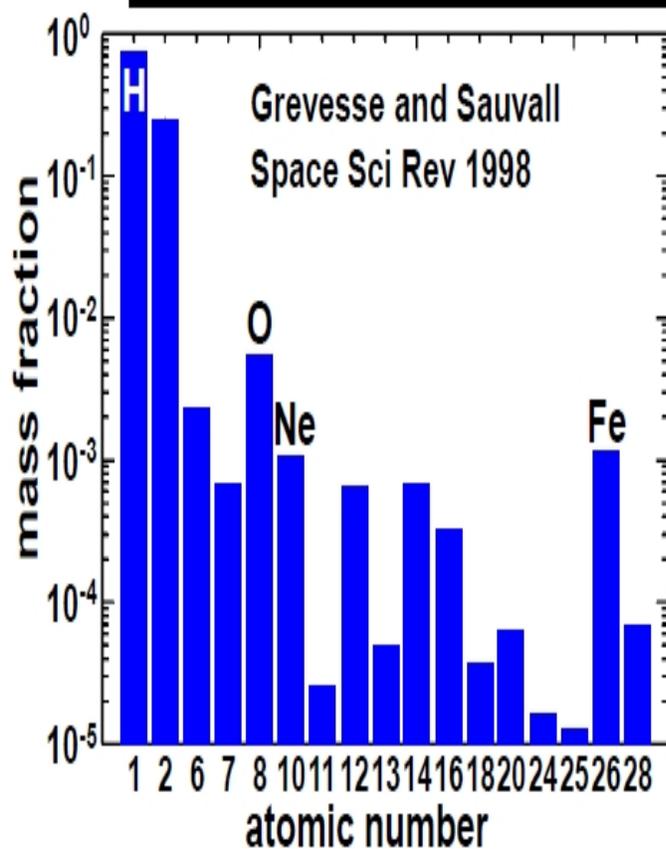


- At the convection zone boundary, R_{CZ} , the temperature $T_e \sim 193$ eV (2 MK), density $n_e \sim 10^{23}/cm^3$ (HED - High Energy Density)
- HED condition \rightarrow important elements: O, Ne, especially Fe (Fe XVII-XIX)
- Recent determination from measurements and 3D hydro NLTE models: 30-40% lower abundances of C, N, O, Ne, Ar than the standard abundances
- Optical depth in the sun depends on the interior opacity (κ) through elemental abundances
- The opacity can determine R_{CZ}

DISCREPANCY IN SOLAR RADIATIVE AND CONVECTION ZONES BOUNDARY (R_{CZ})



- The measured boundary, from helioseismology, of R_{CZ} is 0.713
- The calculated R_{CZ} is 0.726 - large



- oxygen, neon, and iron are most important at the CZ base
- The importance of any single element is diluted by the mixture

Example:

Changing Fe opacity by 1.5x causes ~11% change in total mean opacity

Determination of OPACITY: RADIATIVE ATOMIC PROCESSES

1. Photoexcitation - Photon absorption for a bound-bound transition



- Oscillator Strength (f_{ij})

Monochromatic opacity (κ_ν) depends on f_{ij}

$$\kappa_\nu(\mathbf{i} \rightarrow \mathbf{j}) = \frac{\pi e^2}{mc} N_i f_{ij} \phi_\nu \quad (32)$$

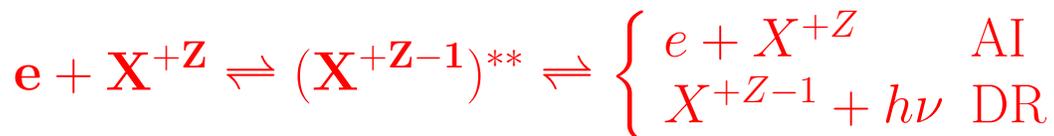
N_i = ion density in state \mathbf{i} , ϕ_ν = profile factor

- κ includes $\sim 100\text{M}$ transitions of mid-Z elements

2. Photoionization - Photon absorption for a bound-free transition: Direct -



3. Autoionization (AI) in photoionization process :



Doubly excited "autoionizing state" \rightarrow resonance

- Photoionization Cross Sections (σ_{PI})

κ_ν depends on σ_{PI}

$$\kappa_\nu = N_i \sigma_{PI}(\nu) \quad (33)$$

κ_ν depends also on processes

• Inverse Bremsstrahlung free-free Scattering:

$$h\nu + [X_1^+ + e(\epsilon)] \rightarrow X_2^+ + e(\epsilon'), \quad (34)$$

Cross section - from the elastic scattering matrix elements for electron impact excitation. An approximate expression for the free-free opacity is

$$\kappa_\nu^{\text{ff}}(1, 2) = 3.7 \times 10^8 N_e N_i g_{\text{ff}} \frac{Z^2}{T^{1/2} \nu^3} \quad (35)$$

where g_{ff} is a Gaunt factor

• Photon-Electron scattering:

a) Thomson scattering when the electron is free

$$\kappa(\text{sc}) = N_e \sigma_{\text{Th}} = N_e \frac{8\pi e^4}{3m^2 c^4} = 6.65 \times 10^{-25} \text{ cm}^2/\text{g} \quad (36)$$

b) Rayleigh scattering when the electron is bound

$$\kappa_\nu^{\text{R}} = n_i \sigma_\nu^{\text{R}} \approx n_i f_t \sigma^{\text{Th}} \left(\frac{\nu}{\nu_I} \right)^4 \quad (37)$$

$h\nu_I =$ binding energy, $f_t =$ total oscillator strength associated with the bound electron.

The total monochromatic opacity is then given by

$$\kappa_\nu = \kappa_\nu(\text{bb}) + \kappa_\nu(\text{bf}) + \kappa_\nu(\text{ff}) + \kappa_\nu(\text{sc}) \quad (38)$$

Mean Opacity: The equation of state (EOS)

- Requires ionization fractions and level populations of each ion of an element in levels with non-negligible occupation probability
- Saha equation - number density for ionization state $n_S(X^i)$ of excited ion X^i at thermal equilibrium as

$$n_S(X^i) = \frac{g_i}{2g_s} \frac{h^3}{(2\pi m_e kT)^{3/2}} \exp\left(-\frac{E_{gi}}{kT}\right) n_e n(X^+) \quad (39)$$

In LTE the level populations are given by the Boltzmann equation,

$$\frac{N_i}{N_j} = \frac{g_i}{g_j} e^{-h\nu/kT} \quad (40)$$

The combination of the Saha-Boltzmann equations specify the Equation-of-State (EOS), in LTE

$$N_{ij} = \frac{N_j g_{ij} e^{-E_{ij}/kT}}{U_j} \quad (41)$$

g_{ij} = statistical factor of level i of ionization state j , the partition function is

$$U_j = \sum_i g_{ij} e^{(-E_{ij}/kT)} \quad (42)$$

There are several ways to modify the Saha-Boltzmann equations to incorporate the effect of plasma interactions into the EOS. The modified density function

is then

$$N_{ij} = \frac{N_j g_{ij} w_{ij} e^{-E_{ij}/kT}}{U_j}, \quad (43)$$

where the partition function is

$$U_j = \sum_i g_{ij} w_{ij} e^{(-E_{ij}/kT)} \quad (44)$$

Rosseland mean $\kappa_R(T, \rho)$:

Harmonic mean opacity averaged over the Planck function, ρ is the mass density (g/cc),

$$\frac{1}{\kappa_R} = \frac{\int_0^\infty \frac{1}{\kappa_\nu} g(u) du}{\int_0^\infty g(u) du},$$

where $g(u)$ is the Planck weighting function

$$g(u) = \frac{15}{4\pi^4} \frac{u^4 e^{-u}}{(1 - e^{-u})^2}, \quad u = \frac{h\nu}{kT}$$

$g(u)$, for an astrophysical state is calculated with different chemical compositions H (X), He (Y) and metals (Z), such that

$$X + Y + Z = 1$$

Solar abundances: $X=0.7, Y=0.28, Z=0.02$

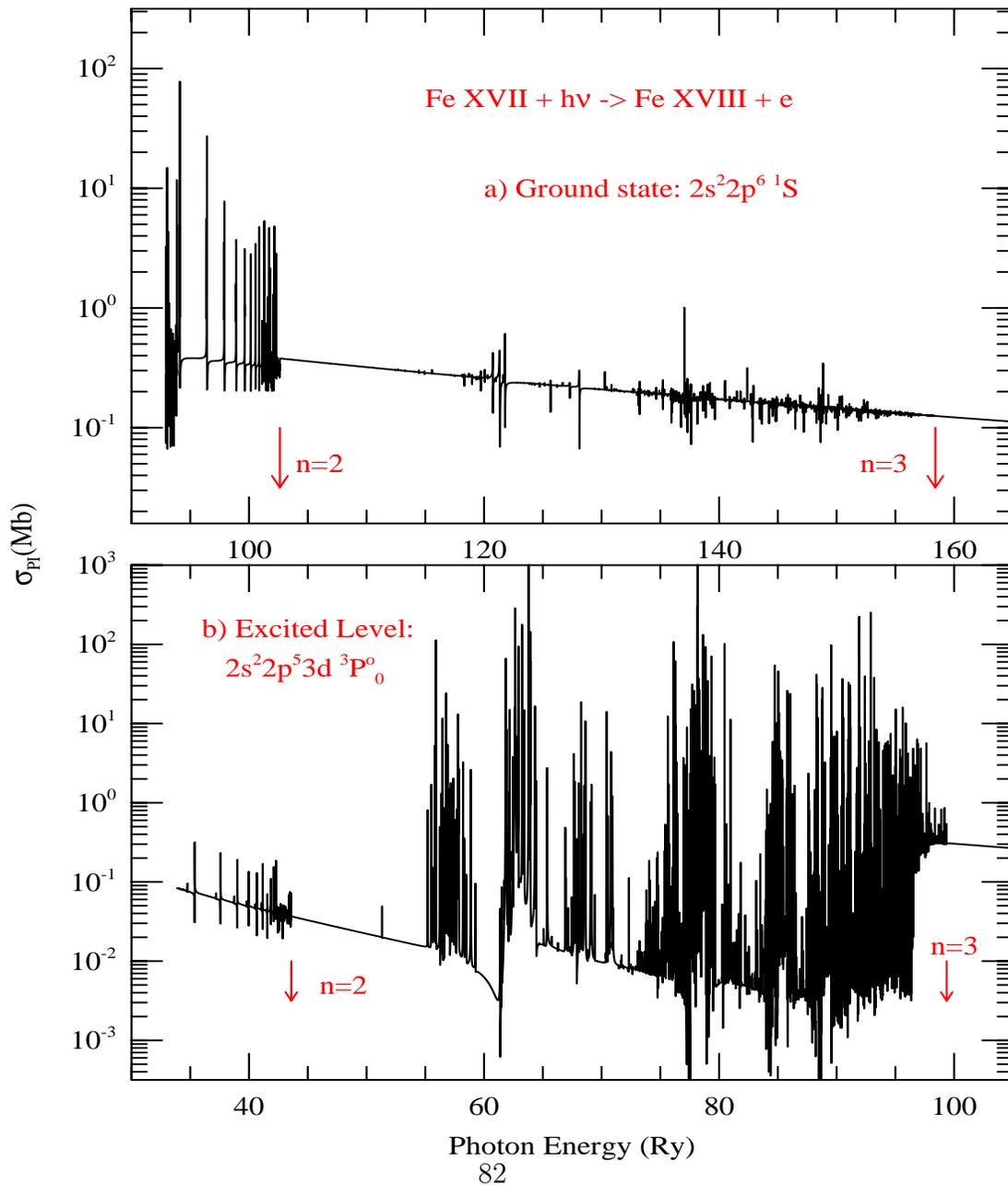
PHOTOIONIZATION CROSS SECTION: Fe XVII

- **Top:** Ground level: $n=2$ resonances are important

- **Bottom:** Excited level: $n=3$ resonances are important

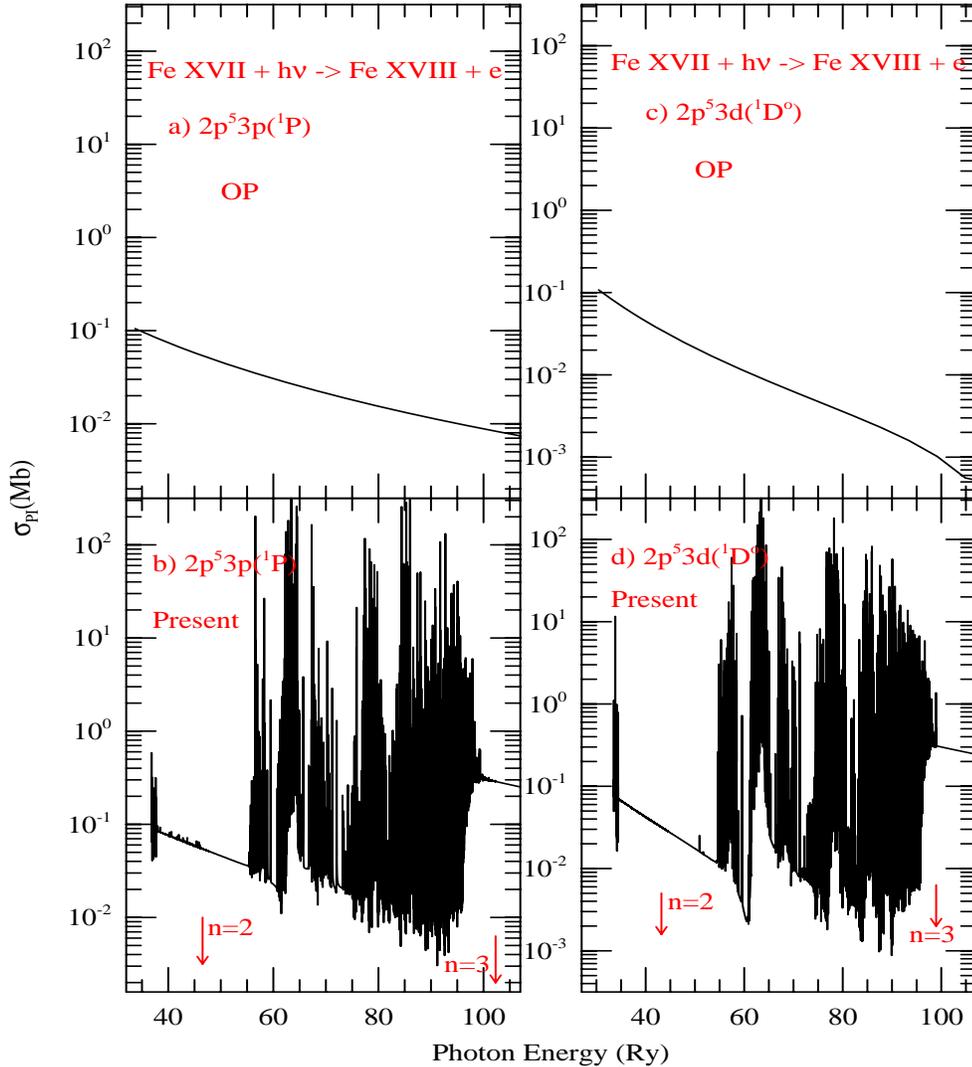
Arrows point energy limits of $n=2$ & 3 core states

Photoionization Cross sections of Fe XVII



COMPARISON OF PHOTOIONIZATION CROSS SECTIONS

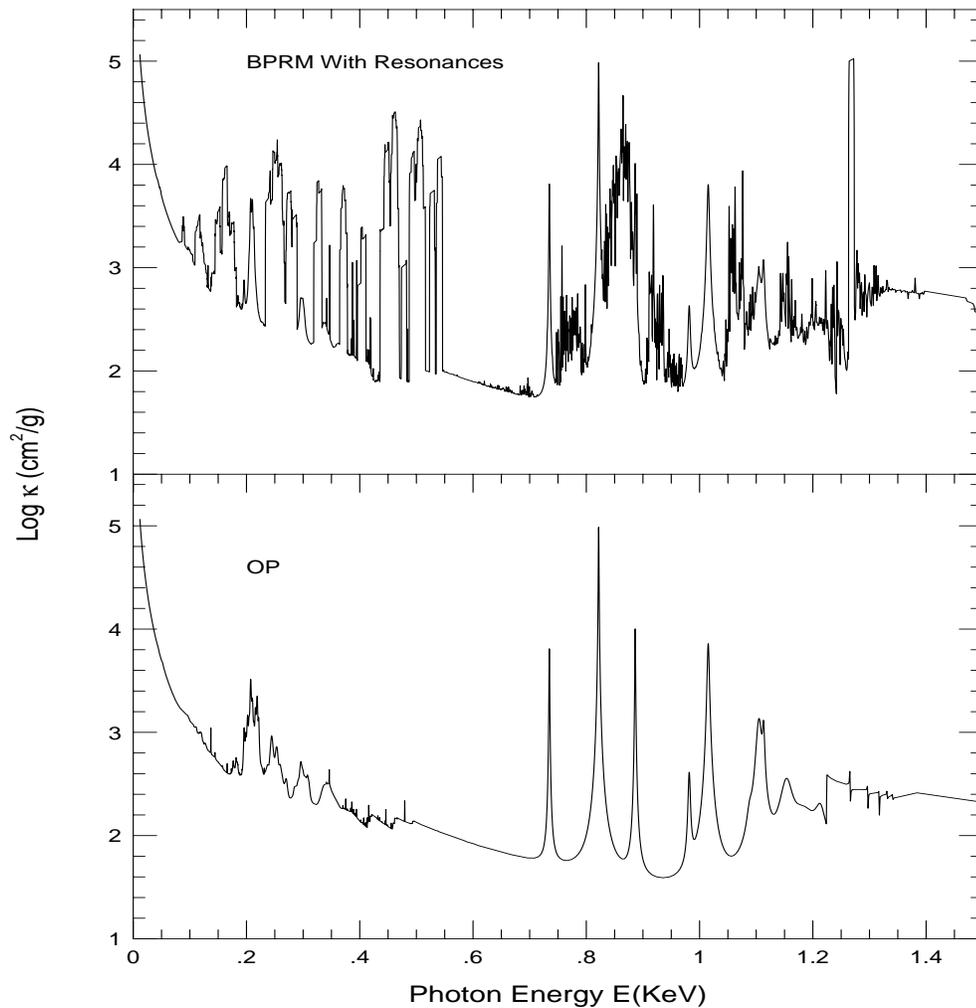
- Photoionization cross sections of Fe XVII: (a,b) level $2p^53p(^1P)$ & (c,d) level $2p^53d(^1D^o)$
- Present σ_{PI} (Nahar et al 2010) is much more highly resolved than those given by the Opacity Project (OP). Without inclusion of $n=3$ core states, σ_{PI} is considerably underestimated



MONOCHROMATIC OPACITY OF Fe XVII

(Nahar et al. 2011)

- Monochromatic opacity κ of Fe XVII at temperature $T = 2.24 \times 10^6$ K and electron density $N_e = 10^{23} \text{ cm}^{-3}$, corresponding to the base of the solar convection zone where the ion is the largest contributor to opacity.



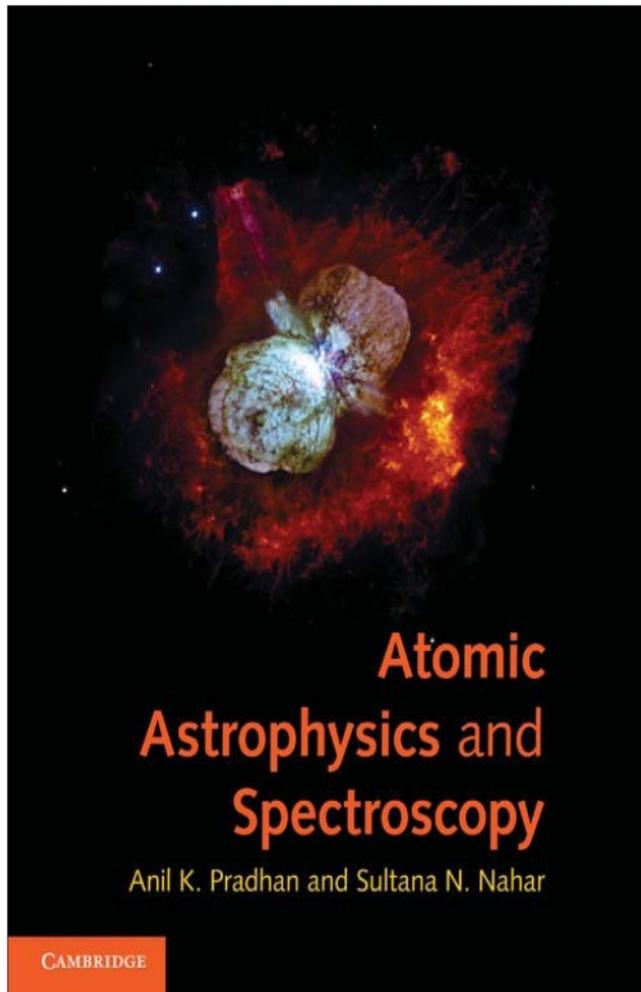


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and Arizona, Germany, Italy



Large Binocular Telescope (LBT)

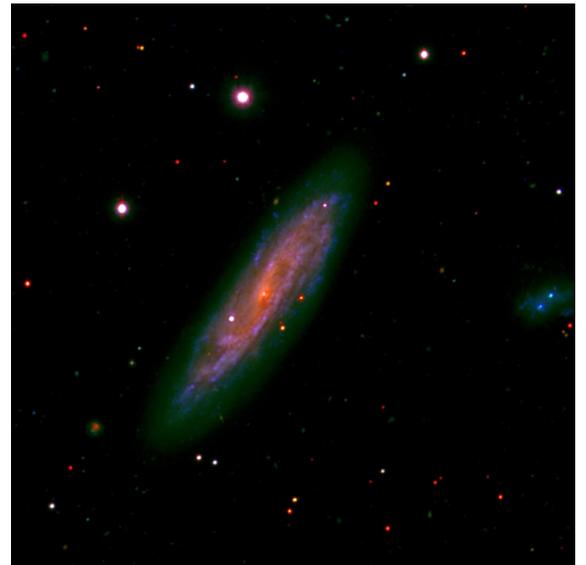
Largest Telescope: 8.4m Mirrors (11.8m), NIR-Optical



Galaxy: M101



Galaxy: M51



Galaxy: NGC 2770